Henderson Lake

CUMMINGS TOWNSHIP
HILL TOWNSHIP
OGEMAW COUNTY
MICHIGAN

WATER QUALITY TESTING
2016

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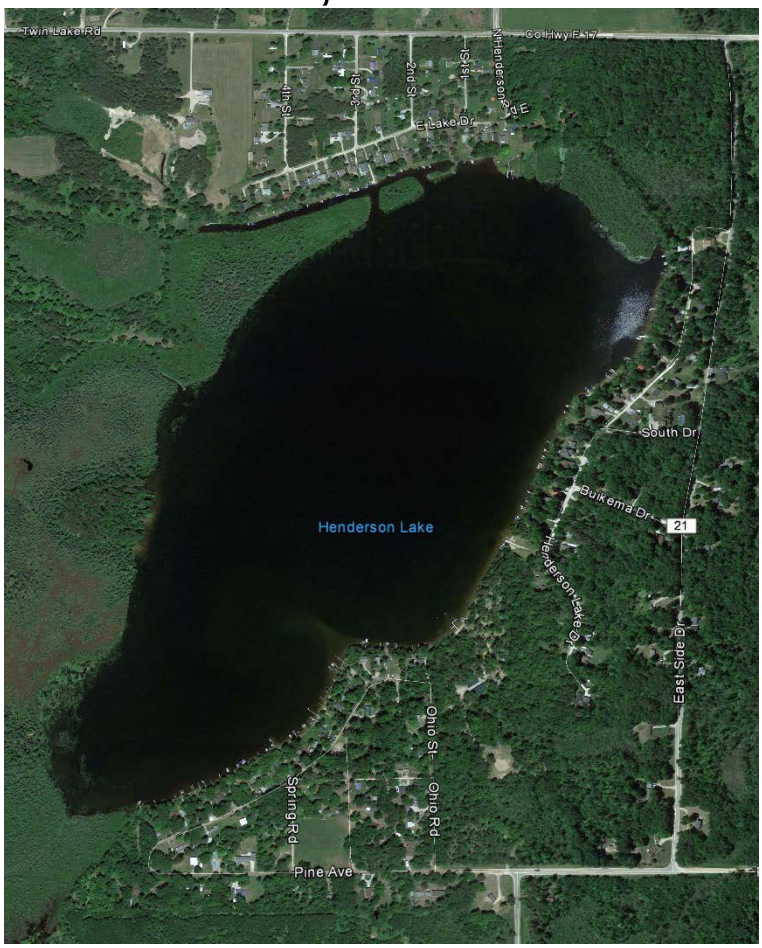


Henderson Lake

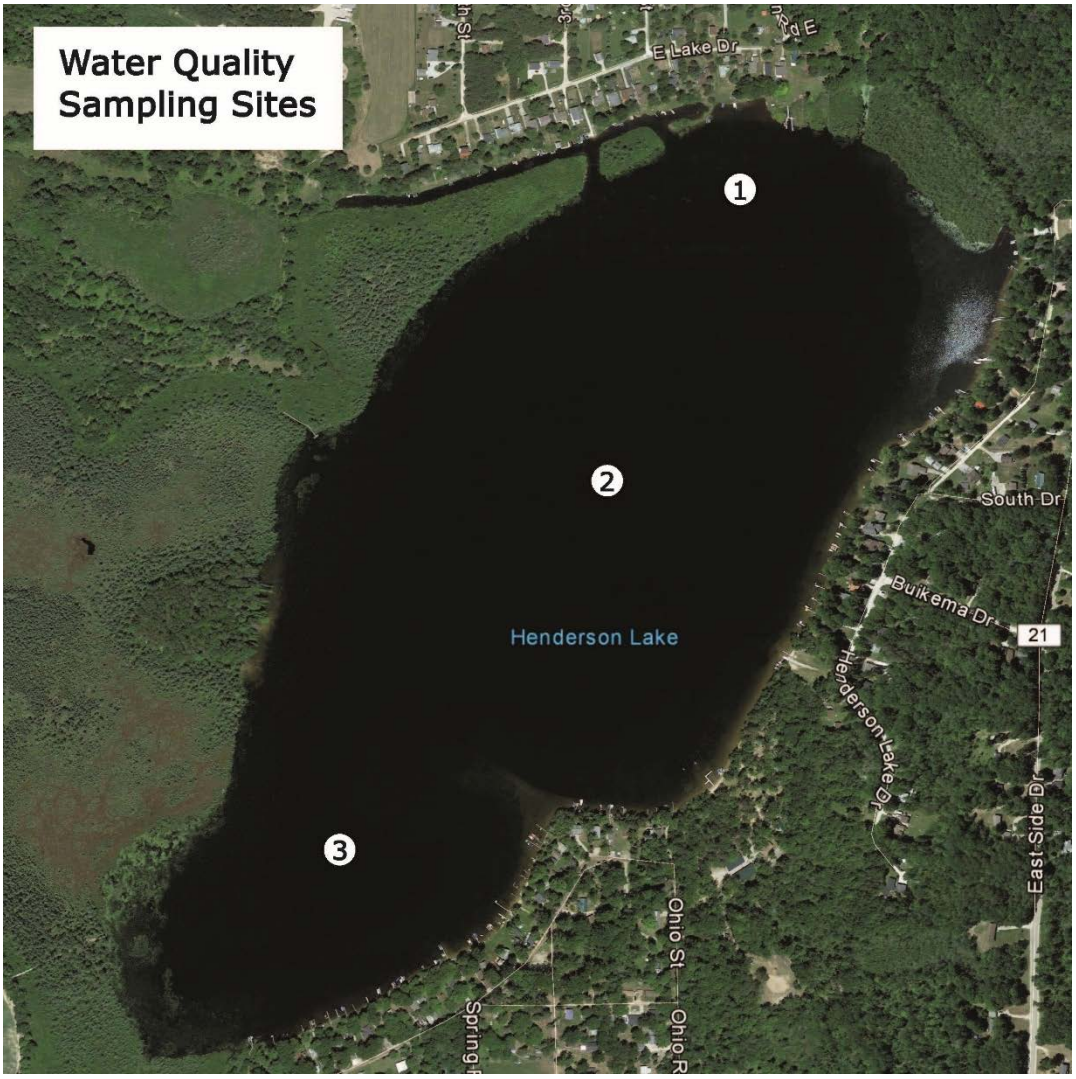
Henderson Lake a 177-acre moderately hard water lake, located in Section 31 of Hill Twp (T23N R4E) and Sections 36, of Cummings Twp (T23N R3E), Ogemaw County, MI. The lake has a maximum depth of 23 feet, and the length of the shoreline is 16,017 feet (3.03 miles). The lake is spring fed, and has one main inlet and one outlet.

THE WATER QUALITY STUDY

During certain periods, Michigan lakes have poorer water quality than the rest of the year. Usually our studies involve sampling the lake in early spring when phosphorus from the bottom sediments may be mixed into the water column causing early spring algal blooms; and late summer when the water is warmest, and the lake is stratified (if it stratifies). Thus, if the lake gets high marks for water quality during early spring and late summer it probably has pretty good water quality all year.

This study looked at the 2016 summer water quality.

THE SAMPLE STATIONS



The locations of the three in-lake sample stations are shown as circles on the map of the lake. The deepest spot on the lake is at site 2.

SAMPLE DATES and ANALYSES

Savin Lake Services personnel collected three surface samples at the stations shown on the map on September 28th, 2016. Top to bottom temperature and dissolved oxygen profile data were also collected at Station 2 in late summer.

Dissolved oxygen, temperature, pH and Secchi disk transparency measurements were conducted in the field. Total phosphorus, conductivity, alkalinity, total nitrate, and chlorophyll α analysis was completed at an independent laboratory.

THE DATA

The data discussed below are found in the table at the end of this report.

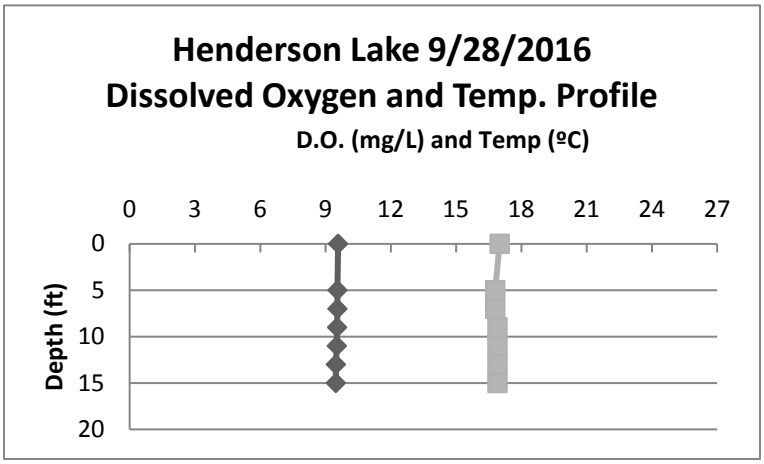
TEMPERATURE AND DISSOLVED OXYGEN

Temperature exerts a wide variety of influences on most lakes, such as the separation of layers of water (stratification), solubility of gases, and biological activity.

Dissolved oxygen is the parameter most often selected by lake water quality scientists as being important. Besides providing oxygen for aquatic organisms in natural lakes, dissolved oxygen is involved in phenomena such as phosphorus precipitation to, and release from, the lake bottom sediments and decomposition of organic material in the lake.

2016

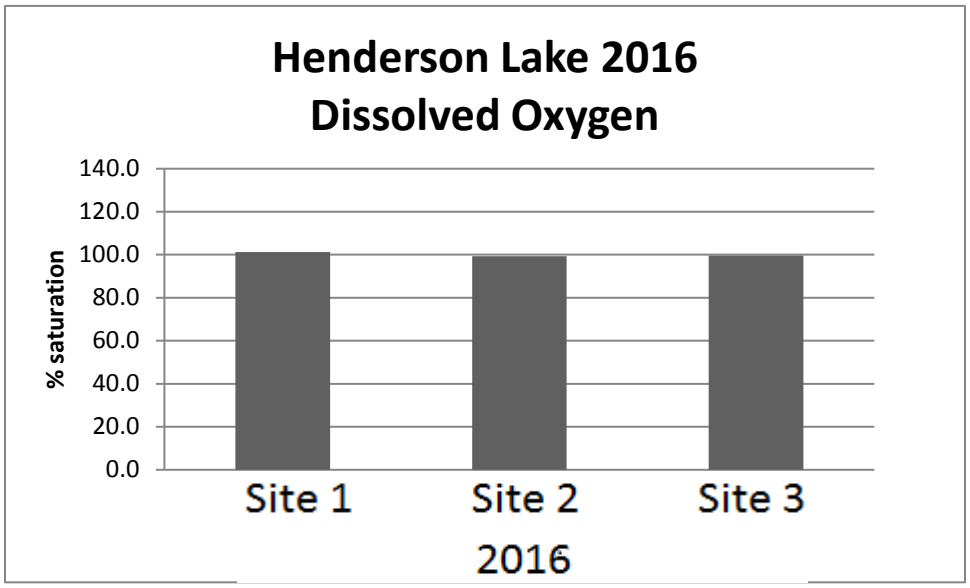
The dissolved oxygen concentrations were good in the late summer for the temperature of the water. The water may have turned shortly before the readings were taken. Therefore, no thermocline was observed at the time of sampling. However, it may be the possibility that no thermocline developed at all. An earlier reading next year will help determine the cause. The dissolved oxygen similarly remained steady throughout the water column.



Temp (ºC)	D.O. (mg/L)	Depth (ft)
17.0	9.58	0
16.8	9.55	5
16.8	9.55	7
16.9	9.53	9
16.9	9.52	11
16.9	9.48	13
16.9	9.48	15

Low dissolved oxygen concentrations (below 4 milligrams per liter) are generally insufficient to support fish life. In most Michigan lakes, there is no dissolved oxygen below the thermocline in late summer. Some experts like to see some dissolved oxygen in the bottom water of a lake, even if it is almost zero. This is because as long as there is some dissolved oxygen in the water at the bottom of the lake, phosphorus precipitated by iron to the bottom sediments will remain there. Once a lake runs out of dissolved oxygen in the water at the bottom iron comes back into solution. When that happens, it releases the phosphorus back into the water. This can cause additional algae to grow when the lake mixes.

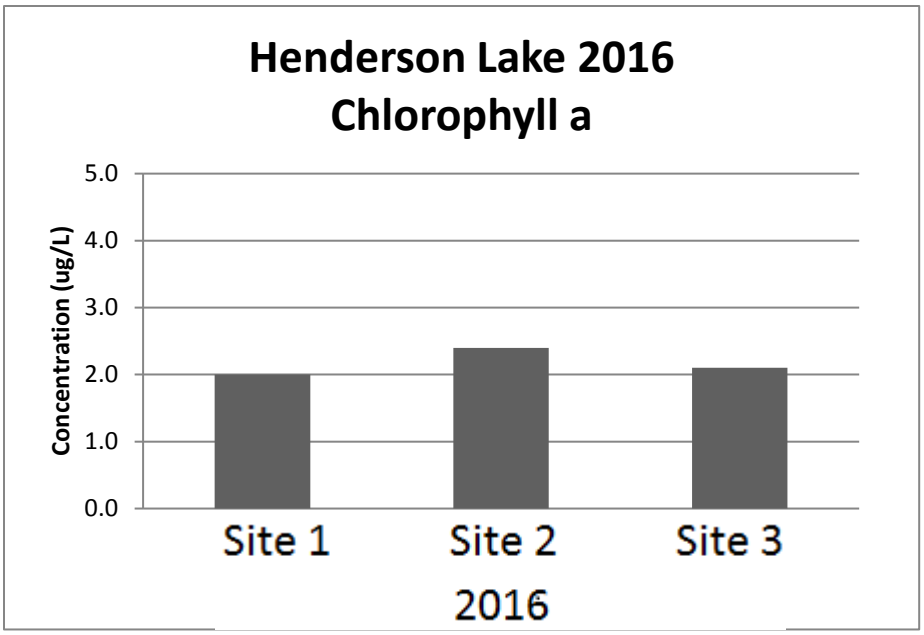
DISSOLVED OXYGEN, PERCENT SATURATION



Because the amount of dissolved oxygen a water can hold is temperature dependent with cold water holding more than warm water, dissolved oxygen saturation is often a better way to determine if oxygen supplies are adequate. The best is between 90 and 110 percent.

The 2016 concentrations ranged from 99 to 102 percent. These values are great for the temperature of the water.

CHLOROPHYLL A



Chlorophyll a is used by lake scientists as a measure of the biological productivity of the water. Generally, the lower the chlorophyll a, the better. High concentrations of chlorophyll a are indicative of an algal bloom in the lake, an indication of poor lake water quality. The highest surface chlorophyll a concentration found by Wallace Fusilier (Water Quality Investigators, WQI) in a Michigan lake was 216 micrograms per liter. Best is below one microgram per liter.

The graph shows Henderson Lake chlorophyll a concentrations were 2.0, 2.4, and 2.1 micrograms per liter. These are good concentrations. This data suggest Henderson Lake contains a low amount of algae growth.

SECCHI DISK TRANSPARENCY (originally Secchi's disk)

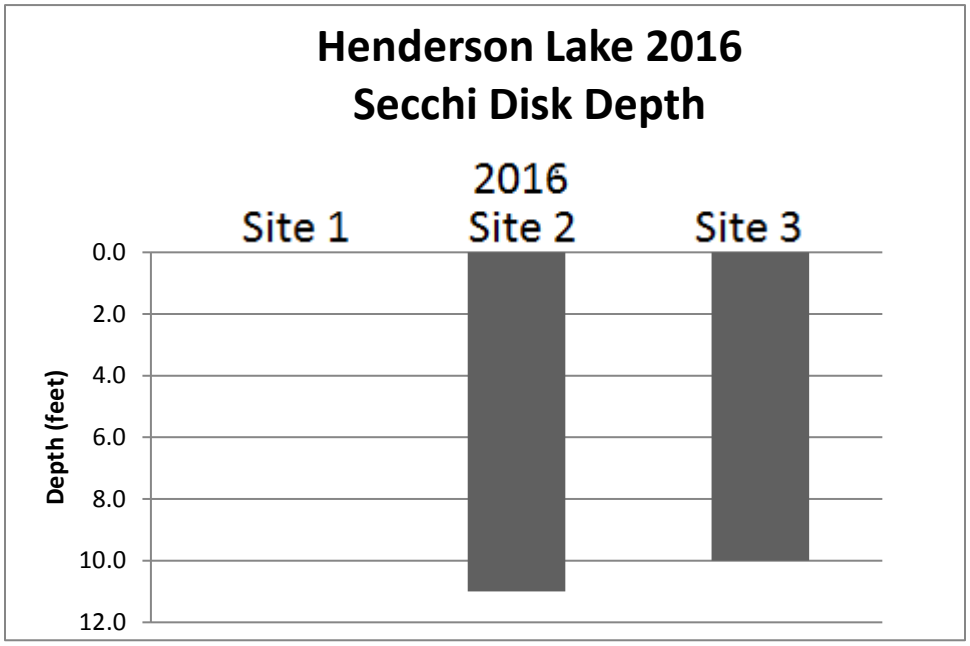
In 1865, Angelo Secchi, the Pope's astronomer in Rome, Italy devised a 20-centimeter (8 inch) white disk for studying the transparency of the water in the Mediterranean Sea. Later an American limnologist (lake scientist) named Whipple divided the disk into black and white quadrants which many are familiar with today.

The Secchi disk transparency is a lake test widely used and accepted by limnologists. The experts generally felt the greater the Secchi disk depth, the better quality the water. However, one Canadian scientist pointed out acid lakes have very deep Secchi disk readings. (Would you consider a very clear lake a good quality lake, even if it had no fish in it? It would be almost like a swimming pool.) Most lakes in southeast Michigan have Secchi disk transparencies of less than ten feet. On the other hand, Elizabeth Lake in Oakland County had 34 foot Secchi disk readings in summer 1996, evidently caused by a zebra mussel invasion a couple of years earlier.

Most limnology texts recommend the following: to take a Secchi disk transparency reading, lower the disk into the water on the shaded side of an anchored boat to a point where it disappears. Then raise it to a point where it's visible. The average of these two readings is the Secchi disk transparency depth.

Secchi disk measurements should be taken between 10 AM and 4 PM. Rough water will give slightly shallower readings than smooth water. Sunny days will give slightly deeper readings than cloudy days. However, roughness influences the visibility of the disk more than sunny or cloudy days.

SECCHI DISK DATA



Henderson Lake's secchi disk readings were 11 feet at site 2, 10 feet at site 3. Site 1's readings reached the bottom of the lake and therefore can't be quantified. These values are ok.

TOTAL PHOSPHORUS

Although there are several forms of phosphorus found in lakes, the experts selected total phosphorus as being most important. This is probably because all forms of phosphorus can be converted to the other forms. Currently, most lake scientists feel phosphorus, which is measured in parts per billion (1 part per billion is one second in 31 years) or micrograms per liter (ug/L), is the one nutrient which might be controlled. If its addition to lake water could be limited, the lake might not become covered with the algal communities so often found in eutrophic lakes.

Based on WQI's studies of many Michigan inland lakes, they've found many lakes were phosphorus limited in spring (so don't add phosphorus) and nitrate limited in summer (so don't add nitrogen).

10 parts per billion is considered a low concentration of phosphorus in a lake and 50 parts per billion is considered a high value in a lake by many limnologists.

In 2016, Henderson Lake's total phosphorus concentrations were all below the detection limit of 8 micrograms/L. These values are great. No graph is displayed due to no exact value being determined.

NITRATE NITROGEN

Nitrate, also measured in the parts per billion range, has traditionally been considered by lake scientists to also be a limiting nutrient. The experts felt any concentration below 200 parts per billion was excellent in terms of lake water quality. The highest value found by Fusilier was 48,000 parts per billion in an Ottawa County river which flowed into Lake Macatawa in Holland, Michigan

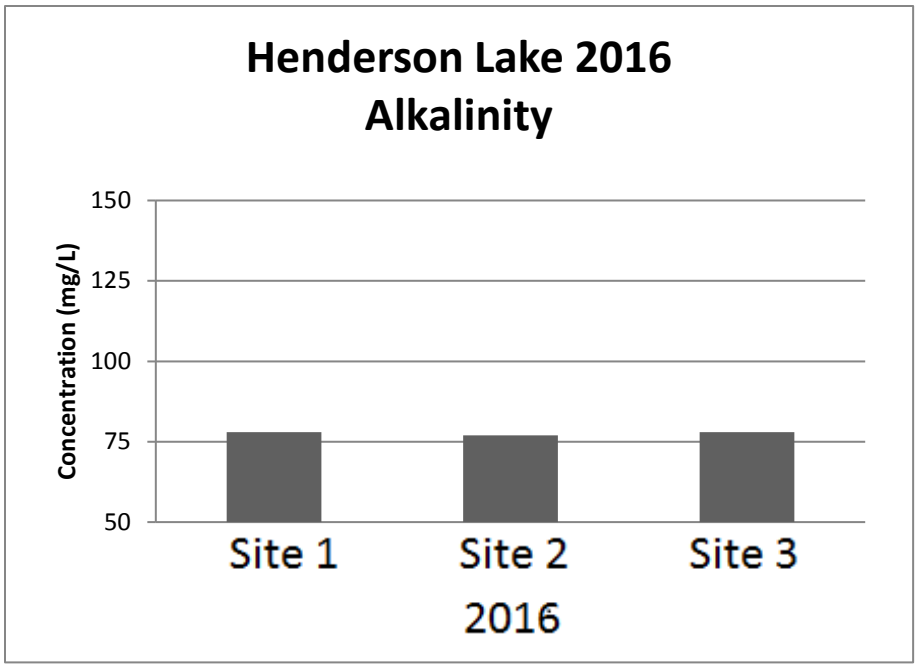
On the other hand, WQI has studied hundreds of Michigan inland lakes, and many times we find them nitrate limited (very low nitrate nitrogen concentrations), especially in summer.

In 2016, the values for all 3 sites were below the detection limit of 60 ug/L. Similarly to phosphorus, no graph is displayed due to a lack of an exact value.

Generally limnologists feel optimal nitrate nitrogen concentrations (which encourage maximum plant and algal growth) are about 10-20 times higher than phosphorus concentrations. The reason more nitrogen than phosphorus is needed is because nitrogen is one of the chemicals used in the production of plant proteins, while phosphorus is used in the transfer of energy, but is not used to create plant material. If the nitrate concentration is less than 10-20 times the phosphorus concentration, the lake is considered nitrogen limited. If the nitrate concentration is higher than 10-20 times the phosphorus concentration, the lake is considered phosphorus limited.

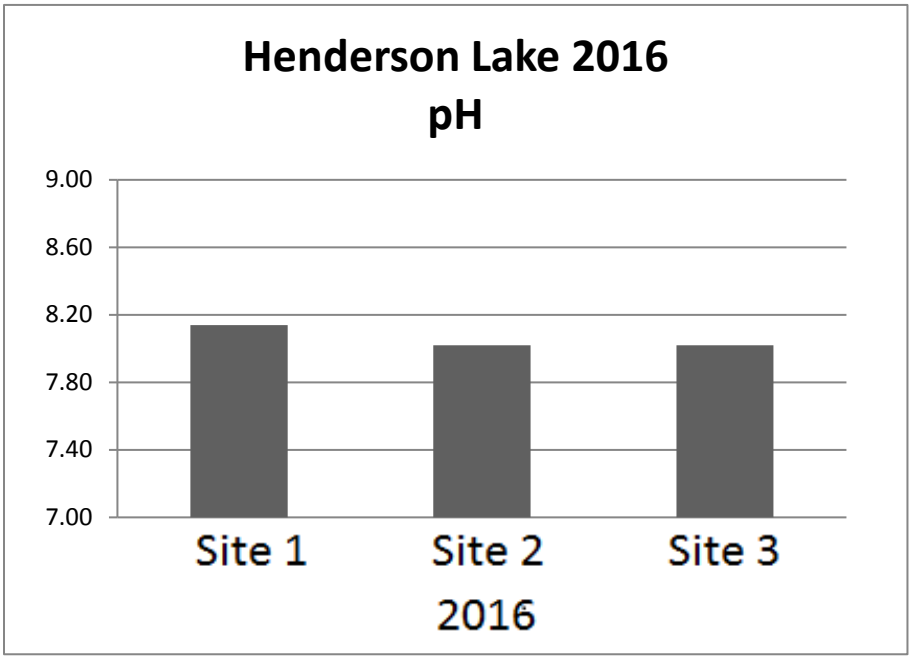
TOTAL ALKALINITY

Alkalinity is a measure of the ability of the water to absorb acids (or bases) without changing the hydrogen ion concentration (pH). It is, in effect, a chemical sponge. In most Michigan lakes, alkalinity is due to the presence of carbonates and bicarbonates which were introduced into the lake from ground water or streams which flow into the lake. In lower Michigan, acidification of most lakes should not be a problem because of the high alkalinity concentrations.



Henderson Lake's surface alkalinity data (77-78 milligrams per liter) indicates it is a moderately hard water lake. This is good, as hard water lakes have the ability to precipitate some of the phosphorus that enters the lake to the bottom sediments as calcium phosphate. This pretty much ties up that phosphorus in the sediments. Soft water lakes lack this ability. Despite the classification of being moderately hard (61-120 milligrams per liter), this is the softest water that Savin Lake Services has tested.

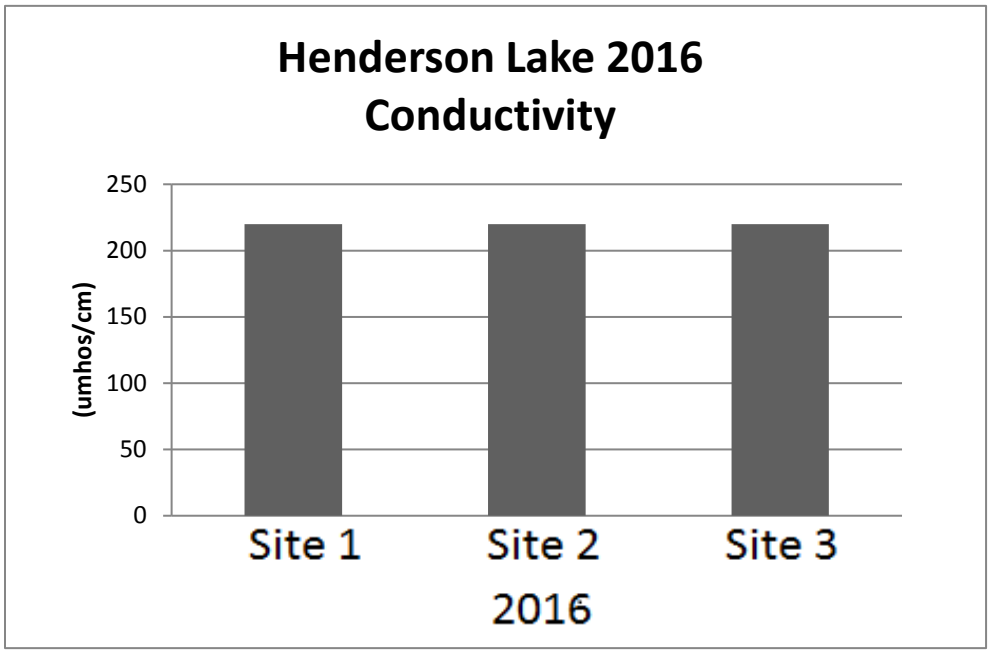
HYDROGEN ION CONCENTRATION (pH)



pH has traditionally been a measure of water quality. Today it is an excellent indicator of the effects of acid rain on lakes. About 99% of the rain events in southeastern Michigan are below a pH of 5.6 and are thus considered acid. However, there seems to be no lakes in southern Michigan which are being affected by acid rain. Most lakes have pH values between 7.5 and 9.0.

Henderson Lake's pH values (8.02 to 8.14) are normal for a Michigan moderately hard water lake.

SPECIFIC CONDUCTIVITY



Conductivity, measured with a meter, detects the capacity of a water to conduct an electric current. More importantly however, it measures the amount of materials dissolved in the water (salts), since only dissolved materials will permit an electric current to flow. Theoretically, pure water will not conduct an electric current. It is the perception of the experts that poor quality water has more dissolved materials than does good quality water.

The graph shows Henderson Lake's conductivities were all 220 umhos/cm. These are great values.

LAKE WATER QUALITY INDIX

The Lake Water Quality Index (LWQI) (Fusilier, 1982) used in this study to define the water quality of Henderson Lake was developed for two reasons. First, there was no agreement among lake scientists regarding which tests should be used to define the water quality of a lake; and second, there was no agreement among lake scientists regarding the meaning of the data collected during lake studies.

Development of the index involved two questionnaires which were sent to a panel of 555 scientists who were members of the American Society of Limnology and Oceanography. The panel was specifically selected because they were chemists and biologists with advanced degrees who studied lake water quality.

The first questionnaire asked the scientists to select tests which they felt should be used to define lake water quality.

The tests most often selected by the scientists became the index parameters (or tests). They were:

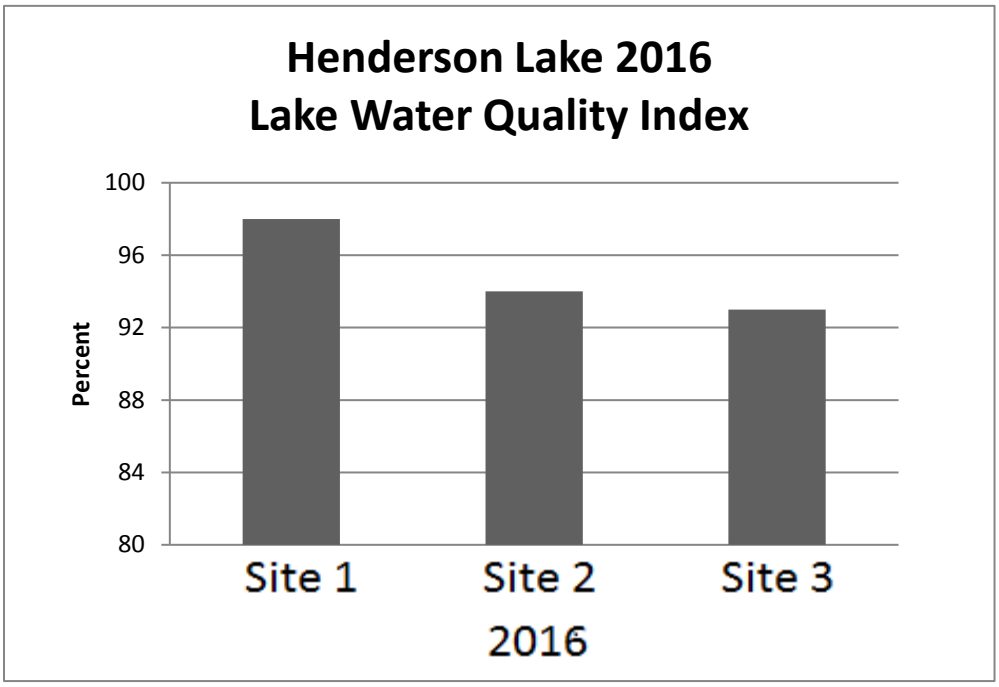
Dissolved oxygen (Percent saturation)
Total alkalinity
Chlorophyll a
Secchi disk depth
Total nitrate nitrogen
Total phosphorus
pH
Temperature
Conductivity

The second questionnaire, sent out after the first was returned, asked the scientists what the results of the tests they selected as good indicators of lake water quality meant.

After the responses to the second questionnaire were tabulated, the nine tests and the accompanying rating curves were combined into a Lake Water Quality Index.

The index ranges from 1 to 100, with 100 indicating excellent lake water quality. The index rated lakes about the same way teachers rate students: 90-100=A, 80-90=B, 70-80=C, 60-70=D, and below 60=E.

The highest index for a Michigan lake studied by Fusilier was Long Lake in Grand Traverse County at 100 in the spring of 1994. The lowest was 16 in an Ottawa County lake.



HENDERSON LAKE 2016 LAKE WATER QUALITY INDICES

The graph shows the water quality of Henderson Lake 98, 94, and 93. All sites received a grade of A.

THE LAKE WATER QUALITY INDEX CALCULATION SHEETS

The Lake Water Quality Index calculation sheets were developed to show graphically what the results of the nine different lake water quality tests meant in terms of lake water quality.

HOW TO READ THE LAKE WATER QUALITY INDEX CALCULATION SHEETS

Listed across the top of the calculation sheets are the tests selected by the panel of experts as being good indicators of lake water quality.

The figures which look like thermometers are graphs which convert the test results (the values found on the outside of the thermometer) to a uniform 0-100 lake water quality rating (found on the inside of the thermometer).

The calculation sheet combines all nine of the individual quality ratings into a single Lake Water Quality Index. The index ranges from 1 (very poor lake water quality) to 100 (excellent lake water quality).
The index is portrayed in three different ways: as a number ranging between 1 and 100 in the circle marked LWQI, and by a color and position on the sheet edge scale. The purpose of the sheet-edge scale is to review quickly large numbers of lakes or sample sites within a lake and determine how the quality of the various lakes or sites compare.

The position of the lines on the thermometer rating scales permits determination of the parameter (or parameters) which cause the index to be depressed. The lower the line, the greater the problem. A glance at the top of the problem rating scale identifies the test and the test results. The rating scales also permit determination of what test results would be considered excellent in terms of lake water quality by the panel of experts surveyed. They are the numbers on the outside the thermometers, near the top.

HENDERSON LAKE WATER QUALITY INDICES CALCULATION SHEETS

Four water quality index calculation sheets are included in this report. Three of the four are from each sample site for the sampling date. The other is an averaged sheet for the sampling date.

Matthew Novotny
Lakes Manager
Savin Lake Services
Hale, Michigan
December 2016

Henderson Lake Water Quality Data

Date	Sample Station Number	Temperature (ºC)	Dissolved Oxygen		Chlorophyll α (ug/L)	Secchi Disk Depth (ft)	Total Nitrate Nitrogen (ug/L)	Alkalinity (mg/L)	pH	Conductivity umhos per cm at 25 ºC	Total Phosphorus (ug/L)	Lake Water Quality Index	Grade
			(mg/L)	Percent Saturation									
9/28/2016	1	17.2	9.77	101.2	2.0	Bottom	ND	78	8.14	220	<9	98	A
9/28/2016	2	17.0	9.58	99.3	2.4	11.0	ND	77	8.02	220	<9	94	A
9/28/2016	3	16.9	9.62	99.7	2.1	10.0	ND	78	8.02	220	<9	93	A

Wallace E. Fusilier, Ph.D. is a highly regarded consulting limnologist. Information and styling found within this report are the result of Fusilier's dedication and professionalism as a limnologist.

CALCULATION SHEET FOR THE UNWEIGHTED MULTIPLICATIVE LAKE WATER QUALITY INDEX

W. Fusilier, Ph.D.

Temperature in degrees C	Dissolved Oxygen, % Saturation	Chlorophylla ug/L	Secchi Disk Depth in feet	Nitrate-N ug/L	Alkalinity mg/L	pH S.U.	Specific Conductivity umhos/cm @ 25C	Total Phosphorus ug/L
17.2	9.8 / 9.7 / 101.2	2.0	N/A	<60	78	8.14	220	<9

Ogemaw
County

Cummings
Township

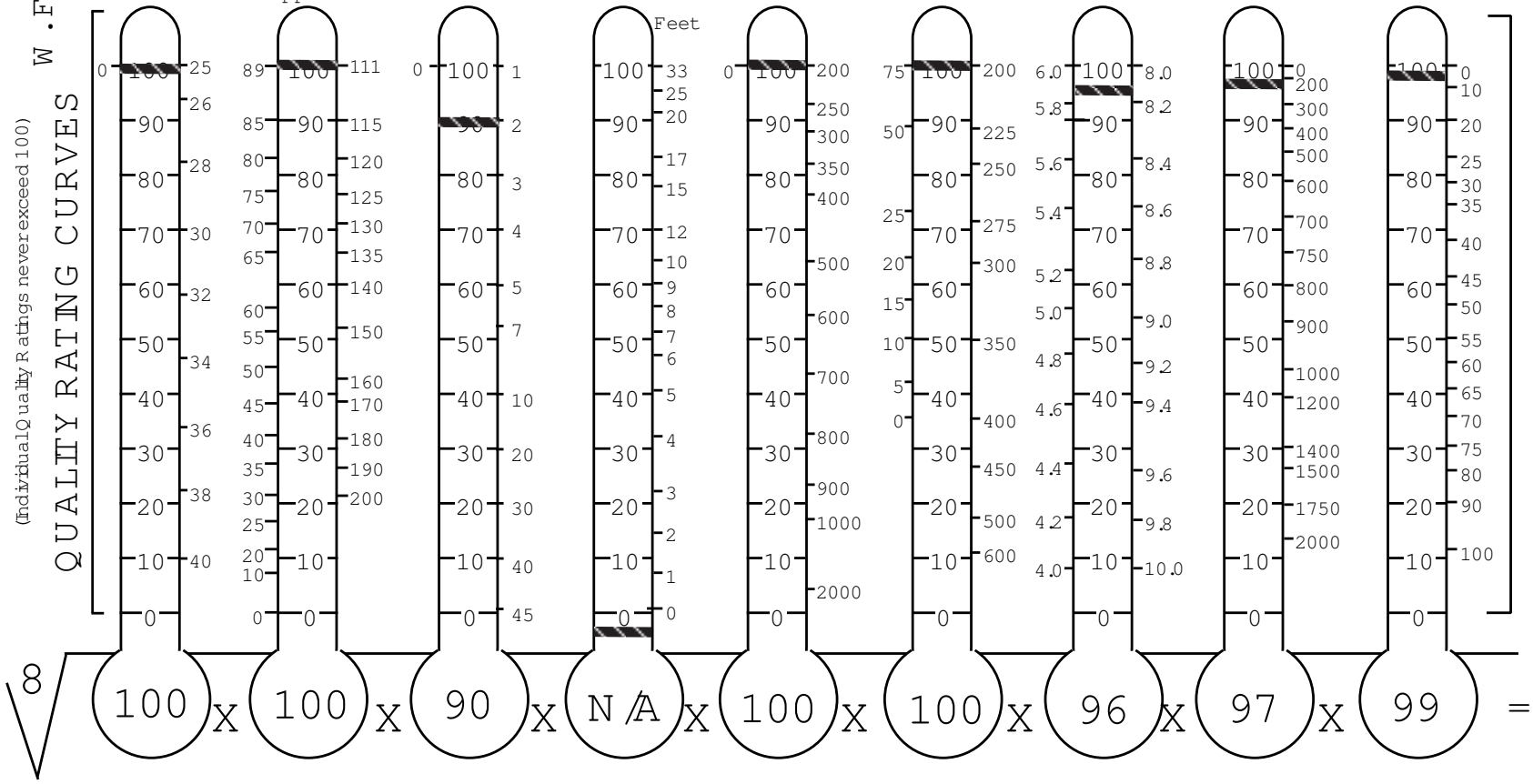
Savin

23 feet
Lake Depth

177 Acres
Lake Area

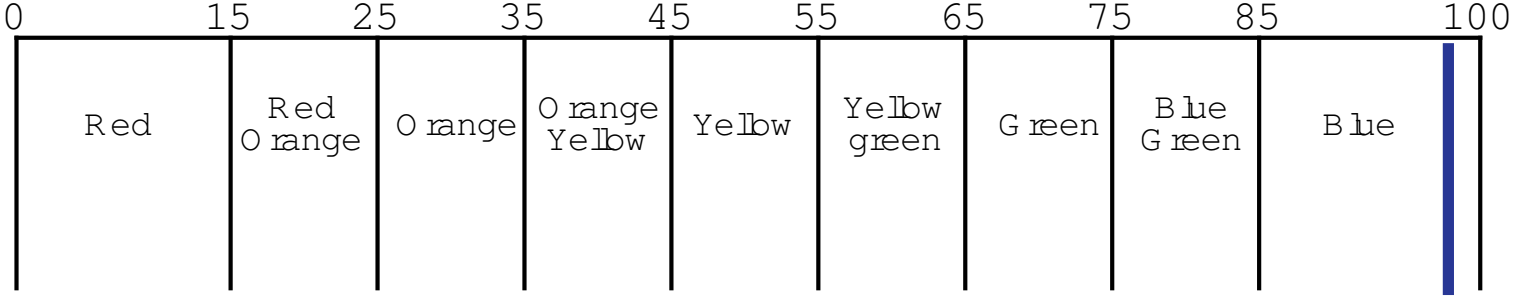
QUALITY RATING CURVES

(Individual Quality Ratings never exceed 100)



SET THE PARAMETER QUALITY RATING AT 1 IF THE EXTERNAL EXTREME VALUE RANGE IS EXCEEDED

LAKE WATER QUALITY INDEX



DATE 09/28/2016

STATION 1

LAKE Henderson Lake

CALCULATION SHEET FOR THE UNWEIGHTED MULTIPLICATIVE LAKE WATER QUALITY INDEX

W. Fusilier, Ph.D.

Temperature in degrees C	Dissolved Oxygen, % Saturation	Chlorophylla ug/L	Secchi Disk Depth in feet	Nitrate-N ug/L	Alkalinity mg/L	pH S.U.	Specific Conductivity umhos/cm @ 25C	Total Phosphorus ug/L
17.0	9.6 / 9.7 (99.3)	2.4	11.0	<60	77	8.02	220	<9

Ogemaw County

Cummings Township

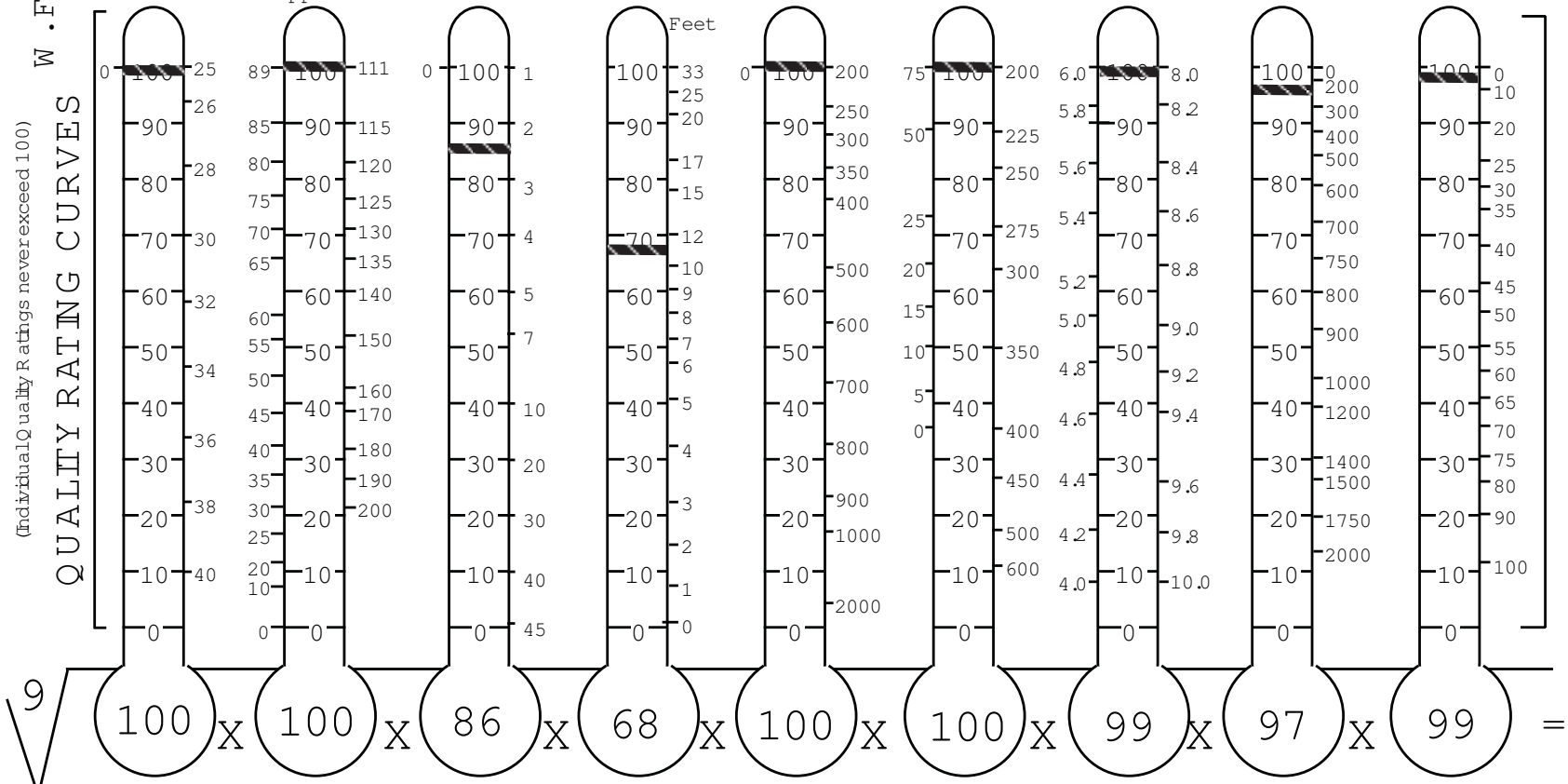
Savin

23 feet Lake Depth

177 Acres Lake Area

QUALITY RATING CURVES

(Individual Quality Ratings never exceed 100)

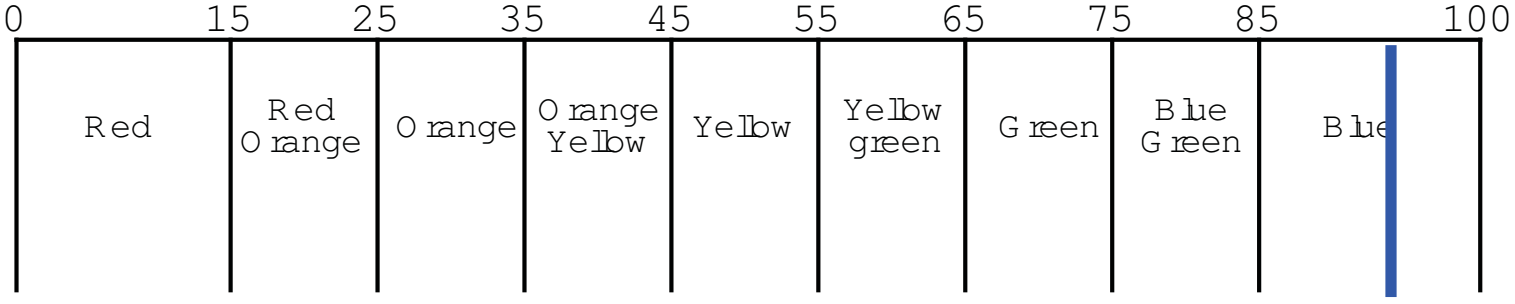


$$\sqrt[9]{100 \times 100 \times 86 \times 68 \times 100 \times 100 \times 99 \times 97 \times 99} = \text{LW QI} = 94$$

SET THE PARAMETER QUALITY RATING AT 1 IF THE EXTERNAL EXTREME VALUE RANGE IS EXCEEDED

LAKE WATER QUALITY INDEX

DATE 09/28/2016



STATION 2

LAKE Henderson Lake

CALCULATION SHEET FOR THE UNWEIGHTED MULTIPLICATIVE LAKE WATER QUALITY INDEX

W. Fusilier, Ph.D.

(Individual Quality Ratings never exceed 100)

QUALITY RATING CURVES

Temperature in degrees C	Dissolved Oxygen, % Saturation	Chlorophyll a ug/L	Secchi Disk Depth in feet	Nitrate-N ug/L	Alkalinity mg/L	pH S.U.	Specific Conductivity umhos/cm @ 25C	Total Phosphorus ug/L
16.9	9.6 / 9.7 / 99.7	2.1	10.0	<60	78	8.02	220	<9
	approx		Feet					

Ogemaw County

Cummings Township

Savin

23 feet Lake Depth

177 Acres Lake Area

$$\sqrt[9]{100 \times 100 \times 89 \times 64 \times 100 \times 100 \times 99 \times 97 \times 99} = \text{LW QI} = 93$$

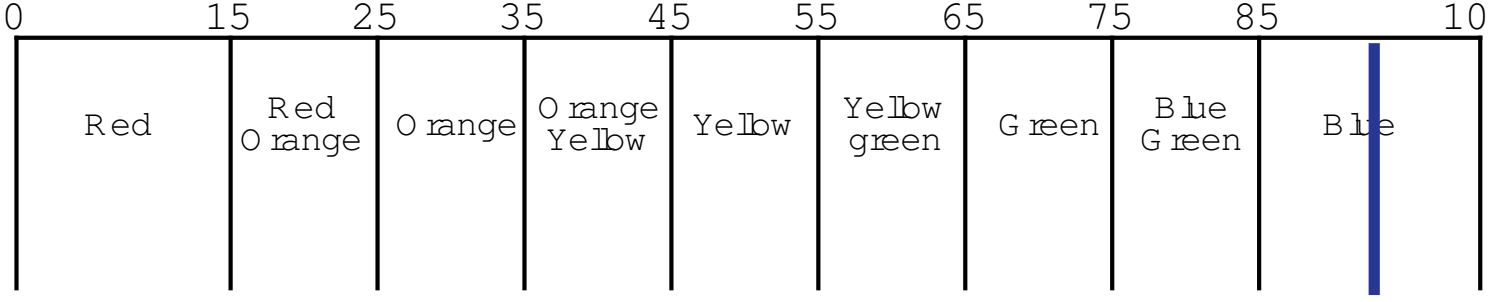
SET THE PARAMETER QUALITY RATING AT 1 IF THE EXTERNAL EXTREME VALUE RANGE IS EXCEEDED

LAKE WATER QUALITY INDEX

DATE 09/28/2016

STATION 3

LAKE Henderson Lake



CALCULATION SHEET FOR THE UNWEIGHTED MULTIPLICATIVE LAKE WATER QUALITY INDEX

W. Fusilier, Ph.D.

Temperature in degrees C	Dissolved Oxygen, % Saturation	Chlorophylla ug/L	Secchi Disk Depth in feet	Nitrate-N ug/L	Alkalinity mg/L	pH S.U.	Specific Conductivity umhos/cm @ 25C	Total Phosphorus ug/L
17.0	9.7 / 9.7 — 100.1	2.2	10.5	<60	78	8.06	220	<9

Ogemaw County

Cummings Township

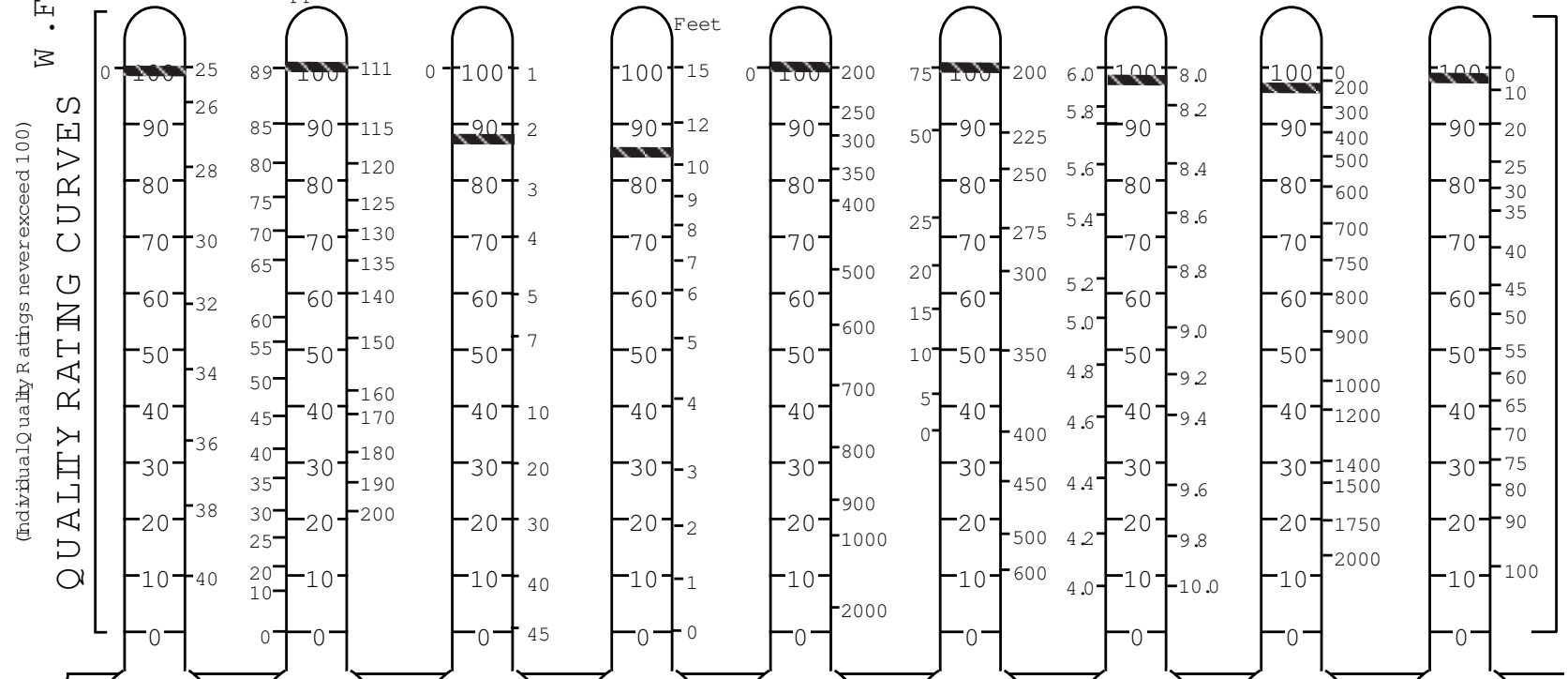
Savin Analyst

23 feet Lake Depth

177 Acres Lake Area

QUALITY RATING CURVES

(Individual Quality Ratings never exceed 100)

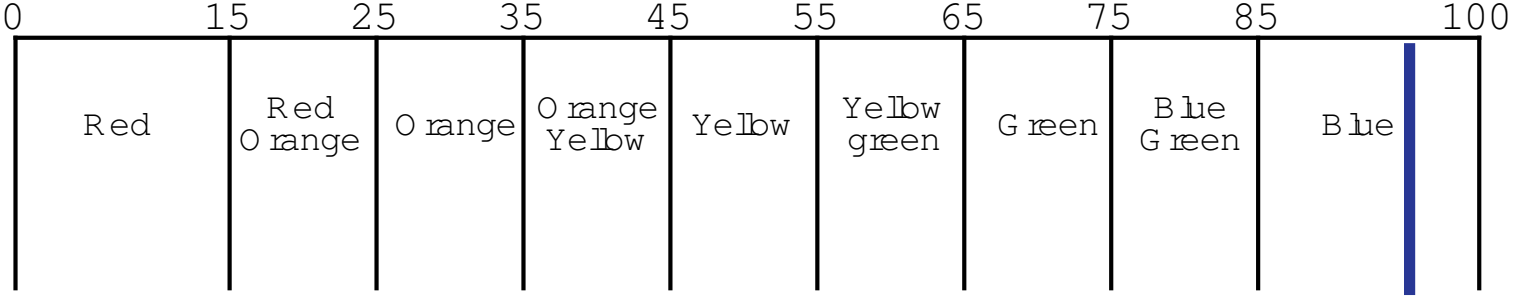


$$\sqrt[9]{100 \times 100 \times 88 \times 85 \times 100 \times 100 \times 98 \times 97 \times 99} = \text{LW QI} = 96$$

SET THE PARAMETER QUALITY RATING AT 1 IF THE EXTERNAL EXTREME VALUE RANGE IS EXCEEDED

LAKE WATER QUALITY INDEX

DATE 09/28/2016



STATION Average

LAKE Henderson Lake