

# EVALUATION OF CHEMICAL INJECTION METHODS FOR HYDRATE MITIGATION

Presentation By:

Aarushi Gupta, Abhinav Singh  
B. Tech Applied Petroleum Engineering  
University of Petroleum and Energy  
Studies, Dehradun

# OBJECTIVES

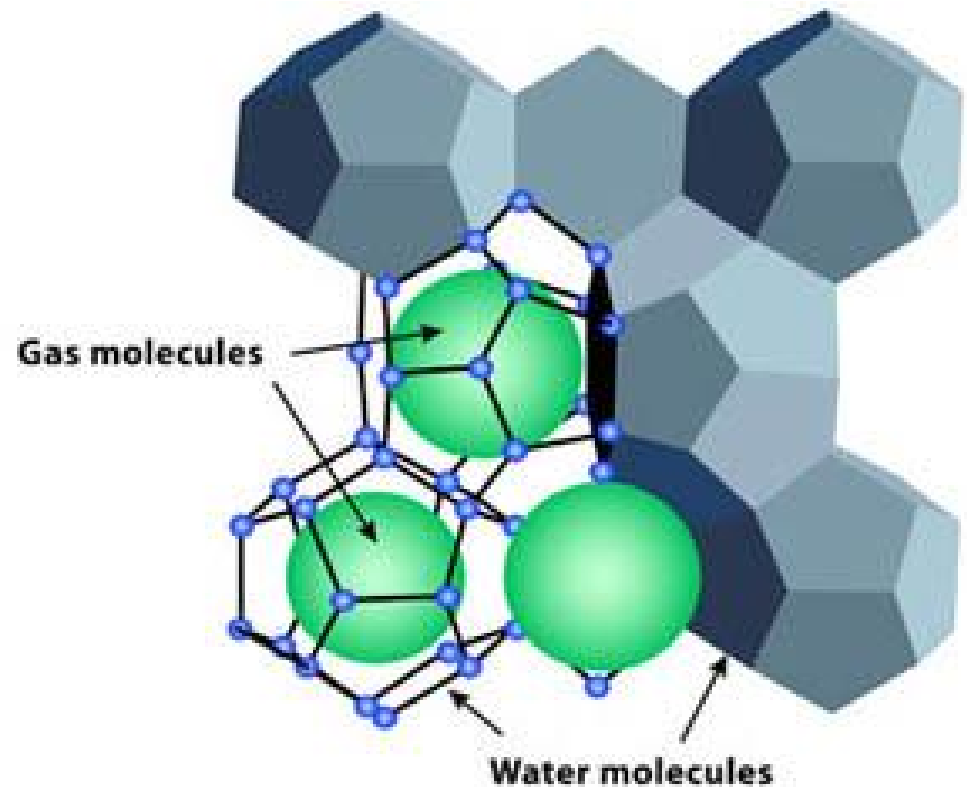
- I. Understanding Thermodynamic Inhibition as a Hydrate Mitigation Technique with the help of a Case Study.
- II. Analyzing the results for the Case Study to understand Kinetic Inhibition as a Hydrate Mitigation Technique.
- III. Analyzing the results for the Case Study to understand Anti- Agglomerants as a Hydrate Mitigation Technique.
- IV. Conclusion

# I. CASE STUDY

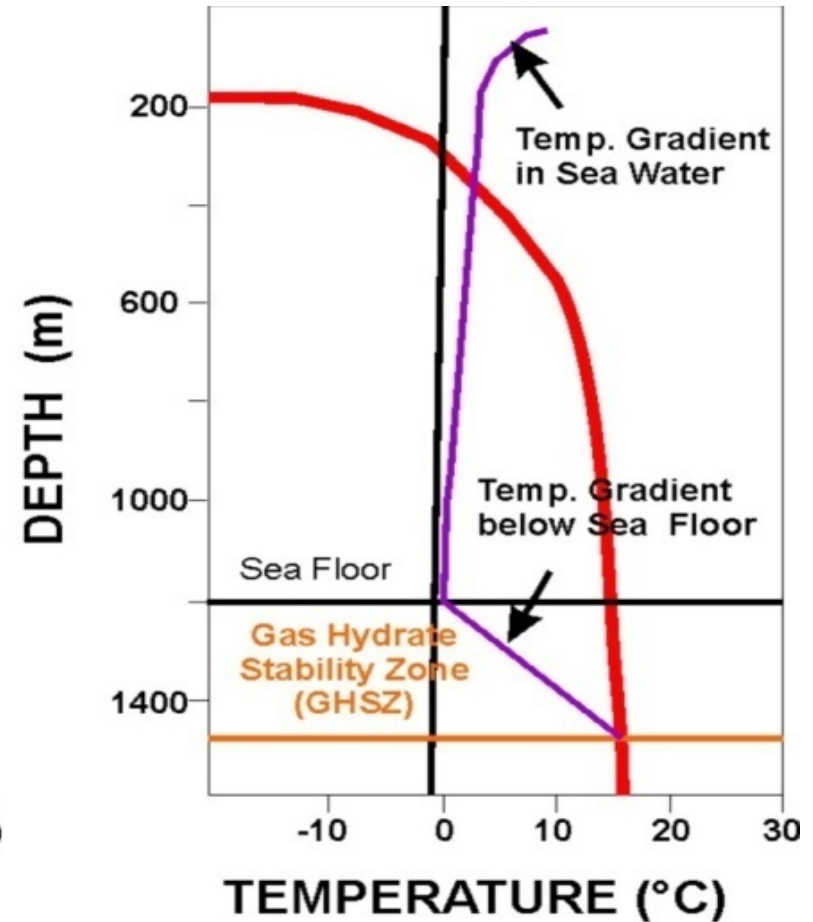
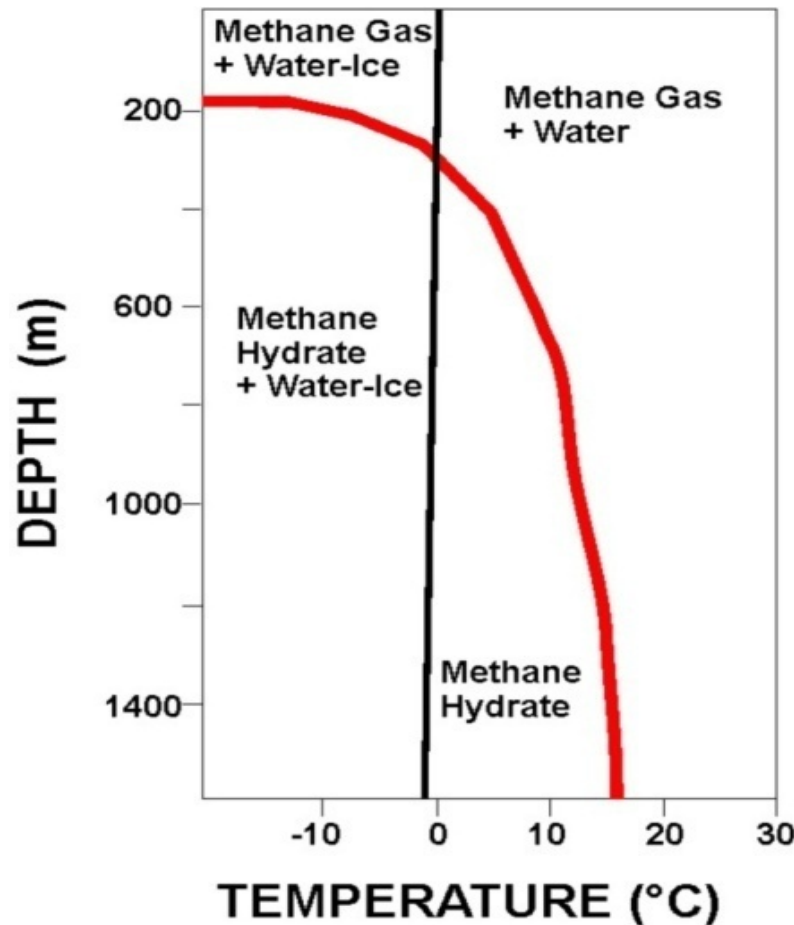
- Polarrev Field, Arctic.
- MEG was chosen as the Inhibitor.
- Used because of its lower cost, lower viscosity and lower solubility in liquid hydrocarbons, and lower vapour pressure giving reduced gas-phases losses.
- 100% pure MEG.
- The total amount of produced water at standard conditions is 518.4 m<sup>3</sup>/day.

# I.I. NATURAL GAS HYDRATES

- Formation
- Structure
- Characteristics
- Potential
- Problems

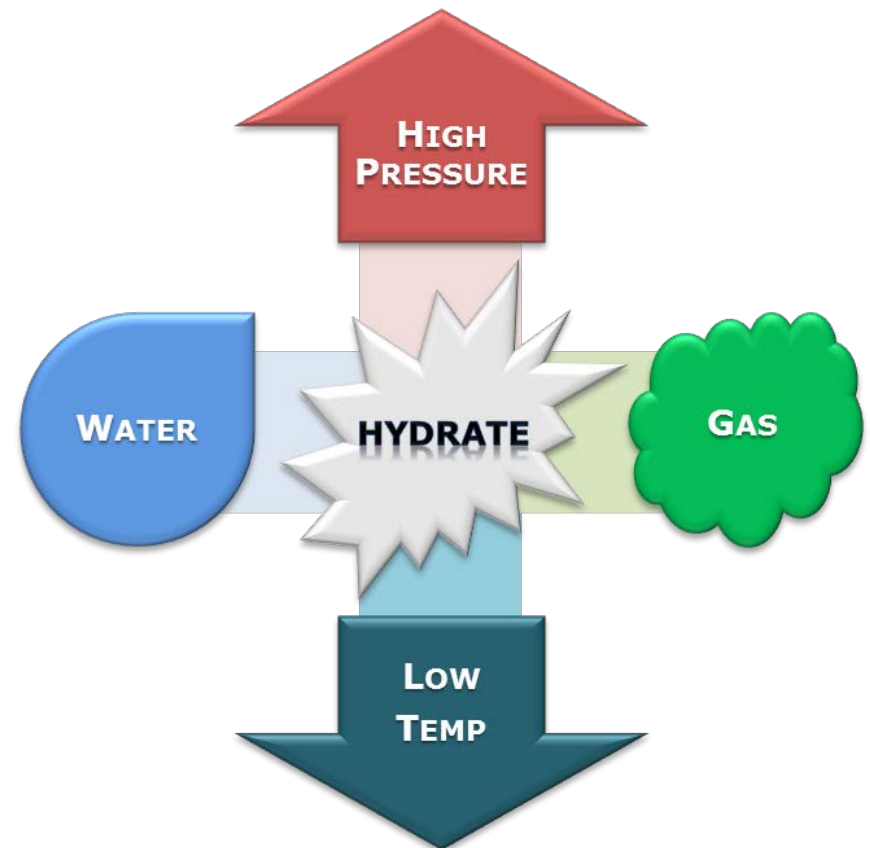


# I.I.a. FORMATION



## I.II. MITIGATION METHODS

- Pressure Control
- Temperature Control
- Remove Supply of Water
- Inject Chemical Inhibitors



## I.III. THERMODYNAMIC INHIBITORS

- Commonly use are Methanol and Monoethylene Glycol.
- May be recovered and re-circulated.
- Economics of recovery is not favorable in most cases.
- Different chemicals used as per temperature conditions of the area.

# I.III. THERMODYNAMIC INHIBITORS (Contd.)

ADVANTAGES	DISADVANTAGES
<ul style="list-style-type: none"> <li>• Proven track-records</li> <li>• Well understood technology</li> <li>• Effective technology</li> <li>• Both oil and gas production</li> <li>• Can be used at all water cuts</li> <li>• Recoverable</li> <li>• Reduced risk of corrosion</li> </ul>	<ul style="list-style-type: none"> <li>• High concentration required</li> <li>• Discharge limitations</li> <li>• Toxic in large amount</li> <li>• Large amount of heat required for regeneration</li> <li>• High CAPEX</li> <li>• High OPEX</li> </ul>



# I. CASE STUDY

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- 100% pure MEG.
- The total amount of produced water at standard conditions is 518.4 m<sup>3</sup>/day.

# I. CASE STUDY (Contd.)

- The amount of MEG required to treat the water phase is 51%.
- The inhibitor losses to the hydrocarbon liquid phase is assumed to be 35% of the amount of ethylene glycol required to treat the water phase i.e. 17.85% wt.
- Thus, the total amount of MEG required is 68.85% wt.

# I. CASE STUDY (Contd.)

- Injection of MEG=  $1.15 \times 10^6$  kg/day.
- Cost of MEG is USD 1.26/kg.
- Total Cost=  $1.26 \times 1.15 \times 10^6$   
= 1, 450, 840 USD/Day

## II. USING KINETIC INHIBITORS IN ABOVE CASE

- Say, we used Poly N-vinyl Pyrrolidone.
- 100% Pure.
- The total amount of produced water at standard conditions is 518.4 m<sup>3</sup>/day.
- The amount of PNP required to treat the water phase is 0.46%.
- The inhibitor losses to the hydrocarbon liquid phase is assumed to be 30% of the amount of PNP required to treat the water phase i.e. 0.138% wt.

## II.I. KINETIC INHIBITORS

- New and evolving technology.
- Requires extensive tests and optimisation to the actual system.
- Works by slowing down the kinetics of the nucleation.
- Low-Dosage-Hydrate-Inhibitors.
- Usually polymers or copolymers.

# II.I. KINECTIC INHIBITORS (Contd.)

ADVANTAGES	DISADVANTAGES
<ul style="list-style-type: none"> <li>• Effective Operation under severe temperature conditions.</li> <li>• Increased stabilization from reduced requirement for pipeline depressurization during cold standby.</li> <li>• Low costs.</li> <li>• Low toxicity due to lower injection amounts.</li> </ul>	<ul style="list-style-type: none"> <li>• Their performance under shut-in conditions.</li> <li>• Hydrate formation at the top of a pipeline under stratified flow.</li> </ul>

## II. USING KINETIC INHIBITORS IN ABOVE CASE

- Say, we used Poly N-vinyl Pyrrolidone.
- 100% Pure.
- The total amount of produced water at standard conditions is 518.4 m<sup>3</sup>/day.
- The amount of PNP required to treat the water phase is 0.46%.
- The inhibitor losses to the hydrocarbon liquid phase is assumed to be 30% of the amount of PNP required to treat the water phase i.e. 0.138% wt.

## II. USING KINETIC INHIBITORS IN ABOVE CASE

- Thus, the total amount of PNP required is 0.598% wt.
- Injection of PNP =  $3.12 \times 10^3$  kg/day
- Total Price =  $\$ 7.01 \times 3.12 \times 10^3$   
 $= \$ 21,861.95/\text{day}$



## III. USING ANTI- AGGLOMERANTS IN ABOVE CASE

- Say, we used Sodium Hydroxide.
- 100% Pure.
- The total amount of produced water at standard conditions is 518.4 m<sup>3</sup>/day.
- The amount of NaOH required to treat the water phase is 0.284%.
- The inhibitor losses to the hydrocarbon liquid phase is assumed to be 30% of the amount of NaOH required to treat the water phase i.e. 0.0852% wt.

## III.I ANTI- AGGLOMERANTS

- New and evolving technology.
- Requires extensive tests and optimisation to the actual system.
- Works by stopping the agglomeration of Gas Hydrate Crystals.
- Low-Dosage-Hydrate-Inhibitors.
- Usually polymers or copolymers.

# III.I ANTI- AGGLOMERANTS

ADVANTAGES	DISADVANTAGES
<ul style="list-style-type: none"> <li>• Low Dosage.</li> <li>• Effective at Severe Conditions.</li> <li>• Effective During Extended Shut-ins.</li> <li>• Total System Approach to Ensure Products are Compatible with Production System.</li> </ul>	<ul style="list-style-type: none"> <li>• Require liquid hydrocarbon phase and water cut &lt; 50-75%.</li> <li>• Do not protect gas phase.</li> <li>• May not be effective at temperatures below 38°F.</li> </ul>

## III. USING ANTI- AGGLOMERANTS IN ABOVE CASE

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- The total amount of produced water at standard conditions is 518.4 m<sup>3</sup>/day.
- The amount of NaOH required to treat the water phase is 0.284%.
- The inhibitor losses to the hydrocarbon liquid phase is assumed to be 30% of the amount of NaOH required to treat the water phase i.e. 0.0852% wt.

## III. USING ANTI- AGGLOMERANTS IN ABOVE CASE

- Thus, the total amount of PNVP required is 0.3692% wt.
- Injection of PNVP =  $1.92 \times 10^3$  kg/day = 1776.09 m<sup>3</sup>/day
- Total Price= \$  $1.63 \times 1.92 \times 10^3$   
= \$ 3,129.6/day

## IV. CONCLUSION

- Using Low- Dosage- Hydrate- Inhibitor in place of Thermodynamic Inhibitors is economical and environmentally safer.
- It might be so that we need to develop different AA's for different crudes.
- KI's have limited usage as they only work down to certain sub-coolings.

**THANK YOU!**

