

December 4, 2018

Henderson Lake Bill Olsen – Primary Contact 2937 Chippewa Trail Lupton MI. 48635

Henderson Lake Property Owners:

It has been a pleasure managing Henderson Lake this past summer. Every year seems to bring a unique set of challenges and we welcome the opportunity to meet these challenges every single year. We hope that you continue to feel that your lake was managed professionally, economically, and effectively.

As we say goodbye to 2018 and look forward to 2019, we are happy to give you a brief re-cap of the work that was performed on Henderson Lake by Savin Lake Services. This year, we treated Henderson Lake (4) times, and conducted water quality tests as follows:

On June 18th, 2018 Savin Lake Services conducted our initial application for non-native and exotic plant communities. Also, during this treatment, we did an algae spray of the entire developed shoreline, treated native vegetation in the canal and around riparian owners' docks where needed, and a Bacterial Augmentation (Mukk Buster) treatment also took place at this time. Savin Lake Services treated less than an acre of Milfoil, 7.5 acres of Curly Leaf Pondweed,10 acres of the lake for algae, and 10 acres of the lake for various pondweeds like Richardson's Pondweed and Baby Pondweed. There was also 5 acres of the lake that received a Mukk Buster treatment.

On June 22nd, 2018 Savin Lake Services sent our Aquatic Biologist out to Henderson Lake to pull a Spring set of water samples. During our telephone conference call in 2017 with myself, Bill Olsen, and many other HLMC Members, it was decided that we would continue to do some water quality studies to ensure the lake was in a healthy state. There were (3) sites that we obtained water samples from on the lake, with one of the sites being the deepest spot in the lake. These samples and sites were analyzed for the following: Temperature, Dissolved oxygen, Total Alkalinity, Total Phosphorus, Total Nitrate Nitrogen, Conductivity, pH (surface only), Chlorophyll a (surface only), and a Secchi disk depth.

On July 24th, 2018 Savin Lake Services conducted our 2nd herbicide application, algae spray, and bacterial augmentation treatment. Savin Lake Services treated 5 acres of Naiad, and Curly Leaf Pondweed, 17.5 acres of various pondweeds like Richardson's Pondweed and Large Leaf Pondweed, and 15 acres of algae control. We only treated the areas that we felt needed to be treated



in developed shoreline areas, in and around docks, and navigational boat ways. During this treatment, we conducted a bacterial augmentation treatment in the same 5 acres we did during our first application.

On August 22nd, 2018 Savin Lake Services conducted another herbicide treatment, algae spray, and Bacterial Augmentation treatment. This treatment consisted of an algae spray in the developed areas of the lake where needed (15 acres), and 10 acres of herbicide treatment for returning Milfoil (very little) and Elodea. We also did a herbicide treatment for 15 acres of Richardson's Pondweed, and treated 40X40 areas for lily pads, and Cattails where treatment was needed. Also, during this time, we did the last Bacterial Augmentation treatment for the year in the same areas as we did earlier this year, as it requires 3 treatments per season for the greatest efficacy.

Our final application for the season was conducted on September 25th, 2018. This application was just a touch up application for algae and residual pondweeds. We treated 7.5 acres of the lake for algae, and 5 acres of the lake for residual pondweeds.

On October 2nd, 2018 Savin Lake Services sent our Aquatic Biologist out to Henderson Lake again to pull a fall set water samples in the same 3 sites for testing. After all the Spring and Fall samples are analyzed there was a report generated and made available for the Henderson Lake Property Owners. This report has already been completed. You can obtain this report by contacting myself, Bill Olsen, or any members of the weed control committee. We would like approval to plan to continue with the Water Quality testing through the 2019 season as it will give us a vast amount of information that we can use for a trending analysis.

Every treatment, survey, or water quality study that Savin Lake Services has performed, we have provided maps, or generated full reports on the work completed. All treatment maps and reports for 2015, 2016, 2017, and 2018 are available for you to view on your website HendersonLake.org.

Please keep in mind that we are a fully integrated lakes management company offering solutions including but not limited to mechanical harvesting, herbicide control, dredging, bio-augmentation, and aeration. Savin Lake Services also offers a complete range of water quality testing, depth contour mapping, individual property solutions, and even aquatic plant density reporting.

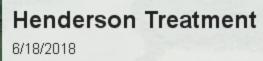
We look forward to serving you in 2019. We plan to keep the same treatment plans for this year as we did last year. We see no reason to deviate from the current treatment plan. We feel this plan is the most effective and efficient way to keep Henderson Lake clean and desirable for use by the Henderson Lake residents. Until then; if you have any questions, comments, or require additional information, feel free to contact us.



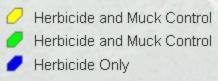
Sincerely,

Paul Bache

Paul Barber - Regional Lakes ManagerSavin Lake Services Inc.3088 Hottis RoadHale, Michigan 48739Hale Office: 989-728-2200Clare Office: 989-386-0600Toll Free: 877-SAV-LAKEFax: 989-516-5900Email: Paulbarber@LakeAndPond.com



Legend



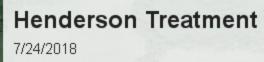
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2000 ft

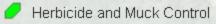
Henderson Lake

Google Earth

© 2018 Google







/ Herbicide Only

2000 ft

🧷 Herbicide and Muck Control

A N

Henderson Lake

Google Earth

© 2018 Google

Henderson Lake treatment map

September 25th, 2018



Gnipp

Pine-Ave-







Algae and Herbicide control Algae control

Peters Rd

eeth-

21

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Henderson Lake Treatment Map 8-22-18

Write a description for your map.

Henderson Lake Henderson Lake

Pine Ave

the bridge of the second

Google earth

6350

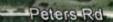
© 2018 Google





21

Herbicide Control Muck and Herbicide control





1000 ft



3088 Hottis Rd. Hale, MI 48739 Hale: 989.728.2200 Clare: 989.386.0600 Fax: 989.516.5900 Henderson Lake 2018 Water Quality Report

Summary:

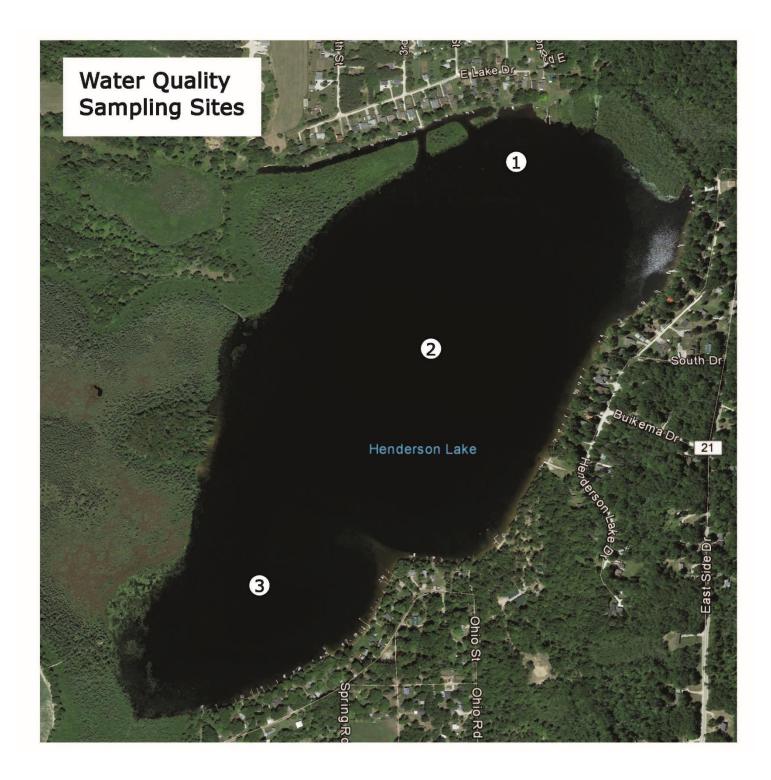
Water Quality Testing was completed 2 times on Henderson Lake in 2018 at 3 different locations around the lake. Of the parameters tested, Temperature, Dissolved Oxygen, Secchi Disk, and pH were sampled while on the lake. Chlorophyll α , Nitrate-N, Phosphorus, Alkalinity, and Conductivity were sampled by sending the water in sample bottles to an independent laboratory, White Water Associates located in Amasa, MI, where the analysis was ran.

A well known limnologist named Wally Fusilier developed a grading scale for various parameters of water quality. Data collected in 2018 is shown below and given a grade based on Fusilier's scale. Additionally, historical data and parameter descriptions are provided at the end of this report.

Because herbicide treatment of aquatic vegetation has occurred on Henderson Lake, it should be noted that the application of herbicide no direct impact to the water quality of Henderson Lake.

Overall in 2018 based on the analysis results, Henderson Lake had excellent water quality figures again. In both the spring and the fall Henderson Lake had average grades of A. Fall chlorophyll a values were slightly high. But other than that the lake is great!







2018 Results:

122

77-69; 122-

131

69-62; 131-

140

<62; >140

3-4

4-5

>5

16-12

12-9

<9

С

D

F

28.5-30

30-31.5

>31.5

Date 6/22/2018			6/22/2018			6/22/2018			10/2/2018			10/2/2018			10/2/2018			
Station Numbe		1			2			3			1			2			3	
Temp (º	C)	25.	3	Α	24.7	Α	2	4.8	А		15.8	Α		15.3	А		15.1	Α
Dissolve Oxyger (mg/L)	ר)	8.9			9.20			0.03			10.16			10.03			9.89	
Dissolve Oxyger (%saturat	า	108	.0	A	111.7	A	10	09.6	A		103.1	A		99.6	A		98.2	A
Chlorophyll a (ug/L)		1.4	1	A	1.4	Α	-	1.0	A		2.1	В		2.4	В		3.8	С
Secchi Disk Depth (ft)		Botte	om	D	11.0	D	Во	ttom	D		Bottom	С		15.0	С		Bottom	С
Total Nitrate Nitrogen (ug/L)		<8	0	A	<80	A	<	<80	A		<80	A		<80	A		<80	A
Alkalinity (mg/L)		79.	0	A	80	A		81	A		74	A		72	A		72	A
рН	pH		8.2		8.21	A 8.14		8.14	4 A		8.20	A C		7.82	Α		7.83	Α
Conductivity (umhos/cm)		200	0.0	A	200	A	2	200	A		200	A		200	A		200	A
Total Phosphorus (ug/L)		8>	3	A	<8	A		12	A		<8	A		<8	A		<8	A
Overall Grade				A		A			A			A			A			A
	S	cale:			•					·						·		
Grade	Grade Ten		p Dissolved Oxygen		Chloro- phyll α			Total Nitrate Nitrogen		Alkalinity	рH	рН		Conduc- tivity		Total Phosphorus		
Α	A 0-1		5.5 85-115		0-2	>	>19		0-275		50-225	5.75-8.27		5-8.27	0-380		0-20	
В	B 26.5-28.5		85-7	7; 115-	2-3	19	9-16	.6 275-		_	50-35;	5.75-5.55;		5-5.55;	380-590		20-28	

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360-450

450-540

>540

225-255

35-23;

255-280

23-17;

280-310

<17; >310

8.27-8.47

5.55-5.33;

8.47-8.69

5.33-5.14;

8.69-8.88

<5.14; >8.88

28-39

39-46

>46

590-720

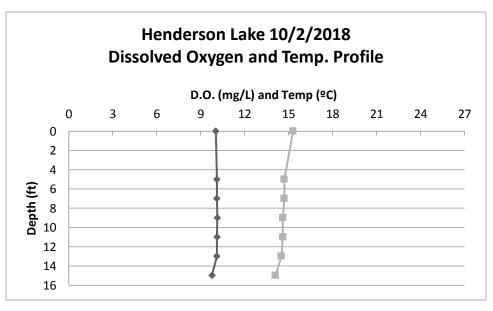
720-800

>800



Dissolved Oxygen and Temp. Profile

D.O.	Depth				
(mg/L)	(ft)				
10.03	0				
10.10	5				
10.10	7				
10.13	9				
10.11	11				
10.10	13				
9.76	15				
	(mg/L) 10.03 10.10 10.10 10.13 10.11 10.10				



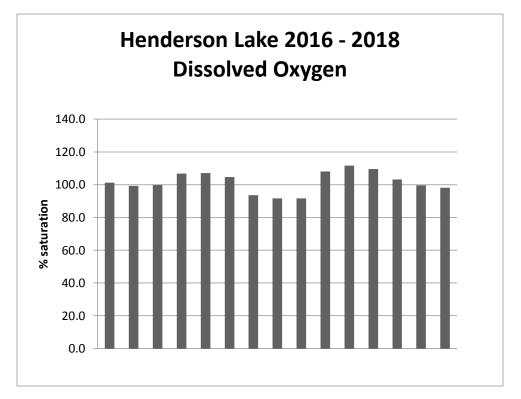
Matt Novotny

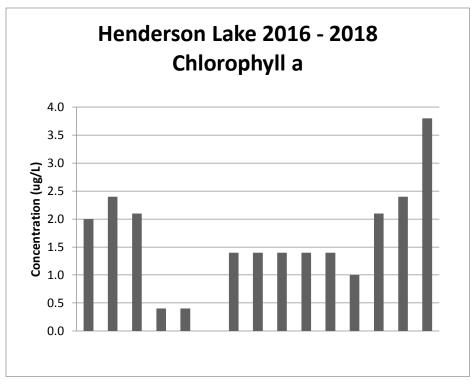
atthew Novotny

Environmental Scientist



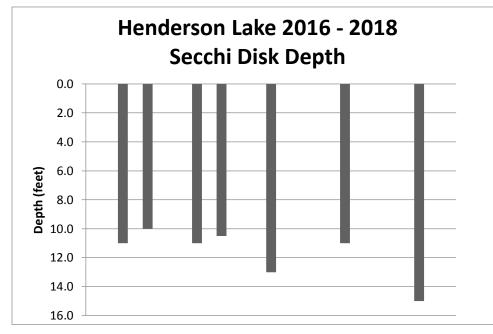
Historical Data:

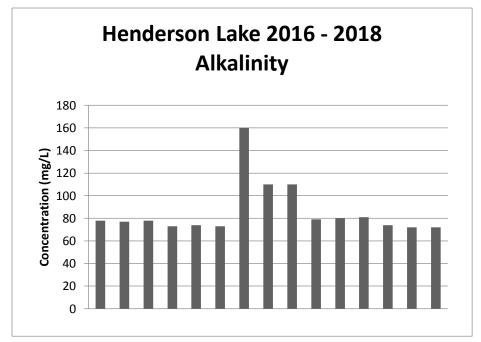




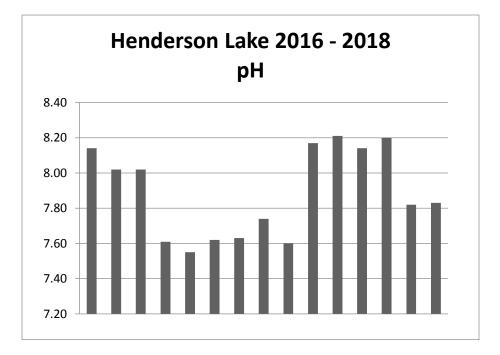


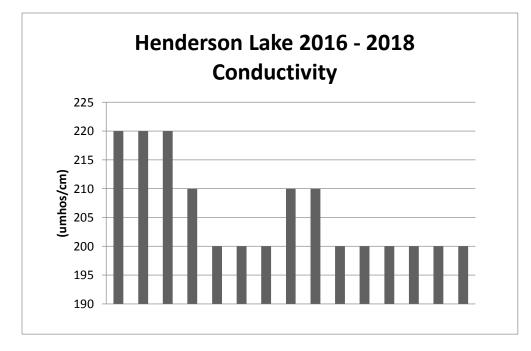
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(Nitrate and Total Phosphorus not shown due to majority of undefined values)



					Henderson	Lake Water	Quality Dat	a ,				
Date	105	Temperat ure (ºC)	Dissolv	ed Oxygen	Chlorophyli a (ug/L)	Secchi Disk Depth (ft)	Total Nitrate Nitrogen (ug/L)		рH	Conductivity umhos per cm at 25 ºC	Total Phosphor us (ug/L)	Grade
	Sample Station Number		(mg/L)	Percent Saturation				Alkalinity (mg/L)				
9/28/2016	1	17.2	9.77	101.2	2.0	Bottom	ND	78	8.14	220	<9	А
9/28/2016	2	17.0	9.58	99.3	2.4	11.0	ND	77	8.02	220	<9	А
9/28/2016	3	16.9	9.62	99.7	2.1	10.0	ND	78	8.02	220	<9	А
6/7/2017	1	23.4	9.14	106.8	0.4	Bottom	ND	73	7.61	210	13	А
6/7/2017	2	23.1	9.17	107.1	0.4	11.0	ND	74	7.55	200	13	А
6/7/2017	3	22.4	9.13	104.7	0.0	10.5	ND	73	7.62	200	41	А
10/4/2017	1	14.0	9.64	93.6	1.4	Bottom	ND	160	7.63	200	ND	А
10/4/2017	2	14.3	9.44	91.7	1.4	13.0	ND	110	7.74	210	ND	А
10/4/2017	3	14.1	9.43	91.6	1.4	Bottom	ND	110	7.60	210	ND	А
6/22/2018	1	25.3	8.90	108.0	1.4	Bottom	<80	79.0	8.17	200.0	<8	А
6/22/2018	2	24.7	9.20	111.7	1.4	11.0	<80	80.0	8.21	200.0	<8	А
6/22/2018	3	24.8	9.03	109.6	1.0	Bottom	<80	81.0	8.14	200.0	12.0	А
10/2/2018	1	15.8	10.2	103.1	2.1	Bottom	<80	74.0	8.2	200.0	<8	А
10/2/2018	2	15.3	10.0	99.6	2.4	15.0	<80	72.0	7.8	200.0	<8	А
10/2/2018	3	15.1	9.9	98.2	3.8	Bottom	<80	72.0	7.8	200.0	<8	А



Parameter Descriptions: TEMPERATURE AND DISSOLVED OXYGEN

Temperature exerts a wide variety of influences on most lakes, such as the separation of layers of water (stratification), solubility of gases, and biological activity.

Dissolved oxygen is the parameter most often selected by lake water quality scientists as being important. Besides providing oxygen for aquatic organisms in natural lakes, dissolved oxygen is involved in phenomena such as phosphorus precipitation to, and release from, the lake bottom sediments and decomposition of organic material in the lake.

Low dissolved oxygen concentrations (below 4 milligrams per liter) are generally insufficient to support fish life. In most Michigan lakes, there is no dissolved oxygen below the thermocline in late summer. Some experts like to see some dissolved oxygen in the bottom water of a lake, even if it is almost zero. This is because as long as there is some dissolved oxygen in the water at the bottom of the lake, phosphorus precipitated by iron to the bottom sediments will remain there. Once a lake runs out of dissolved oxygen in the water at the bottom iron comes back into solution. When that happens, it releases the phosphorus back into the water. This can cause additional algae to grow when the lake mixes.

DISSOLVED OXYGEN, PERCENT SATURATION

Because the amount of dissolved oxygen a water can hold is temperature dependent with cold water holding more than warm water, dissolved oxygen saturation is often a better way to determine if oxygen supplies are adequate. The best is between 90 and 110 percent.



Chlorophyll α is used by lake scientists as a measure of the biological productivity of the water. Generally, the lower the chlorophyll α , the better. High concentrations of chlorophyll α are indicative of an algal bloom in the lake, an indication of poor lake water quality. The highest surface chlorophyll α concentration found by Wallace Fusilier (Water Quality Investigators, WQI) in a Michigan lake was 216 micrograms per liter. Best is below one microgram per liter.

SECCHI DISK TRANSPARENCY (originally Secchi's disk)

In 1865, Angelo Secchi, the Pope's astronomer in Rome, Italy devised a 20centimeter (8 inch) white disk for studying the transparency of the water in the Mediterranean Sea. Later an American limnologist (lake scientist) named Whipple divided the disk into black and white quadrants which many are familiar with today.

The Secchi disk transparency is a lake test widely used and accepted by limnologists. The experts generally felt the greater the Secchi disk depth, the better quality the water. However, one Canadian scientist pointed out acid lakes have very deep Secchi disk readings. (Would you consider a very clear lake a good quality lake, even if it had no fish in it? It would be almost like a swimming pool.) Most lakes in southeast Michigan have Secchi disk transparencies of less than ten feet. On the other hand, Elizabeth Lake in Oakland County had 34 foot Secchi disk readings in summer 1996, evidently caused by a zebra mussel invasion a couple of years earlier.

Most limnology texts recommend the following: to take a Secchi disk transparency reading, lower the disk into the water on the shaded side of an anchored boat to a point where it disappears. Then raise it to a point where it's visible. The average of these two readings is the Secchi disk transparency depth.



3088 Hottis Rd. Hale, MI 48739 Hale: 989.728.2200 Clare: 989.386.0600 Fax: 989.516.5900 Secchi disk measurements should be taken between 10 AM and 4 PM. Rough water will give slightly shallower readings than smooth water. Sunny days will give slightly deeper readings than cloudy days. However, roughness influences the visibility of the disk more than sunny or cloudy days.

TOTAL PHOSPHORUS

Although there are several forms of phosphorus found in lakes, the experts selected total phosphorus as being most important. This is probably because all forms of phosphorus can be converted to the other forms. Currently, most lake scientists feel phosphorus, which is measured in parts per billion (1 part per billion is one second in 31 years) or micrograms per liter (ug/L), is the one nutrient which might be controlled. If its addition to lake water could be limited, the lake might not become covered with the algal communities so often found in eutrophic lakes.

Based on WQI's studies of many Michigan inland lakes, they've found many lakes were phosphorus limited in spring (so don't add phosphorus) and nitrate limited in summer (so don't add nitrogen).

10 parts per billion is considered a low concentration of phosphorus in a lake and 50 parts per billion is considered a high value in a lake by many limnologists.

NITRATE NITROGEN

Nitrate, also measured in the parts per billion range, has traditionally been considered by lake scientists to be a limiting nutrient. The experts felt any concentration below 200 parts per billion was excellent in terms of lake water quality. The highest value found by Fusilier was 48,000 parts per billion in an Ottawa County river which flowed into Lake Macatawa in Holland, Michigan.



3088 Hottis Rd. Hale, MI 48739 Hale: 989.728.2200 Clare: 989.386.0600 Fax: 989.516.5900 On the other hand, WQI has studied hundreds of Michigan inland lakes, and many times they find them nitrate limited (very low nitrate nitrogen concentrations), especially in summer.

WQI was finding many lakes have lower nitrate nitrogen concentrations in summer than in spring. This is probably due to two factors. First, plants and algae growing in lakes as water warms can remove nitrates from the water column. And second, bacterial denitrification (where nitrates are converted to nitrogen gas by bacteria) also occurs at a much faster rate in summer when the water is warmer.

Generally limnologists feel optimal nitrate nitrogen concentrations (which encourage maximum plant and algal growth) are about 10-20 times higher than phosphorus concentrations. The reason more nitrogen than phosphorus is needed is because nitrogen is one of the chemicals used in the production of plant proteins, while phosphorus is used in the transfer of energy, but is not used to create plant material. If the nitrate concentration is less than 10-20 times the phosphorus concentration, the lake is considered nitrogen limited. If the nitrate concentration is higher than 10-20 times the phosphorus concentration, the lake is considered phosphorus limited.

TOTAL ALKALINITY

Alkalinity is a measure of the ability of the water to absorb acids (or bases) without changing the hydrogen ion concentration (pH). It is, in effect, a chemical sponge. In most Michigan lakes, alkalinity is due to the presence of carbonates and bicarbonates which were introduced into the lake from ground water or streams which flow into the lake. In lower Michigan, acidification of most lakes should not be a problem because of the high alkalinity concentrations.



3088 Hottis Rd. Hale, MI 48739 Hale: 989.728.2200 Clare: 989.386.0600 Fax: 989.516.5900 HYDROGEN ION CONCENTRATION (pH)

pH has traditionally been a measure of water quality. Today it is an excellent indicator of the effects of acid rain on lakes. About 99% of the rain events in southeastern Michigan are below a pH of 5.6 and are thus considered acid. However, there seems to be no lakes in southern Michigan which are being affected by acid rain. Most lakes have pH values between 7.5 and 9.0.

SPECIFIC CONDUCTIVITY

Conductivity, measured with a meter, detects the capacity of a water to conduct an electric current. More importantly however, it measures the amount of materials dissolved in the water, since only dissolved materials will permit an electric current to flow. Theoretically, pure water will not conduct an electric current. It is the perception of the experts that poor quality water has more dissolved materials than does good quality water