

Copper-containing mineral sulfides as thermoelectric materials

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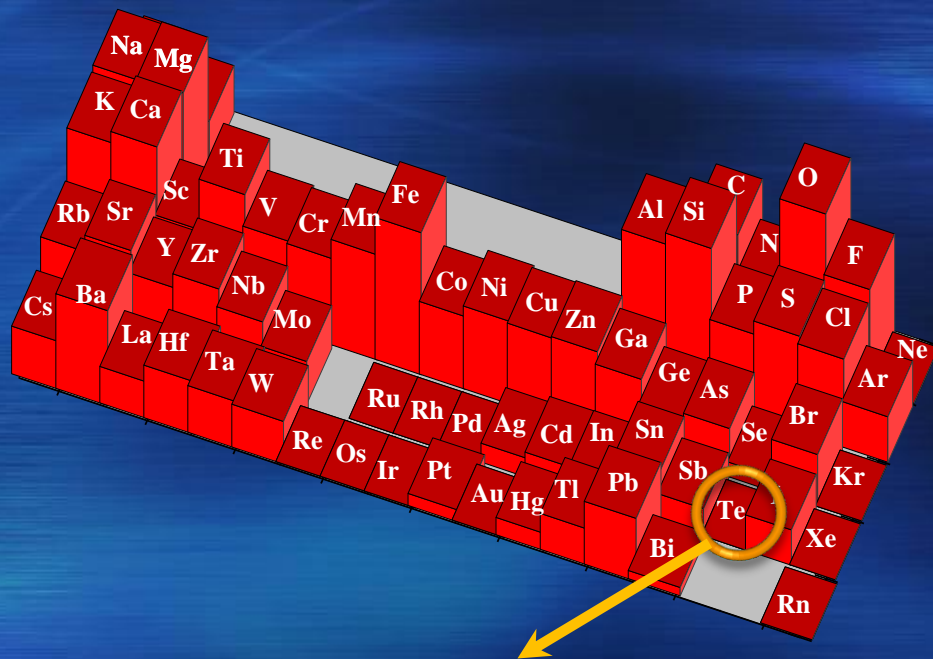
Group web pages:

<http://www.personal.reading.ac.uk/~qf906281/>

Need for new materials

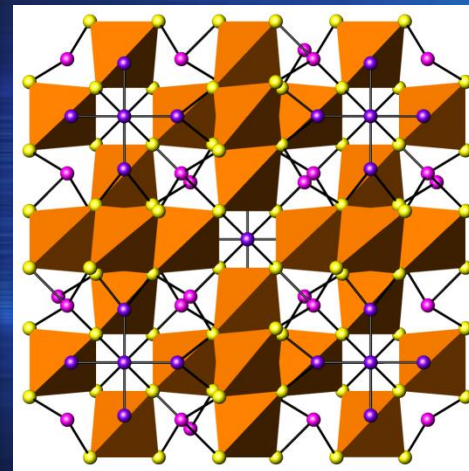
Commercial thermoelectric material:
 Bi_2Te_3 (doped) $ZT \sim 1$ at 25°C

Abundance of elements

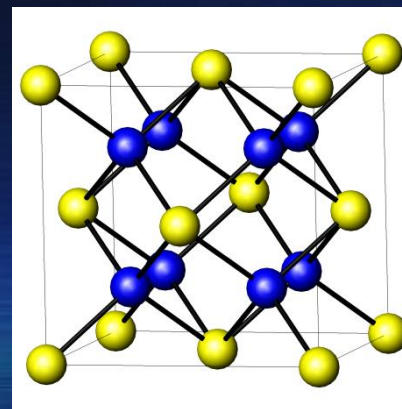


1 ppb in the Earth's crust

Materials made at Reading:



Tetrahedrite
(doped)
 $\text{Cu}_{12}\text{Sb}_4\text{S}_{13}$
 $ZT \sim 1$ at 450°C
Chem. Mater. 2017,
29, 4080.



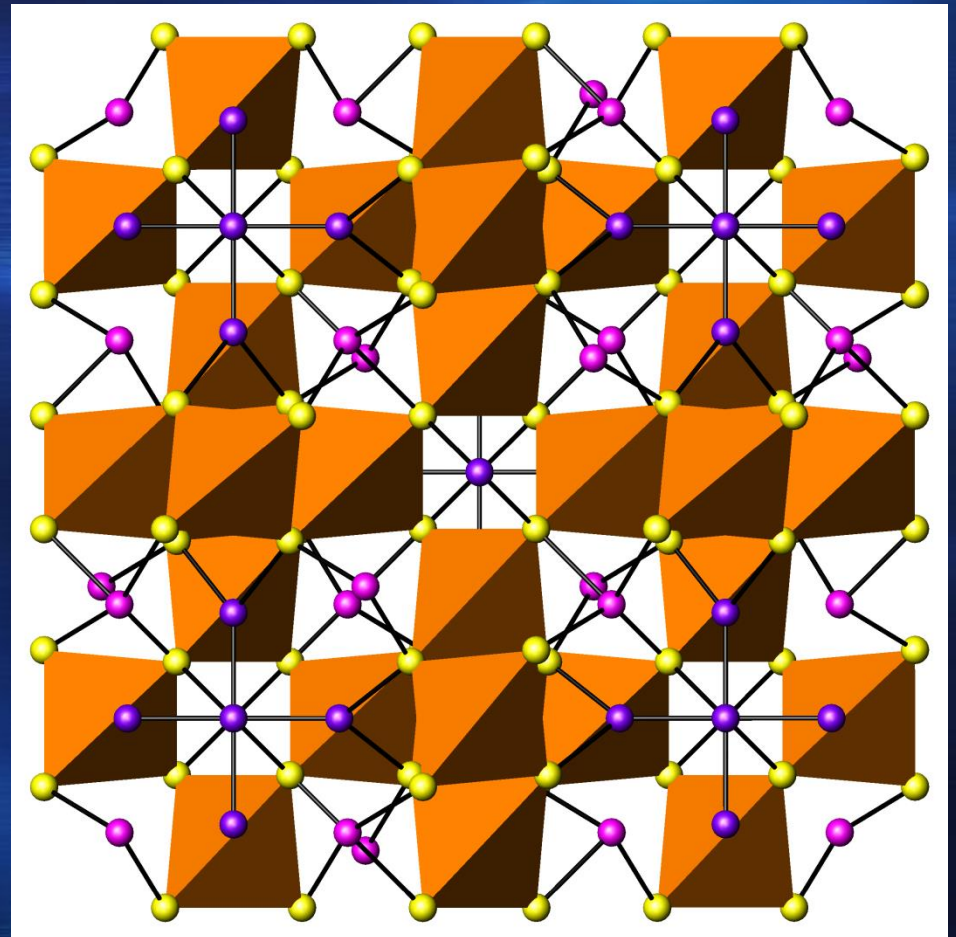
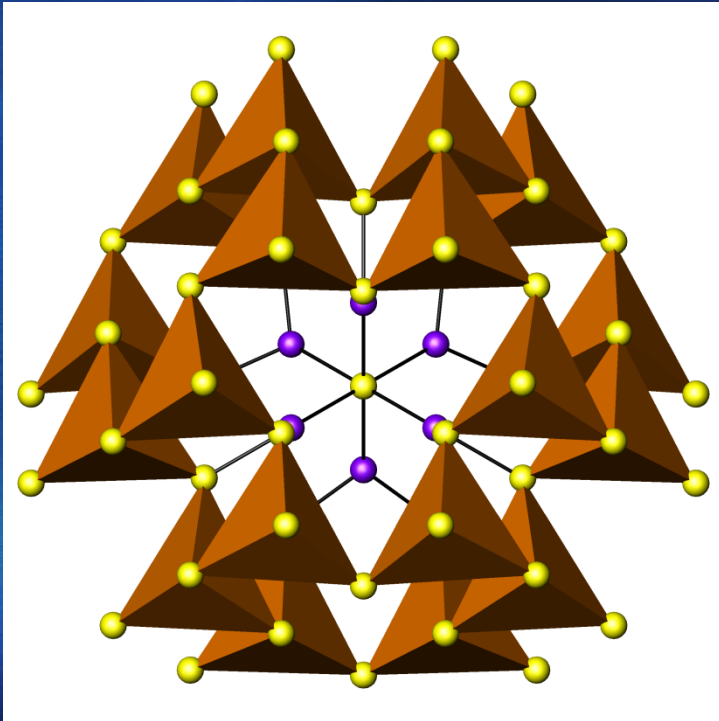
Bornite (doped)
 Cu_5FeS_4
 $ZT \sim 0.8$ at 280°C
Chem. Mater. 2018,
30, 456.

Tetrahedrite

Tetrahedrite is a mineral with the formula $\text{Cu}_{12}\text{Sb}_4\text{S}_{13}$; a common copper ore.

Space group: $\overline{1}43m$

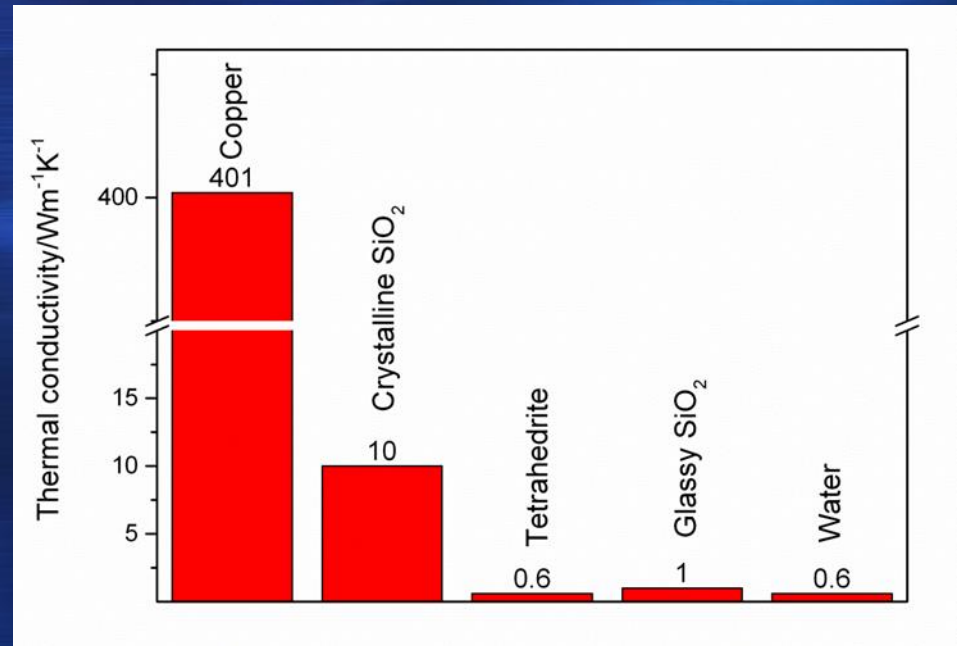
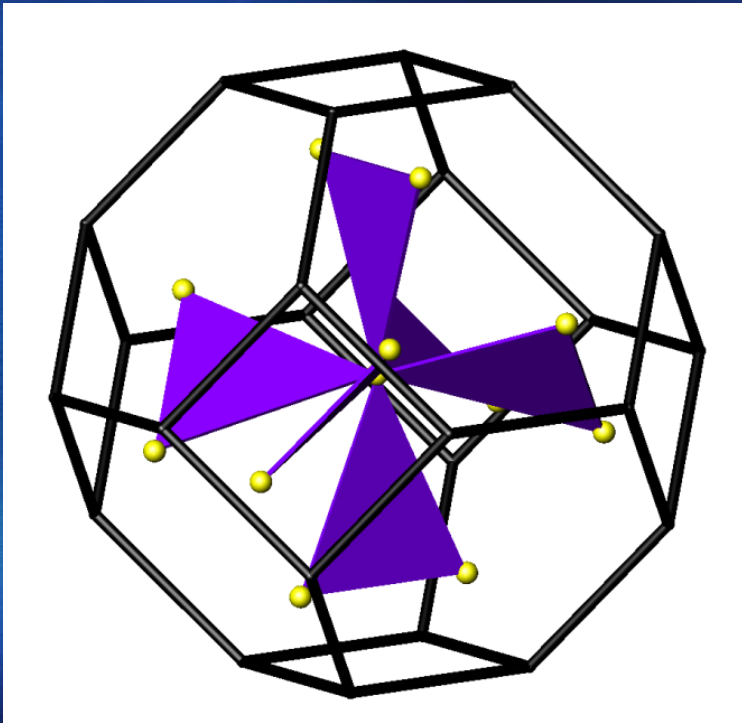
- **Sb** trigonal pyramidal site
- **Cu(1)** tetrahedral site
- **Cu(2)** trigonal planar site



Tetrahedrite: a collapsed sodalite

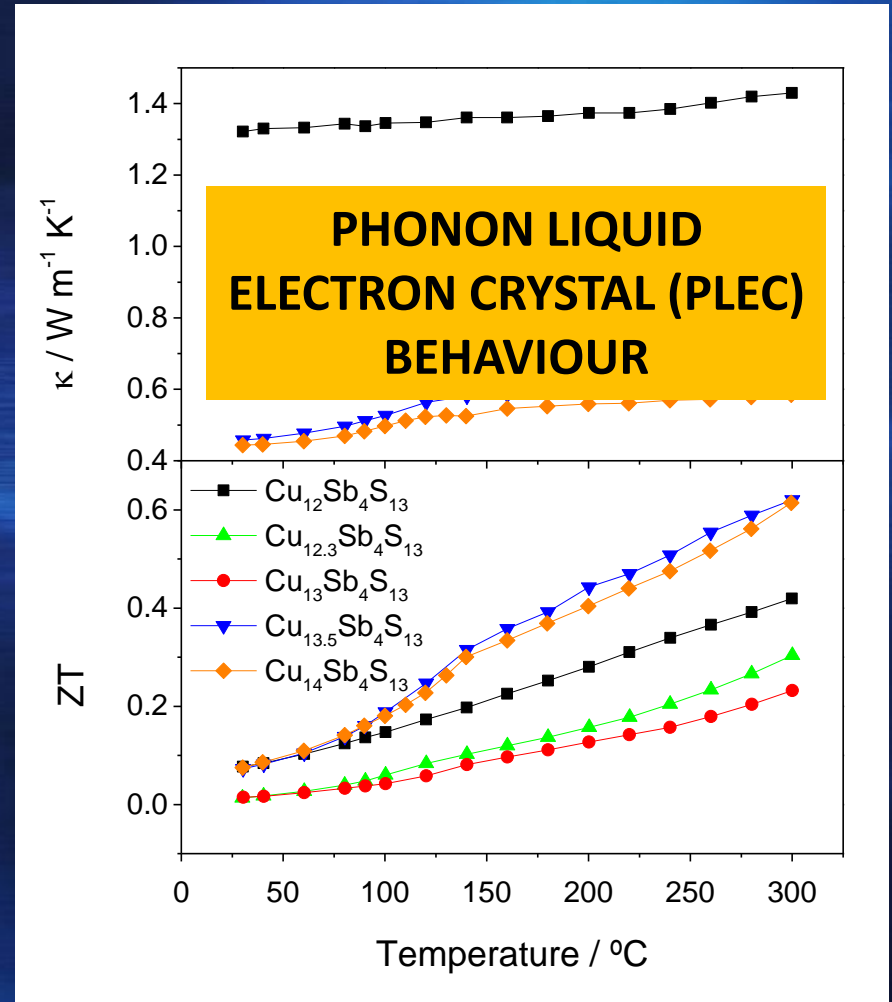
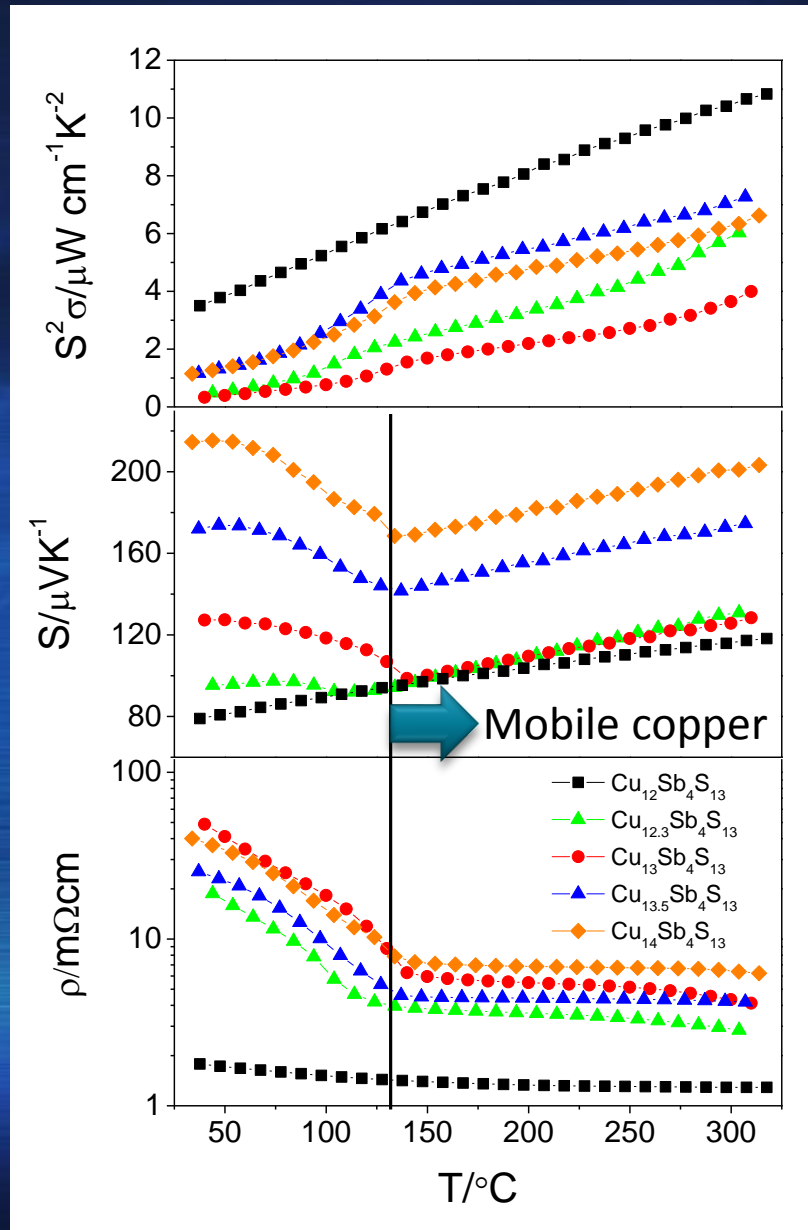
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- **Sb** trigonal pyramidal site
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$ZT \sim 1$ at 450°C

Copper-rich tetrahedrites: $\text{Cu}_{12+x}\text{Sb}_4\text{S}_{13}$

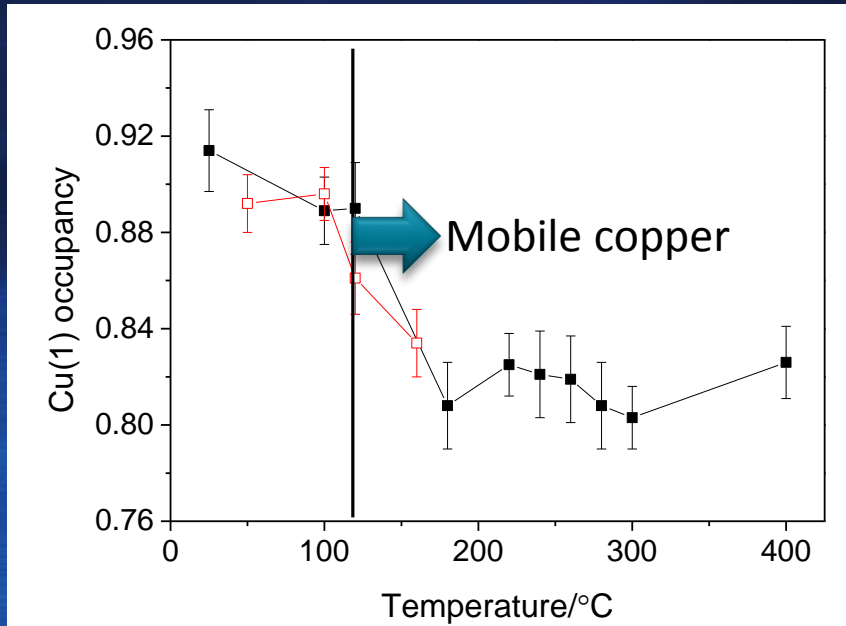


$ZT \sim 0.6$ at 300°C

Estimated $ZT \sim 1$ at 450°C

Copper mobility

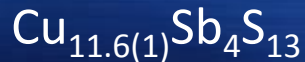
Cu(1)S₄ in “Cu₁₄Sb₄S₁₃”



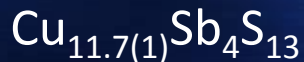
Anomaly in lattice parameters at the onset of copper mobility.

At 400°C:

Refined composition for “Cu_{12.3}Sb₄S₁₃”:



Refined composition for “Cu₁₄Sb₄S₁₃”:



LETTERS

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nature
materials

Copper ion liquid-like thermoelectrics

Huili Liu^{1,2}, Xun Shi^{1,3*}, Fangfang Xu³, Linlin Zhang³, Wenqing Zhang³, Lidong Chen^{1*}, Qiang Li⁴, Ctirad Uher⁵, Tristan Day⁶ and G. Jeffrey Snyder⁶

Advanced thermoelectric technology offers a potential for converting waste industrial heat into useful electricity, and an emission-free method for solid state cooling^{1,2}. Worldwide efforts to find materials with thermoelectric figure of merit, *zT* values significantly above unity, are frequently focused on crystalline semiconductors with low thermal conductivity³. Here we report on Cu_{2–3}Se, which reaches a *zT* of 1.5 at 1,000 K, among the highest values for any bulk materials. Whereas

convergence⁹, nanostructures^{10,11}, and strong electron–phonon coupling by charge density waves¹².

Crystalline semiconductors usually possess high heat conductivity because the phonon mean free path is long in a periodic structure. Disrupting the periodicity or adding defects reduces the phonon mean free path (scattering phonons) to lower κ_L , but such a reduction is limited to the κ of a glass. Whereas a solid glass propagates some heat through transverse shear vibrations a

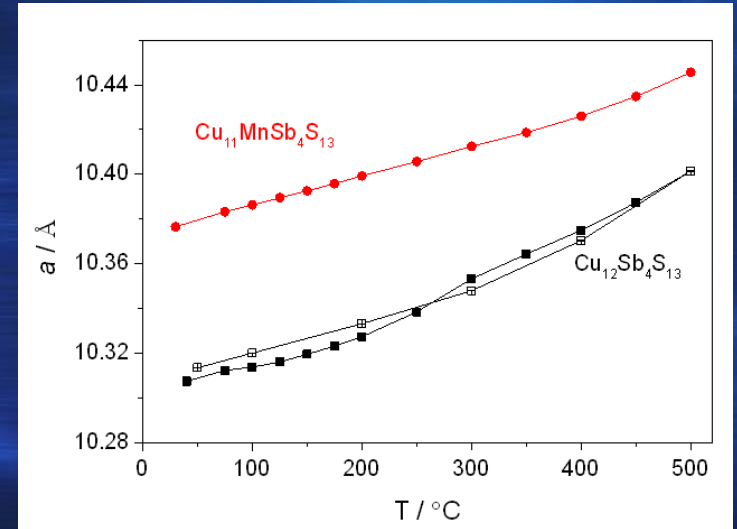
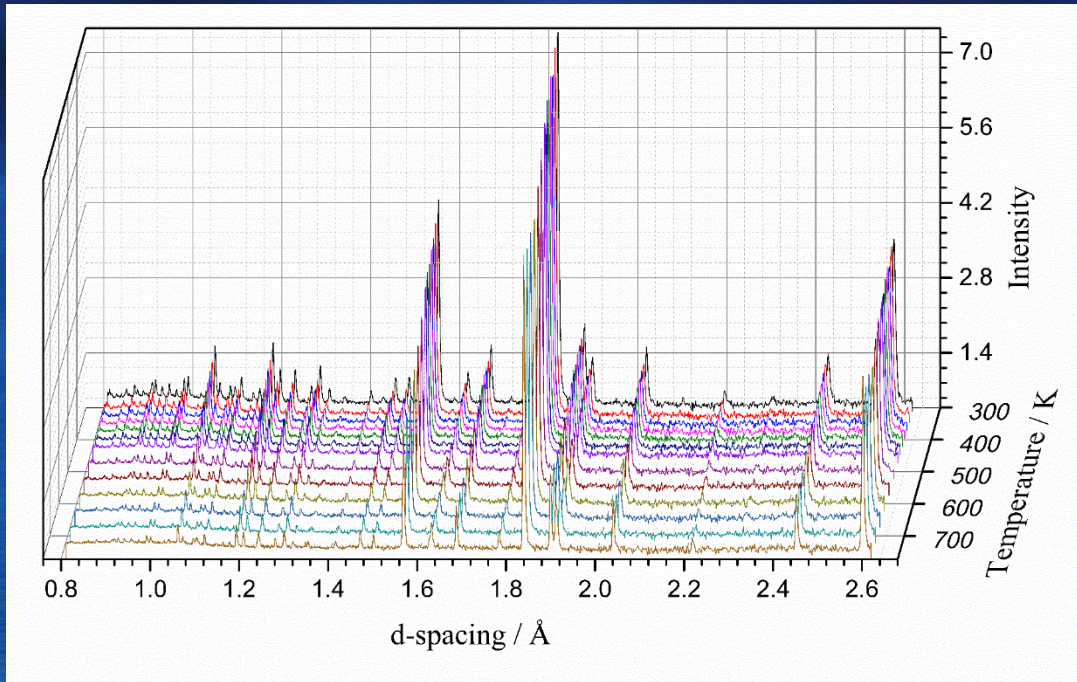


Cu₂Se after DC current for 24 hours...

Adv. Energy Mater. 2014, 1301581

Are all tetrahedrites PLEEC materials?

Neutron diffraction data for $\text{Cu}_{12}\text{Sb}_4\text{S}_{13}$
POLARIS diffractometer (ISIS)



NO EVIDENCE FOR COPPER MOBILITY

Thermal expansion

$$\text{Cu}_{12}\text{Sb}_4\text{S}_{13} \quad 2.31(6) \times 10^{-4} \text{ \AA K}^{-1}$$

$$\text{Cu}_{11}\text{MnSb}_4\text{S}_{13} \quad 1.39(4) \times 10^{-4} \text{ \AA K}^{-1}$$



Sample in sealed ampoule

Manuscript in preparation

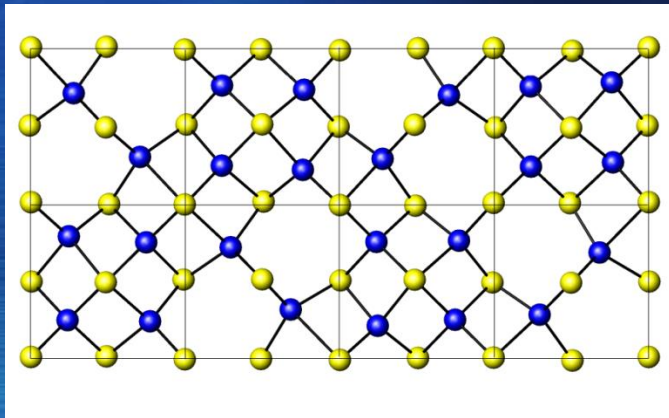
Bornite: $\text{Cu}_5\text{Fe}\square_2\text{S}_4$

- Mineral $\text{Cu}_5\text{Fe}\square_2\text{S}_4$, where □ represents a vacancy
- Structure related to antifluorite / zinc blende

Antifluorite $\text{M}_2\text{S} \equiv \text{M}_8\text{S}_4$

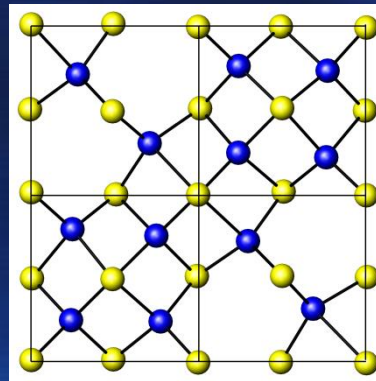
Bornite M_6S_4

Zinc blende $\text{MS} \equiv \text{M}_8\text{S}_8$



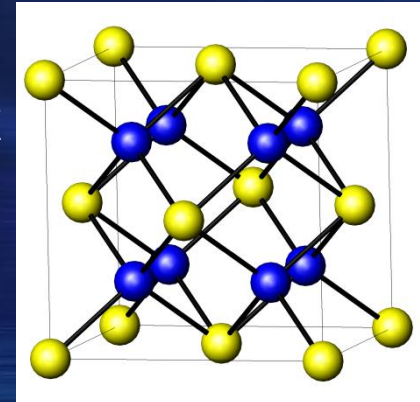
Room-temperature phase
 $4a \times 2a \times 2a$

$T > 473 \text{ K}$



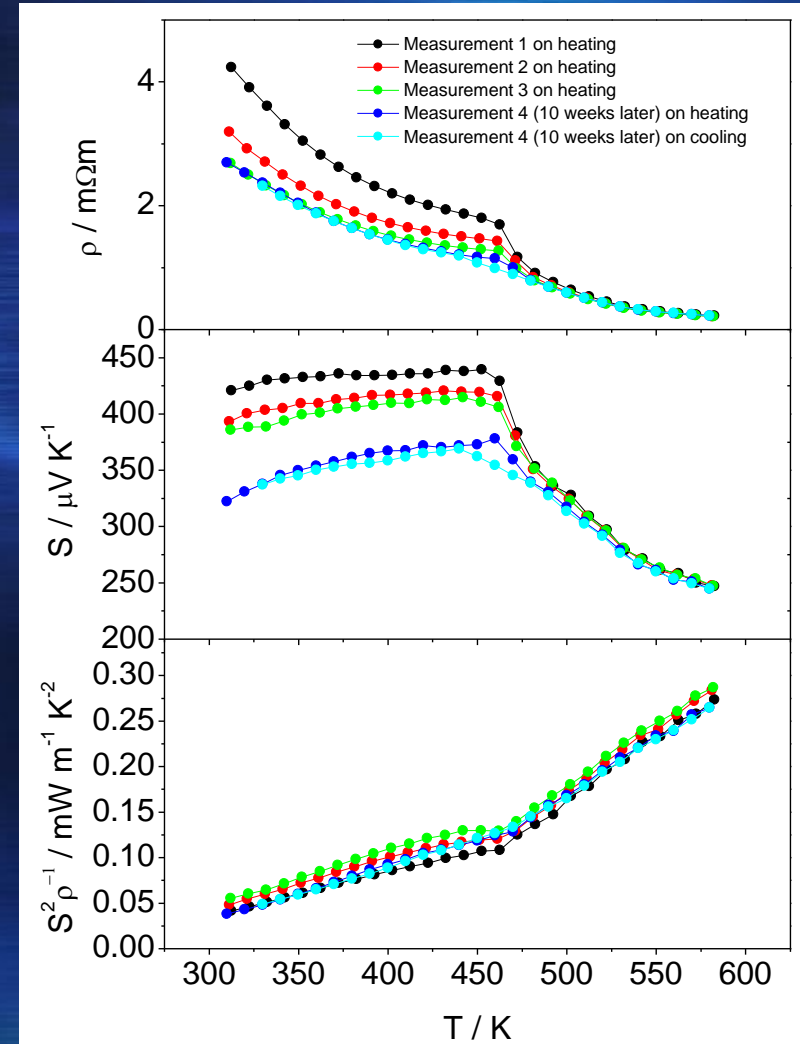
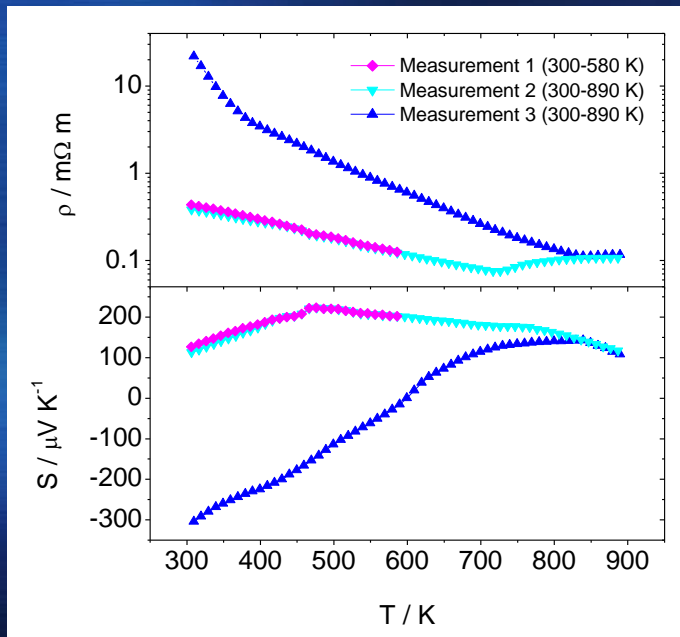
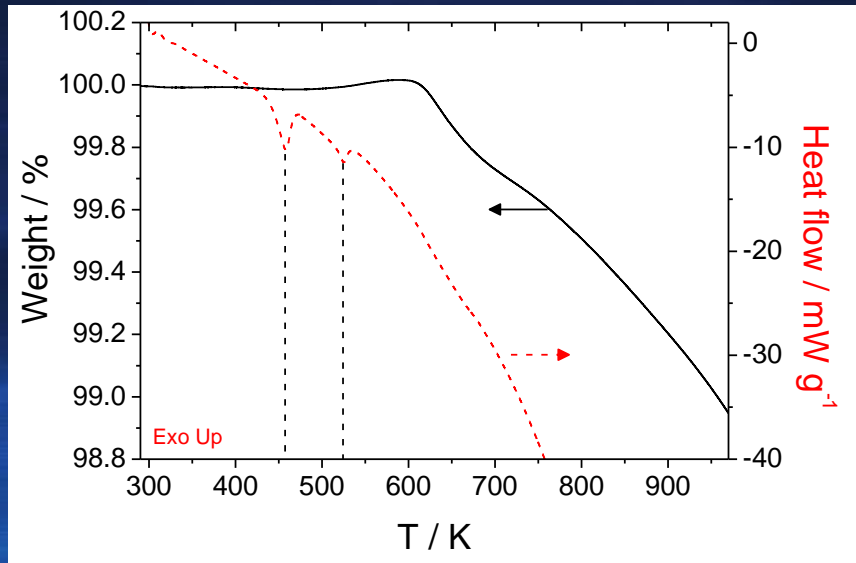
Intermediate phase
 $2a \times 2a \times 2a$

$T > 543 \text{ K}$

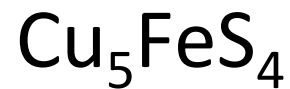
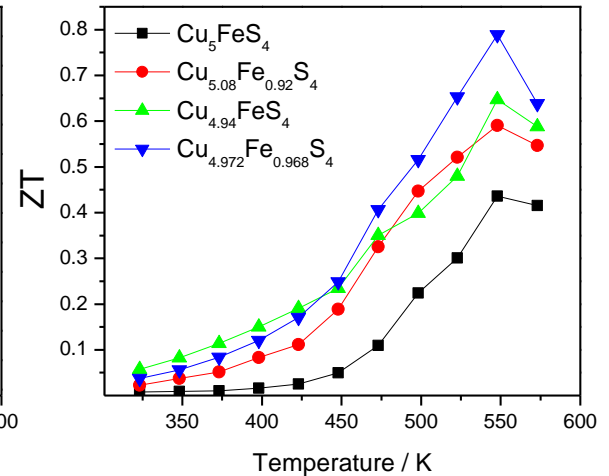
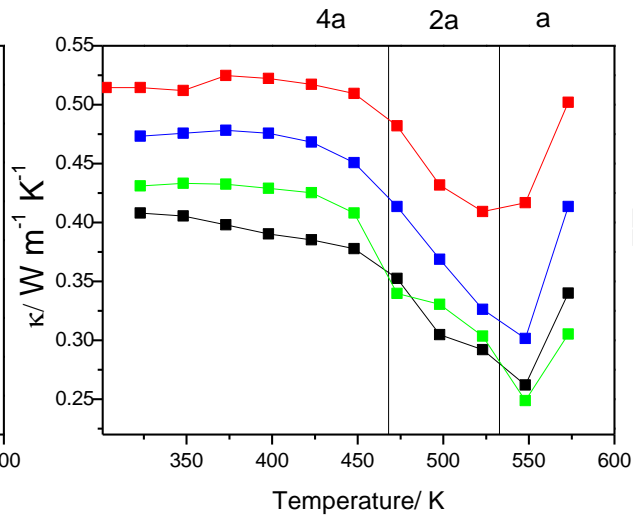
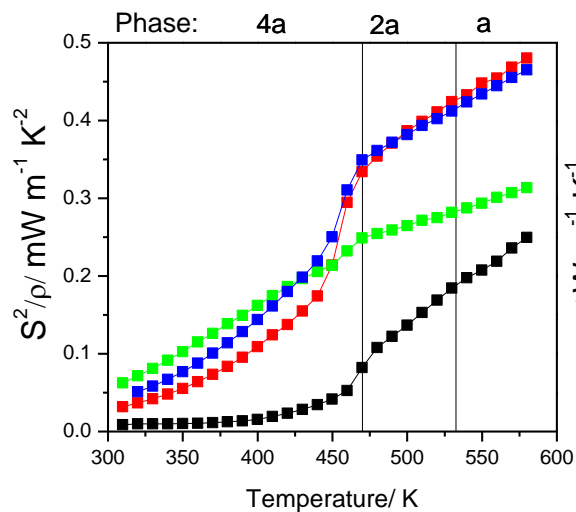


High-temperature phase
 $a \times a \times a$

Thermoelectric properties of bornite



Optimising ZT of bornite



Cu+Fe=6, Parent Phase

Low S^2/ρ , Low κ



Cu+Fe<6, Iron Rich

Improved S^2/ρ , Low κ



Cu+Fe=6, Copper Rich

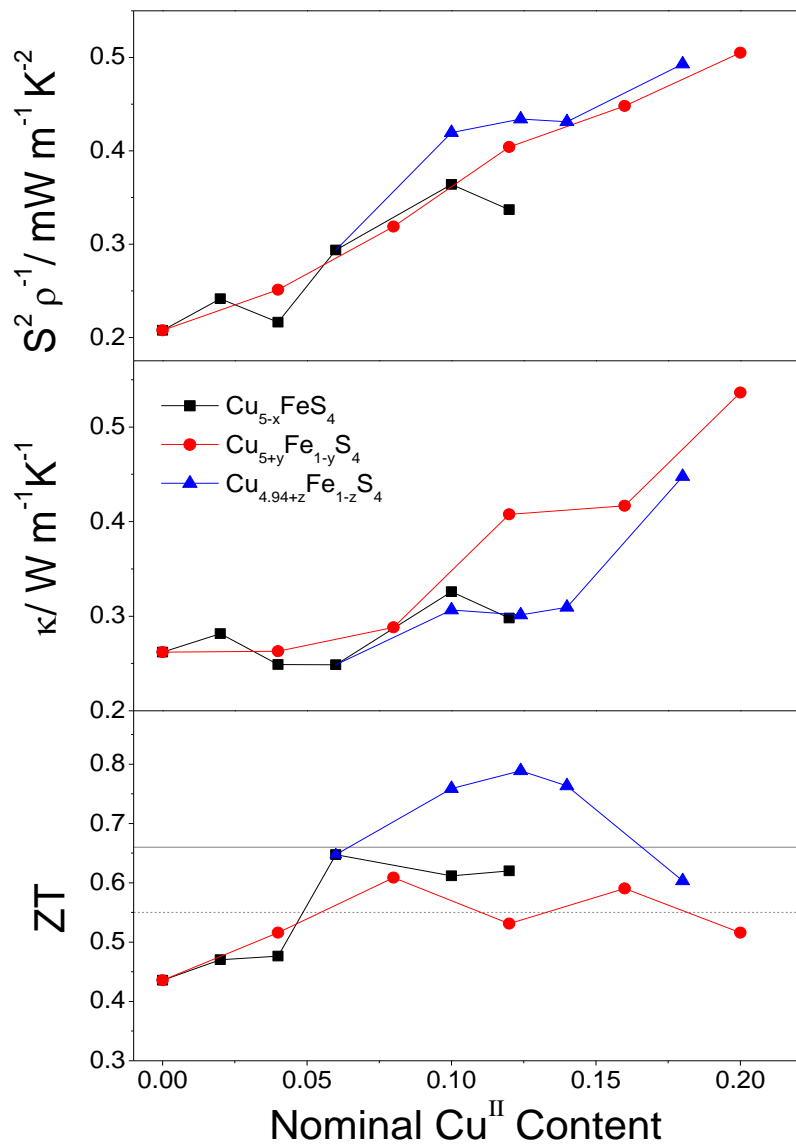
Great S^2/ρ , Increased κ



Cu+Fe=5.94, Copper Rich

Great S^2/ρ , Low κ

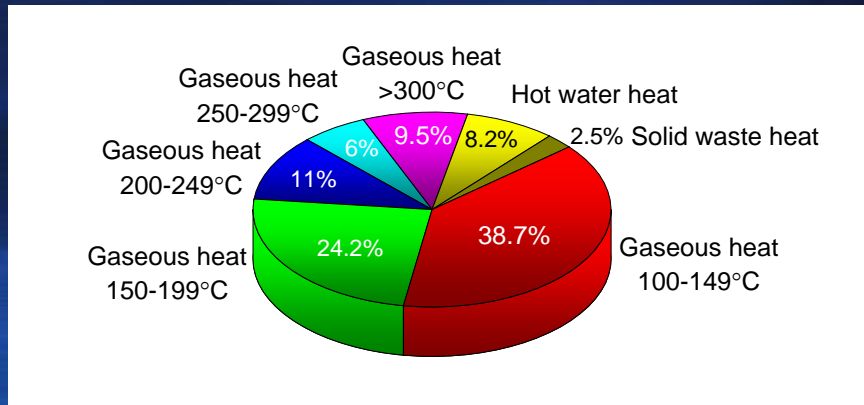
Optimising ZT of bornite



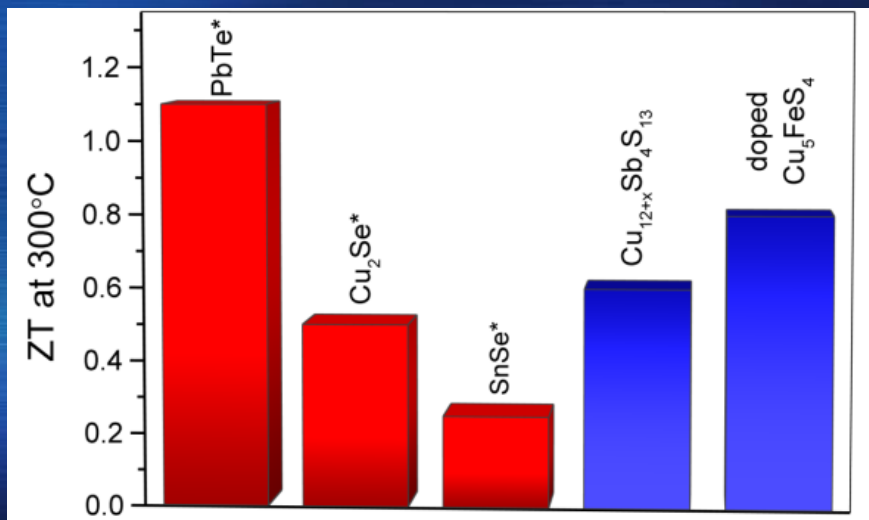
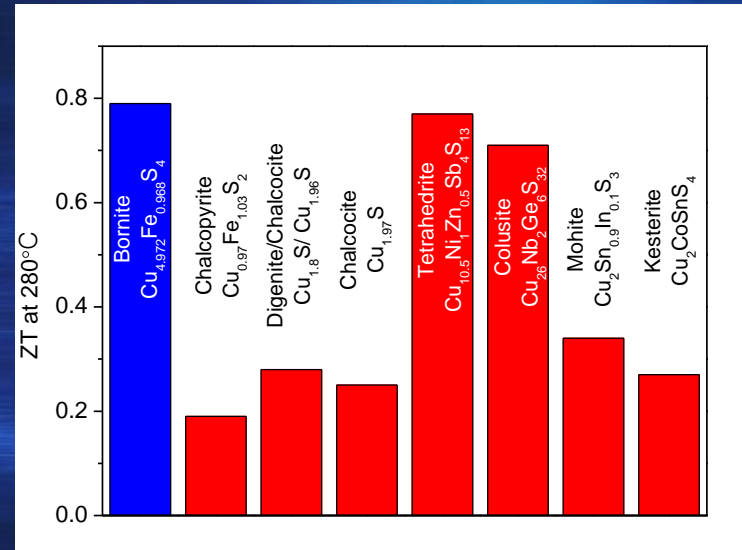
- Cu_5FeS_4 contains Cu(I) and Fe(III)
- Nominal Cu(II) content gives hole concentration.
- Substitution of Fe by Cu has greater impact on the nominal Cu(II) content than Cu deficiency.
- A critical level of vacancies (ca. 1%) is required in order to achieve a high ZT.

Comparison with existing materials

Waste heat recovery



Copper-containing sulfide minerals



Bornite, together with tetrahedrites and colusites, are promising p-type materials for applications at temperatures below 300°C.

* *Nature*, 2012, **489**, 414 ; 2014, **508**, 373; *Nature Materials*, 2012, **11**, 422.

Conclusions so far...

- In copper-rich tetrahedrites, low thermal conductivity is related to copper ionic mobility. These materials are likely to be phonon liquid electron crystals (PLEC).
- There is no evidence of PLEC behaviour in stoichiometric tetrahedrites ($\text{Cu}_{12-x}\text{TM}_x\text{Sb}_4\text{S}_{13}$). Weak copper bonding results in low-energy vibrational modes.
- Bornite is a promising p-type thermoelectric material.
- In bornite, deviations from the ideal Cu_5FeS_4 stoichiometry result in significant enhancements in ZT.

Acknowledgements

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