

Write formulas for the following:

Write names for the following:

1.  $\text{CuCl}_2$
2.  $\text{NaNO}_3$
3.  $\text{FeAsO}_4$
4.  $\text{Mg}(\text{H}_2\text{PO}_4)_2$
5.  $\text{SnBr}_2$
6.  $\text{LiOH}$
7.  $\text{NH}_4\text{CN}$
8.  $\text{Ca}(\text{ClO})_2$
9.  $\text{SrI}_2$
10.  $\text{K}_2\text{Cr}_2\text{O}_7$
11.  $\text{CuNO}_2$
12.  $\text{Na}_2\text{Cr}_2\text{O}_4$
13.  $\text{Ag}_2\text{SO}_4$
14.  $\text{CsHSO}_4$
15.  $\text{Ba}_3(\text{PO}_3)_2$
16.  $\text{Be}(\text{OH})_2$
17.  $\text{Al}(\text{IO}_3)_3$
18.  $\text{Ga}_2\text{Se}_3$
19.  $\text{RbC}_2\text{H}_3\text{O}_2$
20.  $\text{Pb}(\text{CrO}_4)_2$
21.  $\text{FeCO}_3$
22.  $\text{Cd}_3(\text{PO}_4)_2$
23.  $\text{CrO}$
24.  $\text{NaIO}$
25.  $\text{Ni}_3(\text{AsO}_3)_2$
26. barium bromide
27. manganous carbonate
28. calcium hydrogen phosphate
29. cobalt (II) bicarbonate
30. ferrous bisulfate
31. ammonium sulfite
32. potassium phosphide
33. manganic chlorite
34. ferric nitrate
35. mercurous bromate
36. cupric perchlorate
37. potassium permanganate
38. chromium (II) dihydrogen phosphate
39. calcium acetate
40. sodium oxide
41. magnesium arsenide
42. cadmium hydrogen sulfate
43. chromic chloride
44. stannous phosphate
45. barium cyanide
46. cobaltic dichromate
47. silver sulfide
48. strontium nitride
49. plumbic fluoride
50. zinc hypobromite