Phytoremediation- A Green and Clean Technology for remediation of Industrial Contaminants

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Abstract - In world of science and technology, along the global development new challenges of environment protection and conservation has been ahead. Almost every country of world raised voice for sustainable development. For sustainable development of a country, soil and water resources most valuable resources which are unfortunately maximum exploited and polluted in the race of development. The various harmful pollutants (organic as well as inorganic) in soil, air and water impacts both on natural resources and ecosystems. Phytoremediation is a new innovative technology that is cleaning of environment with the help of environment. Phytoremediation involves different plants which has been utilized to clean environment contaminants through different mechanism. Moreover, phytoremediation technology is cost effective as compare to other well-known methods to remediate industrial contaminants. Earlier, one major problem associated with this technology is production of large biomass after remediation process but now that biomass could be utilized in the form of bioenergy. This review focused on different phytoremediation mechanisms and their potential to remediate various pollutant especially heavy metal and textile dyes.

Keywords- Phytoremediation, contaminants, Heavy-metals, Textile Dyes.

I. INTRODUCTION

Phytoremediation is a new innovative and emerging technology that uses plants to degrade, extract, or immobilize contaminants from air, water and soil. As an innovative, cost effective and eco-friendly approach of phytoremediation technology receive more attention as compare to other alternative and established conventional methods used for cleaning pollutants. The conventional methods used for the treatment of contaminants such as leaching of pollutant, solidification/stabilization. electro-kinetical treatment of ionic species migrates to electrodes inserted into the soil, chemical precipitation, reverse osmosis , chemical oxidation and reduction, coagulation-flocculation, ion exchange, and ultrafiltration have multiple disadvantages such as release of toxic substances, formation of a large amount of sludge, cost-issue etc [1-2]. Because of which bioremediation methodologies can be used as alternative technologies for the removal of industrial wastes. Conventional techniques for bioremediation that involve the digging up of contaminated soils and disposal of the wastes to a landfill, lead to contamination elsewhere and can

create significant risks in excavation, handling and hazardous material [3]. transportation of Microbial bioremediation processes for the removal of hazardous compounds, have received quite a lot of focus from researchers all over the world because of the high potentiality of microbes to perform a variety of functions. Microbial biodegradation has been somewhat successful for the degradation of certain organic contaminants and mainly but ineffective to toxic metal [4]. Although these conventional methods are used to clean majority of polluted sites as they are fast and can function over a wide range of oxygen, pH, temperature and osmotic potential but they also tends to be costly and disruptive to environment. The high cost of these technologies is primary driving force behind the search for alternative remediation technologies which have potential to be low-cost and low-impact. This concept is called as phytoremediation.

Though this technology has been used for hundreds of years to treat human wastes, reduce soil erosion and protect water quality, still the research focusing mainly on the phytoremediation of contaminated soils has only grown significantly in the last 25 years. The use of phytoremediation processes for the removal toxicants is comparatively less explored methodology since the fact that plants also possess some inherent metabolic pathways that can breakdown a wide range of toxicants [5] was much less realized a two decades ago. As researchers have now begun to realize the potential of plant systems as effective remediating agents, this new area of phytoremediation has started gaining importance from academician and industry [6]. As a result, phytoremediation is being developed as potential remediation solution for thousands of contaminated sites in the USA [7]. So, phytoremediation becomes a subject of intense public and scientific interest and a topic for current research. Phytoremediation become possible only because of the productive interdisciplinary cooperation of biologists, chemists, agronomists, environmental engineers and federal & state regulators.

Phytoremediation takes advantages of the fact that a living plant can be considered a solar-driven pump, which can extract and concentrate elements or compounds from surrounding environment. Phytoremediation also avoids the need for soil excavation and transport and causes less disruption to ecosystems than physical, chemical or microbial remediation [8]. Phytoremediation can also help in controlling run off, soil erosion, lower air, water emissions and secondary waste production thus making it an attractive technology. This technology also offer a modest way to decrease the effects of global change by lowering energy usage, CO₂ and other harmful gases emission and waste emission during waste water treatment [9]. Therefore, the value of plants to counter balance the hazards of industrialization processes is being appreciated [10]. Phytoremediation can thus serve dual purposes. The release of large amount of toxic wastes into water bodies is one of the consequences of increasing urbanization and industrialization in the modern world. A variety of organic (pesticides, explosives such as TNT, petrochemicals, chlorinated solvents, dyes etc.) and inorganic (heavy metals such as mercury, lead, etc) wastes which have toxic effects on the ecosystem have been contaminating our natural resources [6]. So it is very urgent need to remediate dye bearing effluent by such type of emerging technology. The removal of textile dyes mediated by plant systems is still a much unexplored area of phytoremediation research. Hence, there is wide scope of

removal of pollutant with Phytoremediation method. The present research aims at unraveling the remediation potential and mechanism of different plants to remediate industrial effluent and explore the basic mechanism and the application of these technologies for actual practices which will helps to broaden the horizons of phytoremediation technologies.

II. PHYTOREMEDIATION OF HEAVY METALS

Among various water pollutants, heavy metals are of major concern because of their bio-accumulative nature & carcinogenic and mutagenic effects with a potential toxicity to all life forms [11-12]. Heavy metals are metallic chemical elements with a high atomic weight and density much greater (at least five times) than water. They are highly toxic and cause ill effects at very low concentrations e.g. mercury (Hg), cadmium (Cd), arsenic (As), chromium (Cr), thallium (Tl), and lead (Pb) [13]. A number of research reports has been published on heavy metal remediation by aquatic plants. This area can be further explored.

Among the various terrestrial plants, which involve in phytoremediation process, a very few are listed in table1.

| Plant | Heavy metal | References | Plant | Heavy metal | References |
|---------------------------|-------------|------------|-------------------------------|-------------|------------|
| Lactuca sativa | As, Pb | [15] | Black gram | Cd | [20] |
| Lupinus luteus | Pb, Cu, Cd, | [16] | Mustard and pumpkin | Cd | [21] |
| Orychophragmus violaceus. | Zn | [17] | Maize and tomato | Cd, Pb | [22] |
| Maize | Cr, Pb | [18] | Sedum alfredii | Cd, Zn | [23] |
| Brassica juncea | Pb, Zn, Cu | [19] | Barley cultivar | Pb,Cd | [24] |
| Pteris vitata | As | [25] | Alfa-alfa and Sorghum bicolor | Cr | [26] |
| Helianthus indicus | Cu | [27] | Bryophyllum pinnatum | Cd, Cr,Cu | [28] |
| Euclayptus | Pb,Zn, Cu | [29] | Spartina | Cr,Co, Ni | [30] |
| French marigold | Zn | [31] | Kalanchoe | Cu | [31] |
| Euphorbia | Cd | [32] | Helianthus | Pb | [33] |

| Table 1: Terrestrial | plants known f | for their pot | ential to accum | ulate heavy metals |
|----------------------|----------------|---------------|-----------------|--------------------|
| | 1 | | | 2 |

But it was found that aquatic macrophytes are more suitable for wastewater treatment than terrestrial plants because of their faster growth and larger biomass production, relative higher capability of pollutant uptake, and better purification effects due to direct contact with contaminated water. Their ability to hyperaccumulate heavy metals make them interesting research candidates especially for the treatment of industrial effluents and sewage waste water [13]. Table 2 summarizes on phytoremediation potential of some aquatic macrophytes.

| Aquatic Plant | Heavy metal | Accumulation (dry weight basis) | References |
|-----------------------|----------------------------------|---|----------------------|
| | Zn, Cu, Ni | 9.26 - 112.5mg kg ⁻¹ | [34] |
| | | $2.5-19.0$ mg kg $^{-1}$ | |
| | | $0.5-2.2$ mg kg $^{-1}$ | |
| | Hg | 119 mg Hg g ⁻¹ | [35] |
| | Cd | 3992 µg Cd g ⁻¹ | [36] |
| Eichhornia crassipes | Cu | 314 µg Cu g ⁻¹ | [37] |
| | Cr, Cd, Ni | 2.31 mg Cr g ⁻¹ 1.98 mg Cd g ⁻¹ 1.68 mg Ni g ⁻¹ | [38] |
| | Cr | 1258 μg Cr g ⁻¹ | [39] |
| Elodea densa | Hg | 177 ng Hg g ⁻¹ | [35] |
| Eleocharis acicularis | Fe, Pb, Zn, Mn, Cr, Cu, Ni | 500 μg Fe g ⁻¹ 1120 μg Pb g ⁻¹ 964 μg Zn g ⁻¹ 388 μg Mn g ⁻¹ 265 μg Cr g ⁻¹ 235 μg Cu g ⁻¹ 47 μg Ni g ⁻¹ | [40] |
| Lemna gibba | Ur, As | 897 μg Ur g ⁻¹ 1022 μg As g ⁻¹ | [41] |
| | Zn | 4.23–25.81 mg Zn g ⁻¹ | [42] |
| Lemna minor | Ti | 221 µg Ti g ⁻¹ | [43] |
| | Cu | 400 µg Cu g ⁻¹ | [44] |
| | Pb | 8.62 mg Pb g ⁻¹ | [45] |
| Elodea Canadensis | Ni | >3500 µg Ni g ⁻¹ | [46] |
| Pistia stratiotes | Hg Cr. Cd. Ni | $\begin{array}{c} 0.57 \text{ mg Hg g}^{-1} \\ 215 \text{ ng Hg g}^{-1} \\ 83 \mu \text{g Hg g}^{-1} \\ \hline 2.50 \text{ mg Cr g}^{-1} \\ 2.13 \text{ mg Cd g}^{-1} \end{array}$ | [47] [35] [48] |
| | CI, CU, INI | $1.95 \text{ mg Ni g}^{-1}$ | [20] |

Table 2: Aquatic macrophytes known for their potential to accumulate heavy metals

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| | Cr, Ni, Zn | > 9 mg Cr g ⁻¹ > 10 mg Ni g ⁻¹ > 12 mg Zn g ⁻¹ | [49] |
|-------------------------------|------------|--|------|
| Egeria densa | Cd, Cu, Zn | 70.25 mg Cd g^{-1} 45.43 mg Cu g^{-1} 30.40 mg Zn g^{-1} | [50] |
| Ceratophyllum | As | 525 μg As g ⁻¹ | [36] |
| demersum | Cd, Zn | $\begin{array}{ccc} 143 & \mu C dg^{-1} \\ 57 \ \mu g \ Zn \ g^{-1} \end{array}$ | [51] |
| Potamogeton pusillus | Cu | 162 μg Cu g ⁻¹ | [52] |
| Vallisneria spiralis | Cr, Cd, Ni | $\begin{array}{cccc} 2.85 & mgCrg^{-1} \\ 2.62 & mCdg^{-1} \\ 2.14 mg Ni g^{-1} \end{array}$ | [38] |
| | Hg | 158 μg Hg g ⁻¹ | [53] |
| Cattails | Se | 50 ppm of Se | [54] |
| Hydrilla | Pb | 20ppm of Pb | [55] |
| Azolla | Au | 98 mg Au g ⁻¹ | [56] |
| | Cd, Cr | 1623.201 mg/g Cd and 6013.1 mg/g Cr | [57] |
| Salvinia | Pb | 0.186 mg/g | [58] |
| Myriophyllum verticillatum | Pb | 97.60 mg/g | [59] |
| Bacopa | Hg, Cd | 25.5mg/g Hg, 49.9 mg/g Cd | [60] |
| Typha aungustifoila | Cr, Zn, Cu | 210 μg Cr g ⁻¹ 325 μg Zn g ⁻¹ 7,022 μg Cu g ⁻¹ | [61] |

It clearly pointed out that phytoremediation potential of aquatic plants is dependent upon the tolerance level and toxicity of plant genera or species employed in a particular study. Secondly, within a particular plant genus and/or species, there exists a difference in accumulation potential for the same heavy metal. The existing variation is because the phytoremediation potential is regulated by environmental factors like chemical specific and initial concentrations of the metal, temperature, pH, redox potential, salinity, and interaction of different heavy metals among each other. Still a lot of research work is left in area to increase the hyper accumulative capacity of plants.

III. PHYTOREMEDIATION OF TEXTILE DYES

Dye bearing effluents is another major area of concern. Synthetic dyes and pigments are one of major classes of pollutants contributes to industrial pollution. These dyes have a major contribution in the textile industries for dying purpose. As color is a visible pollutant and the presence of even very minute amount of coloring substance makes it undesirable. Discharge of such effluents imparts color to receiving streams and affects its aesthetic value also. The dyes are generally stable to light, oxidizing agents and heat, and their presence in wastewaters offers considerable resistance to their biodegradation and thus get [2] accumulated into aquatic life forms such as fishes. Through the food chain, dyes ultimately posing risks to human beings. Sometimes the products formed after the processing of these dyes themselves are also toxic. In the environment, **Azo** compounds of these dyes are reduced to liberate benzidine and other aromatic amines, which may cause adverse systemic health effects or cancer of urinary bladder cancer, stomach, kidneys, brain, mouth, esophagus, liver, and gall bladder might also be targets [14]. Therefore, the removal of Azo-Dyes from industrial effluents is one of the major problems due to the difficulty in treating such waste waters by conventional treatment methods. A very few aquatic species like Eichhornia crassipes, Typhonium flaglliforme,Salivinia, Pistia etc.. have been reported in this area are listed in Table 3:

| Fable 3: Aquatic mac | ophytes known | for their potential | to remediate | different dy | yes |
|-----------------------------|---------------|---------------------|--------------|--------------|-----|
|-----------------------------|---------------|---------------------|--------------|--------------|-----|

| Plants | Dye remediated | Reference |
|-----------------------------|--|-----------|
| Rheum rabarbarum | Sulfonated anthraquinones | [62] |
| Phragmites australis | Acid Orange 7 | [63] |
| Blumea malcommi | Direct Red 5B | [64] |
| Glandularia pulchella | Sulphonated azo dye green HE4B | [65] |
| Glandularia pulchella | Remazol Orange 3R | [66] |
| Eichhornia,Pistia, Salvinia | Direct dark blue 6B,Direct black H/Y, Direct congo red | [67] |
| Typha | Reactive | [68] |
| Angustifolia | Red 141 | |
| Tagetes patula | Reactive Blue 160 | [69] |
| Cactus | Red HE7B | [70] |
| Zinnia angustifolia | Remazol Black B | [71] |
| Algae | Azo dye | [72] |
| Arabidopsis | Triphenylmethane dyes | [73] |
| Eichhornia | Red RB and Black B | [74] |
| Hydrocotyle vulgaris | Basic Red 46 | [75] |
| Typhonium flagilliform | Brilliant Blue R | [76] |
| Portulaca grandiflora | Reactive Blue 172 | [77,78] |
| Petunia grandiflora | Brilliant Blue G | [79] |

The removal of dyes by plant systems can thus provide a new dimension to phytoremediation research. A large number of industries including textile, paper and pulp, printing, iron– steel, electroplating, coke, petroleum, pesticide, paint, solvent, and pharmaceutical etc. consume large volumes of water and organic chemicals which differ in their composition and toxicity. The discharge of effluents from these industrial units to various water bodies leading to water pollution is a matter of great concern, especially for developing countries like India. According to survey carried out by Central Pollution Control Board (2008) [81], ground water in 40 districts from 13 states of India i.e. Andhra Pradesh, Assam, Bihar, Haryana, Himachal Pradesh, Karnataka, Madhya Pradesh, Orissa, Punjab, Rajasthan, Tamil Nadu, Uttar Pradesh, West Bengal, and five blocks of Delhi is contaminated with heavy metals. Also, according to a report published in Times of India (Jan-2013) which clearly states that Punjab among the worst performing state in the country when it comes to checking water-pollution where they dump their toxic waste directly in the river and lake. In India, where most of the developmental activities are still dependent upon water bodies, industrial water pollution is posing serious environmental and health problems. Currently,

\$6–8 billion per year is spent for environmental cleanup in the United States, and \$25–50 billion per year worldwide [82] and phytoremediation is on an average tenfold cheaper than engineering-based remediation methods Thus, a definite need exists to develop a low cost and eco-friendly technology to remove pollutants thereby improving water quality.

IV. CONCLUSION

Phytoremediation, the use of plants for environmental cleanup, has gained acceptance in the past 10 years as a costnon-invasive alternative or complementary effective, technology for engineering-based remediation methods. Bioremediation involving the use of microorganisms is a widely studied technology since prokaryotic systems has known to possess diverse metabolic pathways that can metabolize a wide variety of toxicants. But, the fact that plants also comprise of dynamic biological systems that can bestow the plants with excellent remediating abilities was much less realized. Nevertheless, the awareness regarding phytoremediation technologies has been on an increase in the last decade. Industrialization is central to economic development and improved prospects for human wellbeing. But, industrialization has also led to the release of a large number of toxic compounds that have been polluting the ecosystem. Most of the research regarding phytoremediation technologies has focused on the removal of inorganic pollutants like heavy metals and certain types of organic compounds such as polyphenols, chlorinated compounds such as trichloroethylene. But, literature survey has revealed that the remediation of dyes which constitute one group of recalcitrant pollutants is one of the most neglected areas of phytoremediation research. Very few studies have reported the application of these technologies for the removal of textile dves. These studies lack the screening of different plant species for the removal of textile dyes as that reported for the removal of metal contaminants. Hence, very few potent textile dye removing plant species have been reported. Moreover, there is lot of scope for heavy metal removal also. To which Phytoremediation offers an attractive alternative to remediate industrial pollutants. Hence there is wide scope of research in field of phytoremediation.

V. REFERENCES

- [1] Barcelo, J. and Poschenrieder, C., "Phytoremediation : Principles and Prespectives" *Contributions to sciences*, vol. 2(3) pp. 333-344, 2003.
- [2] Lorimer, J. P., Mason, T. J., Plattes, M., Phull, S. S. and Walton, D. J., "Degradation of dye effluent" J. Pure Appl. Chem, vol. 73, 12 pp.1957–1968, 2001.
- [3] Vidali, M., "Bioremediation-An overview" J. Pure Appl. Chem, vol.73 (7) pp.1163-1172, 2001.
- [4] Senan, R.C. and Abraham E.T., "Bioremediation of textile azo dyes by aerobic bacterial consortium" *Biodegradation*, vol.15 pp. 275-280, 2001.

- [5] Chaudhry, Q., Zandstra, M.B., Gupta, S. and Joner, E.J., "Utilizing the synergy between plants and rhizosphere organisms to enhance breakdown of organic pollutants in the environment" Environ Sci Pollut R , vol. 12 pp. 34-48,2005.
- [6] Cluis, C., "Junk-greedy Greens: Phytoremediation as a new option for soil decontamination" Biotechnol. J, vol. 2 pp. 61-67, 2004.
- [7] EPA, "Introduction to phytoremediation" EPA/600/R-99/107. National Risk Management Research Laboratory, Cincinnati 2000.
- [8] Doran, P.M., "Application of plant tissue cultures in phytoremediation research:Incentives and limitations" Biotechnol. Bioeng, vol. 103, pp.60-76, 2009.
- [9] Schroder, P., Daubner D., Maier, H., Neustifter J. and Debus R., "Phytoremediation of organic xenobioticsglutathione dependent detoxification in *Phragmites* plants from European treatment sites" Bioresour Technol, vol. 99, pp.7183-7191,2008.
- [10] Cunningham, S.D. and Ow D.W., "Promises and prospects of phytoremediation" J. Plant Physiol, vol.110, pp. 715-719,1996.
- [11] Yadav S.K., Juwarkar A.A., Kumar G.P., Thawale P.R., Singh S.K. and Chakrabarti T., "Bioaccumulation and phyto-translocation of arsenic, chromium and zinc by *Jatropha curcas* L.: Impact of dairy sludge and biofertilizer", Bioresour. Technol., vol. 100, pp.4616– 4622, 2009.
- [12] Chang J.S., Yoon I.H. and Kim K.W., "Heavy metal and arsenic accumulating fern species as potential ecological indicators in As-contaminated abandoned mines" ECOL INDIC, vol. 9 pp.1275–1279, 2009.
- [13] Rai P.K., "Heavy metal pollution in lentic ecosystem of sub-tropical industrial region and its phytoremediation" Int J Phytoremediation, vol.12 pp. 226–242, 2010.
- [14] Bafana, A., Jain, M., Agrawal, G. and Chakrabarti, T., "Bacterial reduction in genotoxicity of Direct Red 28 dye", *Chemosphere*, vol. 74, pp.1404-1406, 2009.
- [15] Gunduz S., Nezihi Uygur F. and Kahramanolu I., "Heavy metal Phytoremediation potentials of *Lepidum sativum* L., *Lactuca sativa* L., *Spinacia oleracea* L. and *Raphanus sativus* L" Int. j. res. agric. food sci., vol.1(1) pp.1-5,2012.
- [16] Dary M., Chamber-Pérez M.A., Palomares A.J. and Pajuelo E. ,"In situ phytostabilisation of heavy metal polluted soils using *Lupinus luteus* inoculated with metal resistant plant-growth promoting rhizobacteria" J Hazard Mater, vol.177, pp. 323-330,2010.
- [17] He C.Q., Tan G.E., Liang X., Du W., Chen Y.L., Zhi G.Y. and Zhu Y. ,"Effect of Zn-tolerant bacterial strains on growth and Zn accumulation in *Orychophragmus violaceus*" APPL SOIL ECOL, vol. 44, pp.1-5,2010
- [18] Braud A., Jezequel, K., Bazot, S. and Lebeau T., "Enhanced phytoextraction of an agricultural Cr- and

Pb-contaminated soil by bioaugmentation with siderophore-producing bacteria" *Chemosphere*, vol.74 pp.280-286, 2009.

- [19] Wu S.C., Cheung K.C., Luo Y.M. and Wong M.H., "Effects of inoculation of plant growth-promoting rhizobacteria on metal uptake by *Brassica juncea*" *Environ pollut.*, vol. 140, pp.124-135, 2006.
- [20] Ganesan V., "Rhizoremediation of cadmium soil using a cadmium resistant plant growth-promoting rhizopseudomonad" Curr Microbiol., vol. 56, pp.403-407, 2008.
- [21] Sinha S. and Mukherjee S.K., "Cadmium-induced siderophore production by a high Cd-resistant bacterial strain relieved Cd toxicity in plants through root colonization" *Curr Microbiol*, vol. 56, pp.55-60,2008.
- [22] Jiang C.Y., Sheng X.F., Qian M. and Wang Q.Y., (2008) "Isolation and characterization of a heavy metal-resistant *Burkholderia* sp. From heavy metal-contaminated paddy field soil and its potential in promoting plant growth and heavy metal accumulation in metalpolluted soil" *Chemosphere*, 72 pp.157-164.
- [23] Li W.C., Ye Z.H. and Wong M.H., "Effects of bacteria on enhanced metal uptake of the Cd/Zn-hyperaccumulating plant, *Sedum alfredii*" *J Exp Bot*, vol.58, pp. 4173-4182,2007.
- [24] Belimov A.A., Kunakova A.M., Safronova V.I., Stepanok V.V., Iudkin L., Alekseev I.U. V. and Kozhemiakov A.P.
 "Employment of associative bacteria for the inoculation of barley plants cultivated in soil contaminated with lead and cadmium" *Mikrobiologiia*, vol. 73, pp.118-25, 2004.
- [25] Mandal,A., Purakayastha, T.J., Patra, A.K., Sanyal, S.K.,
 "Phytoremediation of arsenic contaminated soil by Pteris vittata" Int J Phytoremediation, vol. 14(6), pp.621-628, 2012.
- [26] Kumar, N., "Comparative Phytoremediation of Chromium-Contaminated Soils by Alfalfa (*Medicago* sativa) and Sorghum bicolor" Int. J. of Sci. Res. Environ. Sci., vol.1(3) pp. 44-49,2013.
- [27] Das, S.,Goswami, S., Das Talukdar, A., "Copper Hyperaccumulating Plants From Barak Valley, South Assam,India For Phytoremediation" *International Journal* of *Toxicological and Pharmacological Research*, vol. 5(1) pp.30-32, 2013.
- [28] Odoemelam, S. A and Ukpe, R. A., "Heavy meal decontamination of polluted soils using Bryophyllum pinnatum" Afr J Biotechnol, vol.7 (23) pp. 4301-4303, 2008.
- [29] Coupe, J. S., Sallami, K. and Ganjia, E., "Phytoremediation of heavy metal contaminated soil using different plant species" Afr J Biotechnol, vol. 12(43), pp. 6185-6192, 2013.
- [30] Cambrollé, J., Mateos-Naranjo, E., Redondo-Gómez, S., Luque, T.and Figueroa, M.E. , "The role of

two *Spartina* species in phytostabilization and bioaccumulation of Co, Cr, and Ni in the Tinto-Odiel estuary (SW Spain) "*Hydrobiologia*, vol. 67(1) pp. 95–103, 2012.

- [31] Lai, H.Y., Juang, K.W and Chen, Z. S., "Large-Area Experiment On Uptake Of Metals By Twelve Plants Growing In Soils contaminated With Multiple Metals" Int J Phytoremediation, vol.12 pp.785–797,2010.
- [32] Husnain, A., Ali, S. S., Zaheeruddin and Zafar,R., "Phytoremediation of Heavy Metals Contamination in Industrial Waste Water by *Euphorbia Prostrata*" *Current Research Journal of Biological Sciences*, vol. 5(1), pp.36-41, 2013.
- [33] Jarosz, Z., Dzida, K., Pitura and K. onopińska, J., "The possibility of using 5-aminolevulinic acid in lead phytoextraction process" Mod. Phytomorphol., vol. 3 ,pp. 57–62, 2013.
- [34] Hammad D.A. ,"Cu, Ni and Zn Phytoremediation and Translocation by Water Hyacinth Plant atDifferent Aquatic Environments', Aust. j. basic appl. sci., vol. 5(11), pp.11-22, 2011.
- [35] Molisani M.M., Rocha R., Machado W., Barreto R.C. and Lacerda L.D., "Mercury contents in aquatic macrophytes from two reservoirs in the paraíba do sul:guandú river system", Braz. J. Biol., vol. 66 pp.101–107,2006.
- [36] Mishra K.K., Rai U.N. and Prakash O., "Bioconcentration and phytotoxicity of Cd in *Eichhornia crassipes*" *Environ Monit Assess*, vol.130 pp.237–243,2007.
- [37] Hu C., Zhang L., Hamilton D., Zhou W., Yang T.and Zhu D., "Physiological responses induced by copper bioaccumulation in *Eichhornia crassipes* (Mart.)" *Hydrobiologia*, vol. 579 pp.211–218,2007.
- [38] Verma V.K., Tewari, S. and Rai, J.P.N., "Ion exchange during heavy metal bio-sorption from aqueous solution by dried biomass of macrophytes" *Bioresource Technol.*, vol.99 pp.1932–1938, 2008.
- [39] Paiva, B.L., de Oliveira, J.G., Azevedo, R.A., Ribeiro, D.R., da silva, M.G. and Vitória A.P., "Ecophysiological responses of water hyacinth exposed to Cr³⁺ and Cr⁶⁺" Environ Exper Bot, vol. 65 pp.403–409, 2009.
- [40] Hoang Ha N.T.H., Sakakibara M., Sano S., Hori R.S. and Sera K., "The potential of *Eleocharis acicularis* for phytoremediation: Case study at an abandoned mine site" *Clean Soil, Air, Water*, vol.37 pp.203–208,2009.
- [41] Mkandawire M., Taubert B. and Dudel E.G., "Capacity of *Lemna gibba* L. (Duckweed) for uranium and arsenic phytoremediation in mine tailing waters" Int J Phytoremediation, vol. 6, pp.347–362,2004.
- [42] Khellaf N. and Zerdaoui M., "Phytoaccumulation of zinc by the aquatic plant *Lemna gibba*" *Bioresource Technol.*, vol. 100, pp. 6137–6140, 2009.
- [43] Babic M., Radic S., Cvjetko P., Roje V., Pevalek-KozlinaB. and Pavlica M., "Antioxidative response of *Lemna*

minor plants exposed to thallium(I)-acetate" *Aquat.Bot.*, vol. 91, pp.166–172,2009.

- [44] Boule K.M., Vicente J.A.F., Nabais C., Prasad M.N.V. and Freitas H., "Ecophysiological tolerance of duckweeds exposed to copper" Aquat Toxicol., vol.91, pp.1–9, 2009.
- [45] Uysal Y. and Taner F., "Effect of pH, temperature, and lead concentration on the bioremoval of lead from water using *Lemna minor*" Int J Phytoremediation, vol.11, pp.591–608,2009.
- [46] Maleva, M.G., Nekrasova, G.F., Malec, P., Prasad, M.N.V. and Strzałka K., "Ecophysiological tolerance of *Elodea canadensis* to nickel exposure" *Chemosphere*, vol. 77, pp.392–398,2009.
- [47] Mishra, V.K., Tripathi, B.D. and Kim, K.H., "Removal and accumulation of mercury by aquatic macrophytes from an open cast coal mine effluent" J Hazard Mater, vol. 172, pp.749–754,2009.
- [48] Skinner, K., Wright, N. and Goff, E.P., "Mercury uptake and accumulation by four species of aquatic plants" *Environ. Poll.*, vol. 145, pp.234–237, 2007.
- [49] Mufarrege ,M.M., Hadad, H.A. and Maine, M.A., "Response of *Pistia stratiotes* to heavy metals (Cr, Ni, and Zn) and phosphorous" Arch Environ Contam Toxicol., vol. 58, pp.53–61,2010.
- [50] Pietrobelli, J.M.T., de Módenes, A.N., Fagundes-Kle,n M.R.F. and Espinoza-Quiñones, F.R., "Cadmium, copper and zinc biosorption study by non-living *Egeria densa* biomass" *Water Air Soil Poll.*, vol. 202, pp.385–392,2009.
- [51] Aravind, P., Prasad, M.N.V., Malec, P., Waloszek, A. and Strzałka, K., "Zinc protects *Ceratophyllum demersum* L. (free-floating hydrophyte) against reactive oxygen species induced by cadmium" *Journal of Trace Elements in Medicine and Biology*, vol. 23, pp.50–60,2009.
- [52] Monferran, M.V., Sanchez Agudo, J.A., Pignata, M.L. and Wunderlin, D.A., "Copper-induced response of physiological parameters and antioxidant enzymes in the aquatic macrophyte *Potamogeton pusillus*" *Environmental Pollution*, vol. 157 pp. 2570–2576,2009.
- [53] Rai, P.K. and Tripathi, B.D., "Comparative assessment of *Azolla pinnata* and *Vallisneria spiralis* in Hg removal from G.B. Pant Sagar of Singrauli Industrial region, India" *Environmental Monitoring and Assessment*, vol. 148, pp.75–84,2009.
- [54] Mojiri, A., Hamidi, A., Aziz., Zahed, M. A., and Aziz, S. Q., "Phytoremediation of Heavy Metals from Urban Waste Leachate by Southern Cattail (*Typha domingensis*) " *International Journal of Scientific Research in Environmental Sciences*', vol. 1(4), pp. 63-70, 2013.
- [55] Singh, D., Tiwari, A. and Gupta, R., "Phytoremediation of lead from wastewater using aquatic plants" *Journal of Agricultural Technology*, vol. 8(1), pp. 1-11,2012.

- [56] Valderrama, A., Tapia, J., Peñailillo ,P. and Carvajal , D.E., "Water phytoremediation of cadmium and copper using *Azolla*" *Water Environ J*, vol. 1, pp.1-8,2012.
- [57] Kamel,K. A., "Phytoremediation Potentiality of Aquatic Macrophytes in Heavy Metal Contaminated Water of El-Temsah Lake, Ismailia, Egypt" Middle East J Sci Res, vol. 14 (12), pp.1555-1568, 2013.
- [58] Sukumaran, D., "Phytoremediation of Heavy Metals from Industrial Effluent Using Constructed Wetland Technology" Appl. Ecol. Evs. Res., vol. 1, (5),pp. 92-97,2103.
- [59] Asiye, U., Uyanik, A. and Kutbay, H. G., "Removal of Heavy Metals Using *Myriophyllum verticillatum* (Whorl-Leaf Watermilfoil) in a Hydroponic System" Ekoloji , vol. 22(87), pp. 1-9,2013.
- [60] Hussain, K., Abdussalam, A.K., Ratheesh Chandra, P. and Salim, N., "Heavy metal accumulation potential and medicinal property of *Bacopa monnieri*- a paradox" J. stress physiol. biochem., vol. 7(4) pp. 39-50,2011.
- [61] Firdaus-e-Bareen and Khilji. S., "Bioaccumulation of metals from tannery sludge by *Typha angustifolia*" Afr J Biotechnol, vol. 18, pp.3314–3320,2009.
- [62] Rahul, V., Khandare, N.R., Tatoba, R., Waghmode, R. and Govindwar, S.P., "Bacterial assisted phytoremediation for enhanced degradation of highly sulfonated diazo reactive dye" Environ Sci Pollut Res Int., vol. 19, pp. 1709–1718, 2012.
- [63] Davies, L.C., Carias, C.C., Novais, J.M. and Martins-Dias, S., "Phytoremediation of textile effluents containing azo dye by using Phragmites australis in a vertical flow constructed intermittent feeding constructed wetland" Ecol. Eng., vol. 25, pp.594–605, 2005.
- [64] Kagalkar, A.N., Jagtap, U.B., Jadhav, J.P., Bapat, V.A. and Govindwar, S.P., "Biotechnological strategies for phytoremediation of the sulphonated azo dye Direct Red 5B using Blumea malcolmii Hook." Bioresour Technol', vol. 100, pp. 4104–4110,2009.
- [65] Kabra, A.N., Khandare, R.V., Kurade, M.B., and Govindwar, S.P., "Phytoremediation of a sulphonated azo dye green HE4B by Glandularia pulchella (Sweet) Tronc (Moss Verbena) "Environ Sci Pollut Res Int., vol. 18, pp.1360–1373,2011.
- [66] Kabra, A.N., Khandare, R.V., Waghmode, T.R. and Govindwar, S.P., "Differential fate of metabolism of a sulfonated azo dye Remazol Orange 3R by plants Aster amellus Linn., Glandularia pulchella (Sweet) Tronc. and their consortium" J Hazard Mater , vol. 190, pp.424– 431,2011.
- [67] Anjana, S. and Thanga and S. V. G., "Phytoremediation Of Synthetic Textile Dyes", Asian J Microbiol Biotechnol Environ Sci, vol. 13(1), pp.30-39,2011.

- [68] Nilratnisakorn, S., Thiravetyan, P. and Nakbanpote, W., "Synthetic reactive dye wastewater treatment by narrow-leaved cattails (Typha angustifolia Linn.): effects of dye, salinity and metals" Sci Total Environ., vol. 384. pp.67–76, 2007.
- [69] Patil ,A.V.and Jadhav, J.P.(2013) "Evaluation of phytoremediation potential of Tagetes patula L. for the degradation of textile dye Reactive Blue 160 and assessment of the toxicity of degraded metabolites by cytogenotoxicity" *Chemosphere*, vol. 92(2), pp.225-32,2013.
- [70] Adki, V.S., Jadhav, J.P. and Bapat, V.A., "Exploring the phytoremediation potential of cactus (Nopalea cochenillifera Salm. Dyck.) cell cultures for textile dye degradation" *Int. J. Phytoremediation*, vol. 14(6), pp.554-69, 2012.
- [71] Khandare, V. R., Rane, N. R., Waghmode, T. R., and Govindwar, S. P, "Bacterial assisted phytoremediation for enhanced degradation of highly sulfonated diazo reactive dye" Environ Sci Pollut Res Int., vol. 19(5), pp. 1709-1718,2012.
- [72] Wang, X. S., Chen, J. P., "Removal of the Azo Dye Congo Red from Aqueous Solutions by the Marine Alga *Porphyra yezoensis Ueda*" *CLEAN – Soil, Air, Water*, vol. 37(10), pp.793–798, 2009.
- [73] Zhao, F.J., Hamon, R.E., Lombi, E., McLaughlin, M.J. and McGrath, S.P., "Characteristics of cadmium uptake in two contrasting ecotypes of the hyperaccumulators *Thalspi caerulenscens*" *J. Exp. Bot.*, vol. 53, pp. 535–543,2002.
- [74] Muthunarayanan, V., Santhiya, M., Swabna, V. and Geetha. A.," Phytodegradation of textile dyes by Water Hyacinth (*Eichhornia Crassipes*) from aqueous dye solutions" *Int. J. Environ. Sci.*, vol. 1(7), pp.1702-1717, 2011.
- [75] Vafaei, F., Movafeghi, A., Khataee, A. R., Zarei, M., and Salehi Lisar, S. Y., "Potential of *Hydrocotyle vulgaris* for phytoremediation of a textile dye: Inducing antioxidant response in roots and leaves" Ecotoxicol Environ Saf., vol. 93, pp.128-134, 2013.
- [76] Kagalkar, A.N., Jagtap, U.B., Jadhav, J.P., Govindwar, S.P. and Bapat, V.A., "Studies on phytoremediation potentiality of Typhonium flagelliforme for the degradation of BBR" Planta, vol. 232, pp. 271–285, 2010.
- [77] Khandare, R.V., Kabra, A.N., Kurade, M.B. and Govindwar, S.P., "Phytoremediation potential of Portulaca grandiflora Hook. (Moss-Rose) in degrading a sulfonated diazo reactive dye Navy Blue HE2R (Reactive Blue 172)", Bioresour Technol., vol. 102, pp.6774–6777,2011.
- [78] Khandare, R. V., Kabra, R. V., Awate, A. N. and Govindwar, S. P., "Synergistic degradation of diazo dye Direct Red 5B by Portulaca grandiflora and Pseudomonas putida', Int. J. Environ. Sci. Technol., vol. 10, pp.1039– 1050,2013.

- [79] Watharkar, D. A., Khandare, R. V.,Kamble, A.A., Mulla Y. M.,Govindwar, S. P. and Jadhav, J.P., "Phytoremediation potential of Petunia grandiflora Juss.,an ornamental plant to degrade a disperse, disulfonated triphenylmethane textile dye Brilliant Blue G", Environ Sci Pollut Res Int., vol. 20, pp.939–949, 2013.
- [80] Aubert, S.J., Schwitzguebel, J.P., "Screening of plant species for the phytotreatment of wastewater containing sulphonated anthraquinones" *Water Res.*, vol. 38 pp.3569– 3575, 2003.
- [81] Central Pollution Control Board [CPCB] "Status of Water Supply, Wastewater Generation and Treatment in Class-I Cities and Class-II Towns of India"Control of Urban Pollution Series, CUPS/70/2009-10, New Delhi,2008.
- [82] Tsao, D.T. "Overview of phytotechnologies" Adv Biochem Eng Biotechnol.,vol. 78 pp.1-50, 2003.



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