Nomenclature, Structure, and Properties of Aldehydes & Ketones

- 1. Write a structural formula for each of the following compounds:
 - a. *m*-chlorobenzaldhyde
 - b. 2,2-dibromohexanal
 - c. 3-ethylcycloheptanone
 - d. 1-phenyl-2-butanone
- 2. Name each of the following compounds

$$\begin{matrix} 0 \\ \textbf{II} \\ \textbf{a.} \ \textbf{CH}_{3}\textbf{CH}_{2}\textbf{CCH}_{2}\textbf{CH}_{3} \end{matrix}$$

b.
$$(C_6H_5)_2C=0$$

$$H_3C$$
 CH_3
 CH_3
 CH_3

- 3. Give an example of each of the following:
 - a. cyanohydrin
 - b. enolate

- c. hemiacetal
- d. imine
- 4. Arrange the following in order of decreasing boiling point. Explain why you have arranged them in such an order: 4-heptanone, 2,4-dimethyl-3-pentanone, heptanal.

Synthesis of Aldehydes and Ketones

- 5. Write an equation for the synthesis of 2-hexanone by hydration of an alkyne.
- 6. Write an equation for the synthesis of pentanal from an alcohol.

Reactions of Aldehydes and Ketones

- 7. Write an equation for the reaction, if any, of *p*-bromobenzaldehyde with each of the following reagents, and name the organic product:
 - a. Tollens reagent
 - b. cyanide ion
 - c. methyl amine (CH₃NH₂)
- 8. What simple chemical test can distinguish between the following pair of compounds: benzyl alcohol and benzaldhyde.

9. Complete each of the following equations:

Reactions with Grignard Reagents and Other Nucleophiles

- 10. Write an equation for the reaction of each of the following with methylmagnesium bromide, followed by hydrolysis with aqueous acid:
 - a. acetaldehyde

- b. acetophenone
- 11. Using a Grignard reagent and the appropriate aldehyde or ketone, show how each of the following can be prepared:
 - a. 2-methyl-2-pentanol

b. 3-butene-2-ol

Oxidations and Reductions

12. Give the structure of each product:

Enols, Enolates, and the Aldol Reaction

14. Write the steps in the mechanism for the aldol condensation of butanal.