



*Setting the Standard for Automation™*

# Fertilizer Symposium - 1<sup>st</sup> December 2018

**Advantages of Advanced Radiometric  
Measurement**

ISA-D: "Fertiliser Symposium-2018"

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# Radiometric Measurement Technology

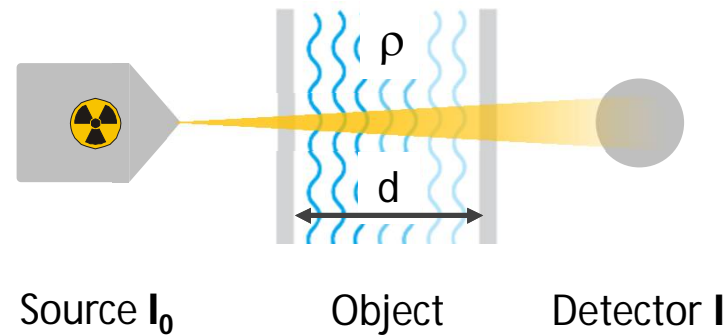
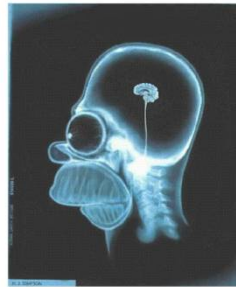
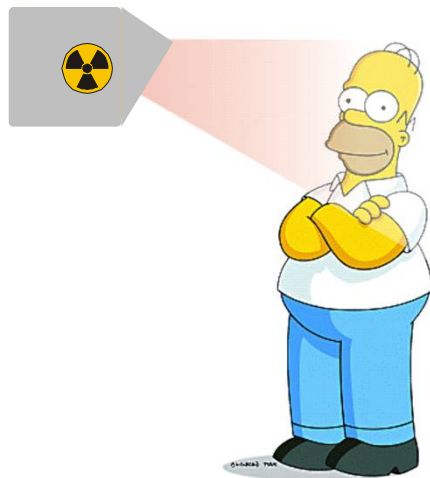
Why gamma measurement technology?

- Contactless, non-invasive
  
- For severe operating conditions
  - High temperature
  - High pressure
  - Foam formation
  - Corrosive media
  - Wall build-ups
  
- Long-term solution
  - Maintenance-free!
  - No recalibration



# Measuring Principle – Radiometry

What is really behind it?

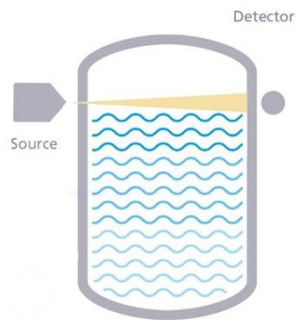


$$I = I_0 \cdot \exp(-\mu \cdot \rho \cdot D)$$

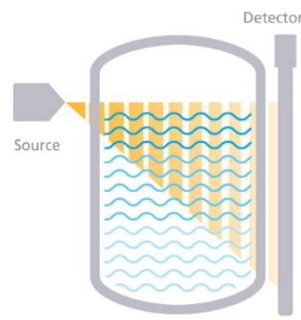
- Source: emits gamma radiation  $I_0$
- Object: attenuation of radiation
- Detector: measures radiation intensity

# Applications – measure where others fail

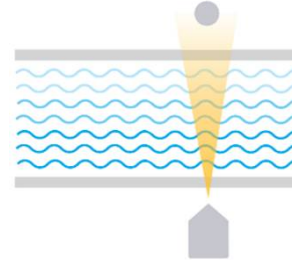
Radiometry is versatile!



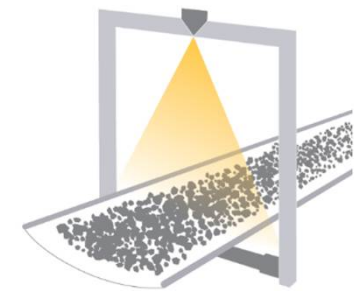
Limit Switch



Level



Density /  
Concentration



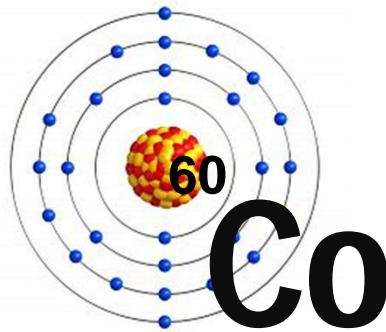
Bulk Flow

Furthermore:

- Interfaces
- Potassium content
- Moisture measurement (neutrons)
- and many more

# Industrial Used Nuclear Isotopes


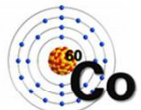
Cobalt – 60 or Cesium - 137



- There are many known natural and artificial isotopes
- However just a few are really used for measurement purposes.
- This is mainly Cobalt – 60 and Cesium – 137
- They vary from decay scheme, energy, intensity, half life time, etc
- Make the best selection for the measurement purpose
- Cobalt – 60 has some unique advantages in fertilizer production

# Isotope Selection



Cobalt-60 or Cesium-137

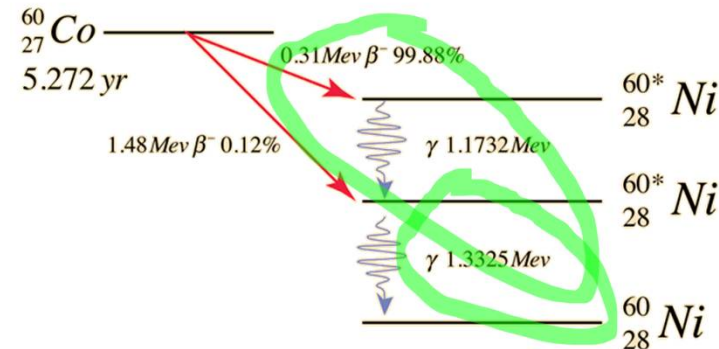
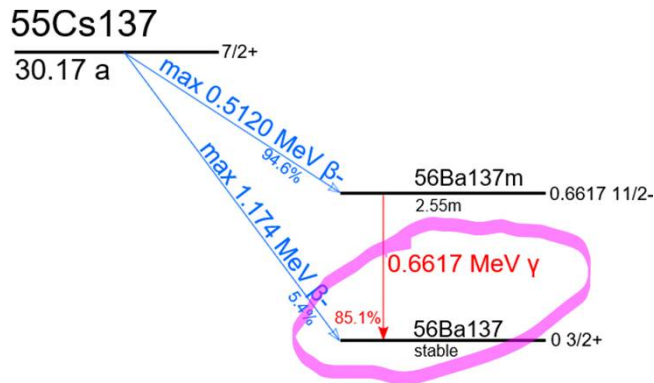
Isotope	Cs-137	Co-60
		
Energy	660 keV	1200 keV
Half-Life Time	30.18 years	5.27 years
Half Value Layer (steel)	16mm / 0.63in	21.6mm / 0.85in
Gamma efficiency per decay	0.85	2

- Half-Life Time  $\neq$  Source Life Time
- Recommended working life of sources (acc. ISO2919) is ~10-15 years and is incorporated by design, independent of isotope incorporated by design
- Depending on governmental requirements
- Real operating lifetime can be even longer

# Cobalt-60 vs. Cesium-137

## Decay Schemes

Isotope	Cs -137	Co-60
		
Energy	660 keV	1200 keV



$$I = I_0 * e^{-\mu * \rho * d} \quad \rightarrow \quad \frac{I_0^{Co}}{I_0^{Cs}} = \frac{2.0}{0.85} = 2.35$$

- Cobalt is more efficient in emitting gamma radiation

## Advantages of Cobalt-60

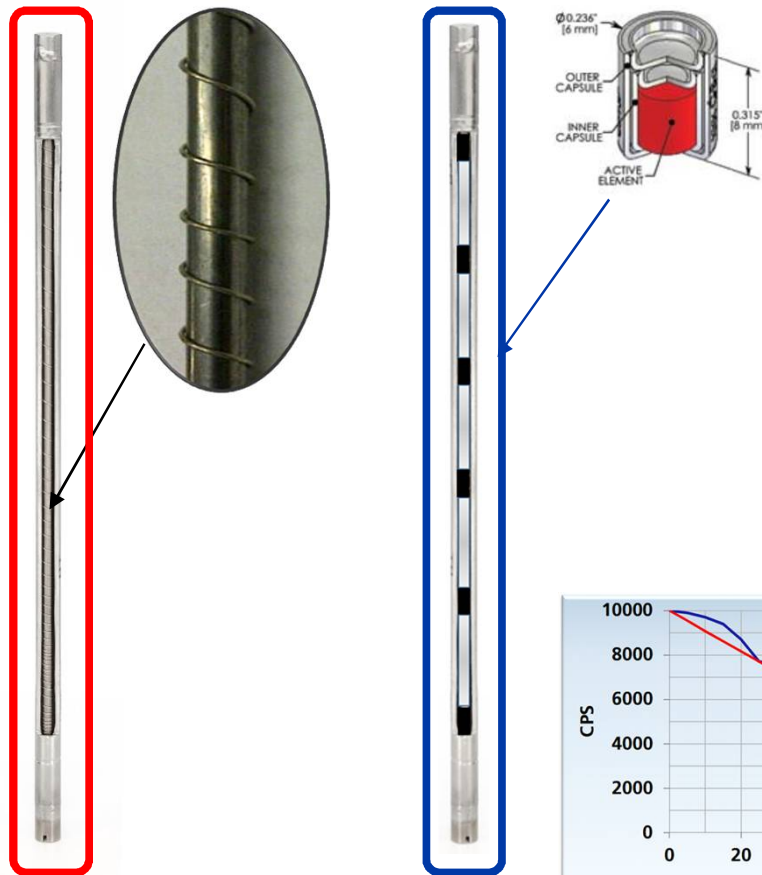
What makes it isotope of choice for fertilizers?

- Higher gamma efficiency ( factor 2.35)m, more countrate per activity
- Higher energy
  - More resilient against gas property changes, especially at lower count rates
  - Better radiation through massive walls
- Cost efficient rod sources
  - Rod sources provide full linearity and highest accuracy

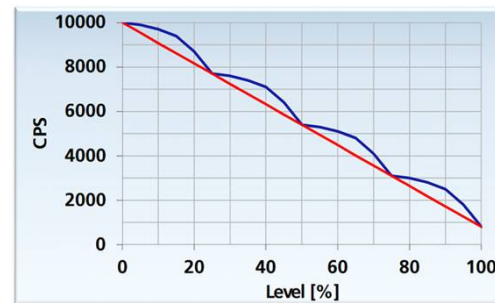


# Advantages of Rod Sources

Why only Cobalt – 60 can make a real rod source!

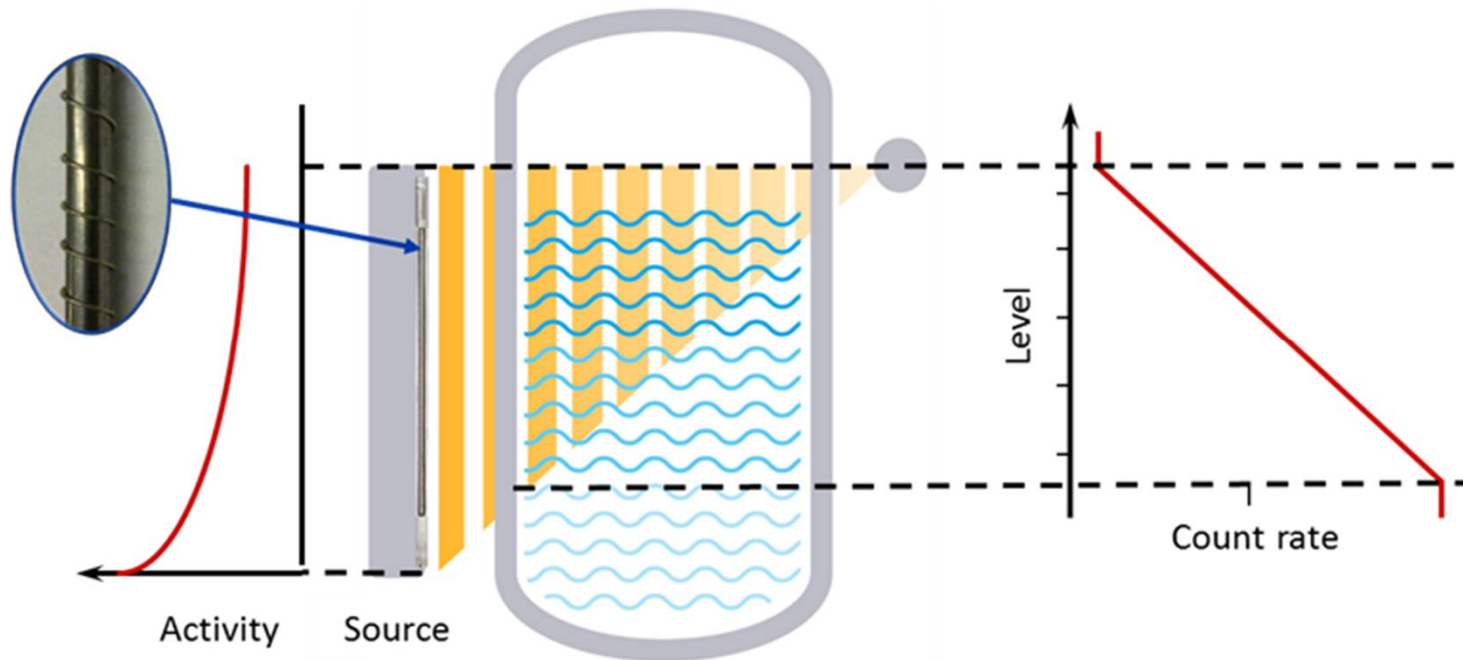


- With Co-60, activity is distributed over the complete source length
- Co-60 wire wound around a core
- By changing the pitch the activity distribution can be controlled
- Highest achievable accuracy and linearity
- Customer specific design  
Adapted to the individual geometry



# Rod Sources

100 % Linearity for highest Accuracy

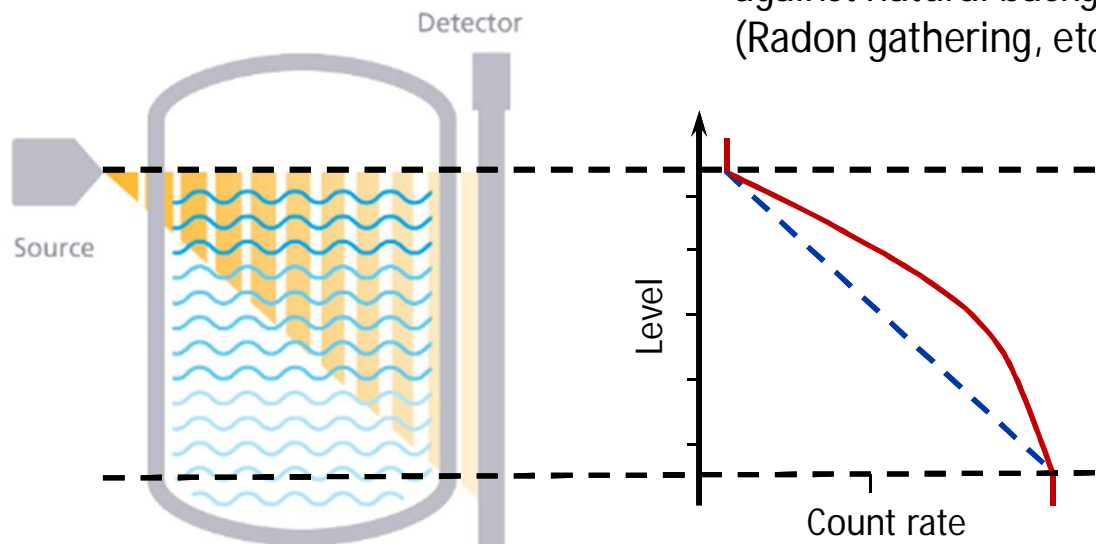


- With a custom specific activity distribution, we can reach a linear measurement curve
- Point detector very tolerant against background variation yield to high accuracy in a level measurement

# Level: Point Source / Rod Detector

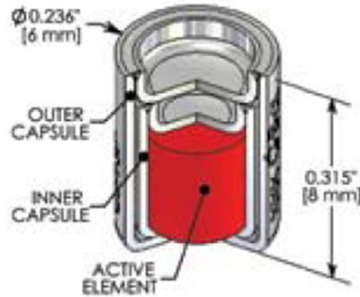
„Industry Standard“

- Typically 2 point calibration at 0% and 100%
- Assumption: Linear Calibration (blue)
- Reality: Not-linear (red)
- Inherent systematic error
  - Lower accuracy
  - However measurement is reproduceable
- Rod detector being very vulnerable against natural background variation (Radon gathering, etc.)



# Safety

Co-60 vs Cs-137



## Cs-137

- Is a bound in a ceramic matrix. If this container breaks it can be distributed as powder in the atmosphere, e.g. in the event of fire
- After entering the body, caesium is more or less uniformly distributed throughout the body, with the highest concentrations in soft tissue.
- Cs-137 itself is very reactive and produces a water soluble compound

## Co-60

- Co-60 is a metal, not water soluble and remains local when exposed to fire
- Molten Co-60 is forming alloys with the metal from the building structure
- Lower activity to dispose

# Typical applications

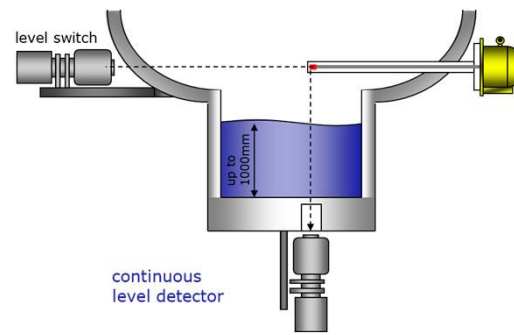
## Stripper in Urea Production



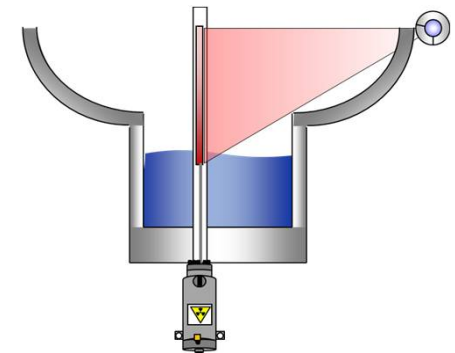
### Challenges:

- Thick walls (> 50mm)
- High gas density – high absorption
- Frequent and regular variations of gas property  
→ Gas density compensation needed
- High Accuracy required

### Solution 1

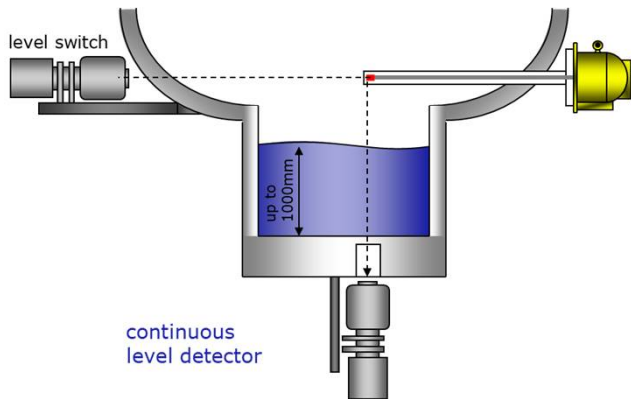


### Solution 2



# Stripper in Urea Production

## Solution 1



### Solution:

- Point Source – Point Detector
- Dip Pipe for source arrangement
- Local thinning of a detector window (blind hole)
- Gas density compensation algorithm

### Benefits:

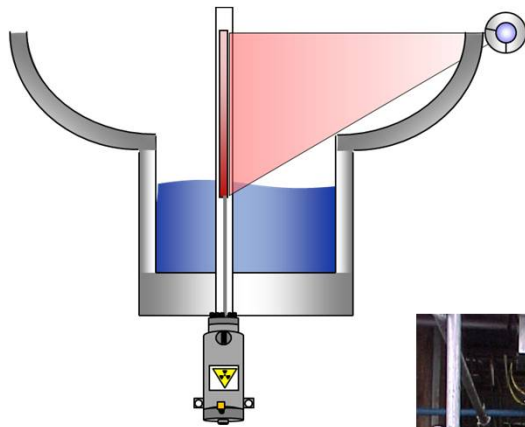
- Easy to calibrate
- Easy handling of source

### Limitation:

- Limited accuracy, due to non-linearity
- Blind hole thinning required

# Stripper in Urea Production

## Solution 2



### Solution:

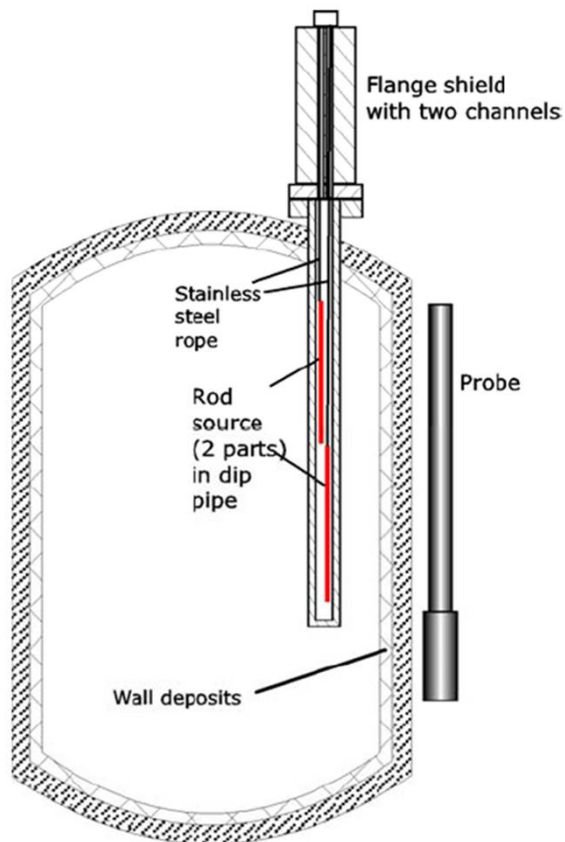
- Rod Source – Point Detector
- Dip Pipe for rod - source arrangement
- Gas density compensation algorithm
- Optimized activity distribution

### Benefits:

- Easy to calibrate
- Rod source shield can cover the whole rod  
→ no radiation exposure during maintenance
- 100% linearity
- Highest accuracy
- Reliable and repeatable

# Typical applications

## Continuous Reactor Level in Urea Production



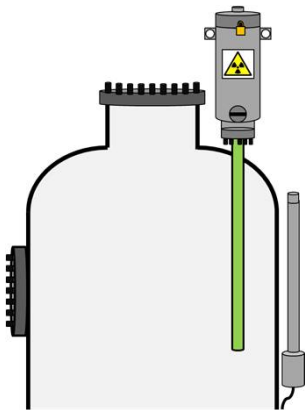
### Challenges:

- Thick walls ( > 50mm)
- High gas density – high absorption
- Frequent and regular variations of gas property  
→ Gas density compensation needed
- High Accuracy required – linear measurement curve
- Source container to contain the complete rod source during maintenance



# Continuous Reactor Level in Urea Production

Rod Source – Rod Detector



## Solution:

- Rod source in dip line
- Rod detector outside
- High Accuracy required – linear measurement curve
- Source shield design to hold rod source and shield completely
- Optimized activity distribution

## Benefits:

- Easy to calibrate
- Rod source shield can cover the whole rod  
→ no radiation exposure during maintenance
- 100% linearity
- Highest accuracy, reliability and repeatability

# Compensation of gas property changes

What is causing the higher absorption?

Changing gas properties simulate level changes, that do not exist. Increasing gas density simulates increasing level

- Changes of gas density
- Change of hydrogen content

$$I = I_0 \cdot e^{-\mu \cdot \rho \cdot d}$$

absorption coefficient

gas density

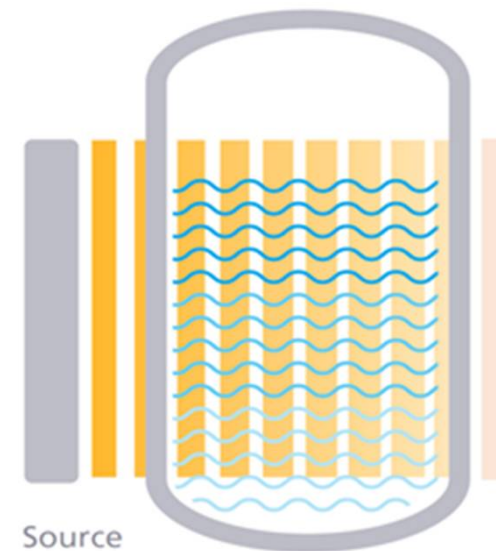
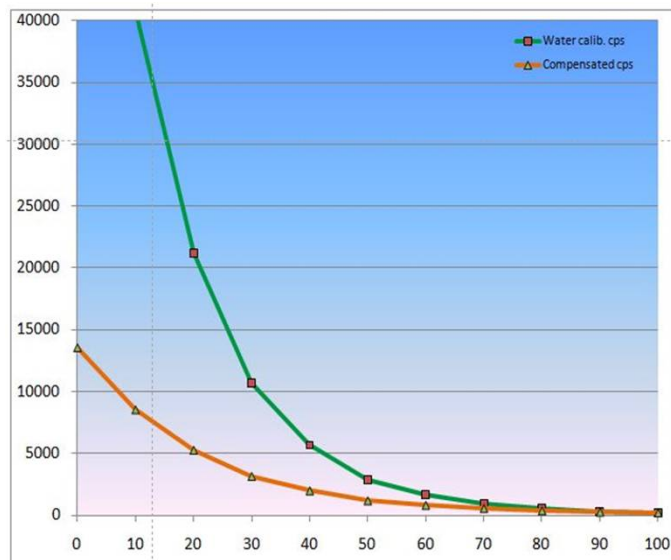
attenuated count rate      original count rate

# Compensation of gas property changes

How to compensate?

Here are two way to mitigate the influence of gas property changes

- If the changes are repeatable and stable, a “calculated” compensation can be factored in
- Real Measurement of gas density, with an additional detector



# Any Questions?



Thanks for your attention!



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