



ISA Delhi Section

Setting the Standard for Automation™

FERTILIZER MEET 2017 16TH DECEMBER 2017

Advanced Analysers for Syn Gas Optimisation

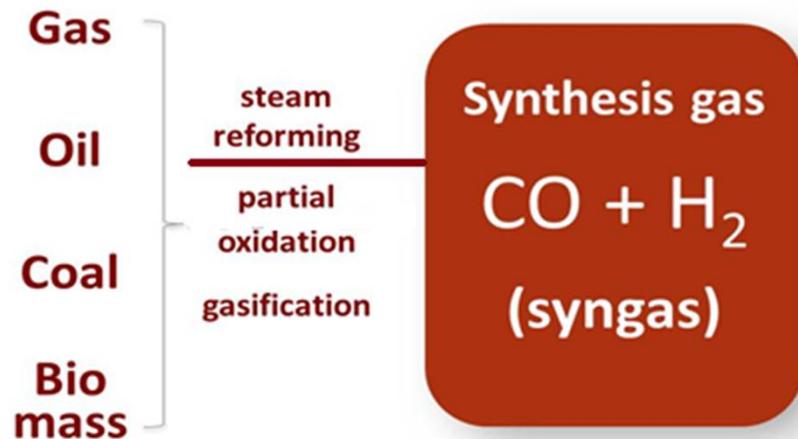
Presenter: Jiwan Jain

Standards
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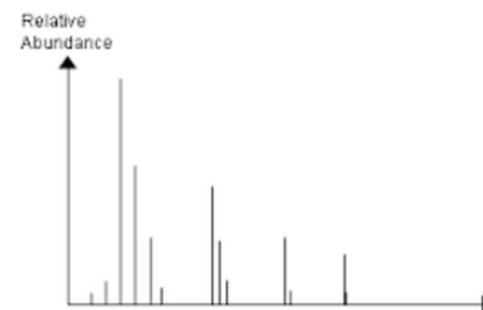
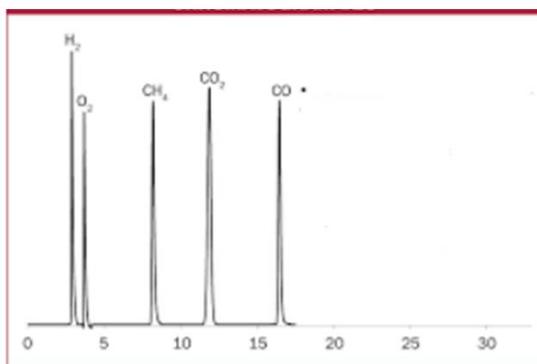
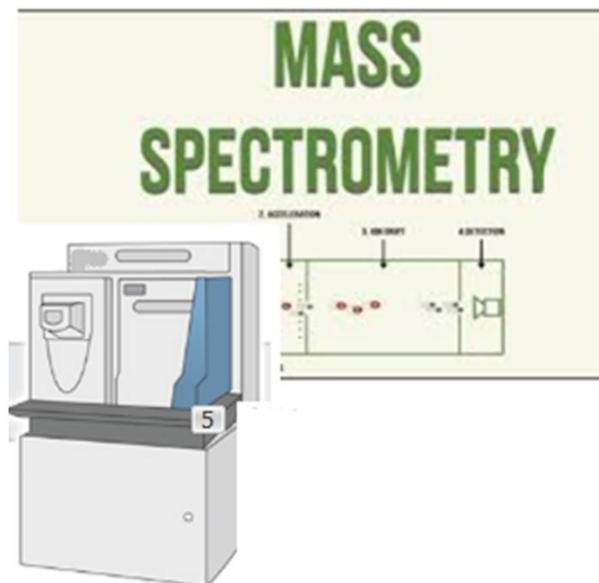
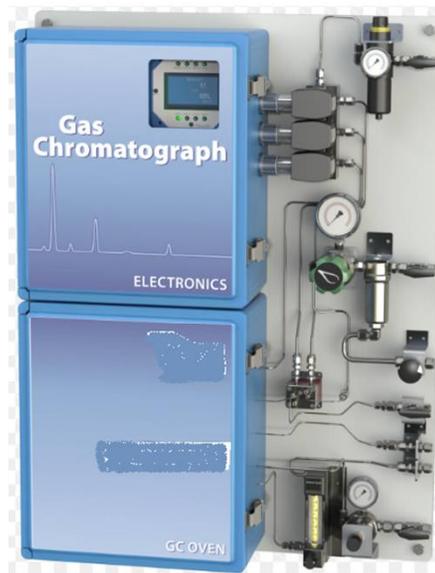
The International Society of Automation
Delhi Section

Syngas production methods

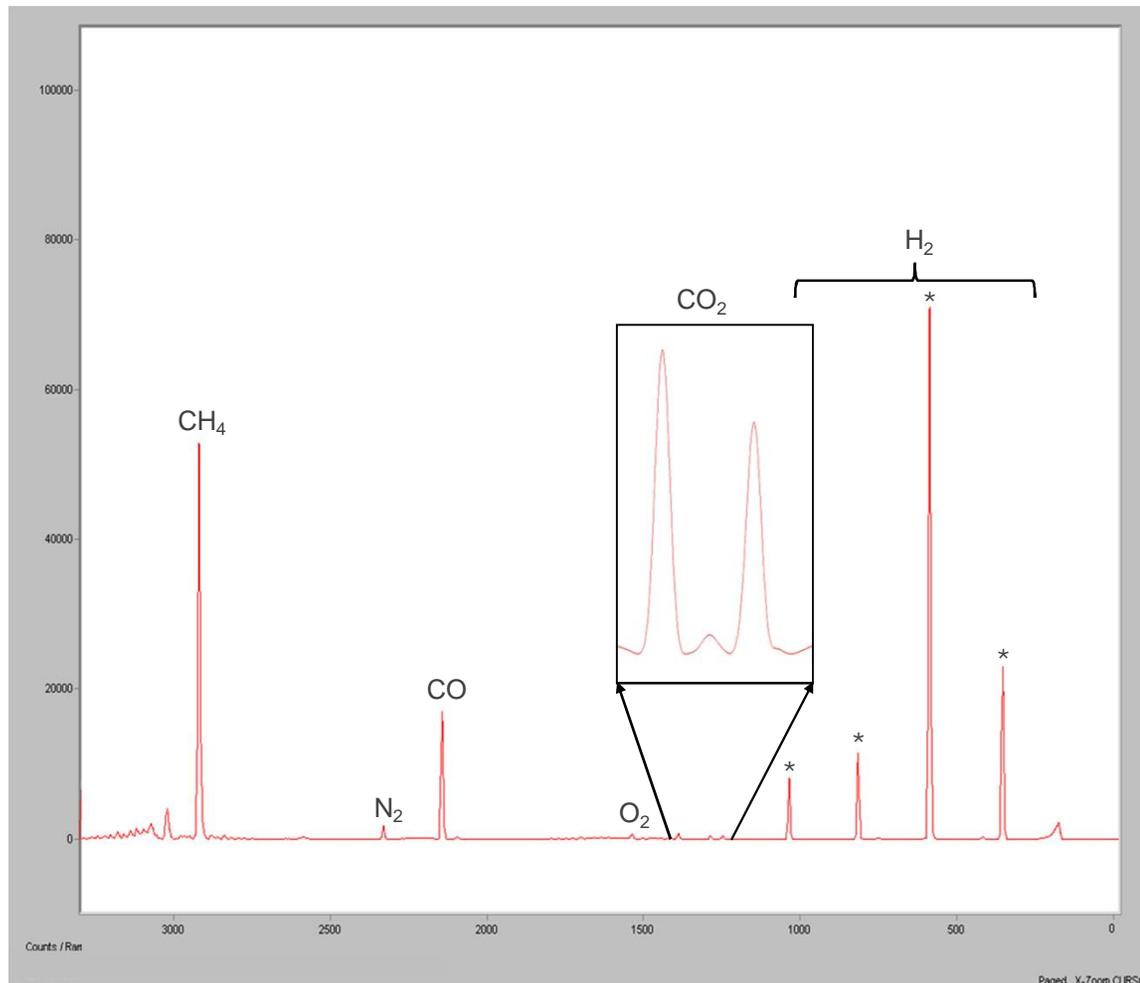
- The most common method is via reforming of natural gas
 - Steam Methane Reforming (SMR), the most common method in use
 - Partial Oxidation (POX)
 - Autothermal Reforming (ATR)
- Gasification of various feedstocks (common in areas that do not have natural gas and have high coal reserves, such as China)
 - Coal
 - Petcoke
 - Biomass



Current methods of Syn Gas analysis



Typical composition picture for Syngas from Methane Reforming

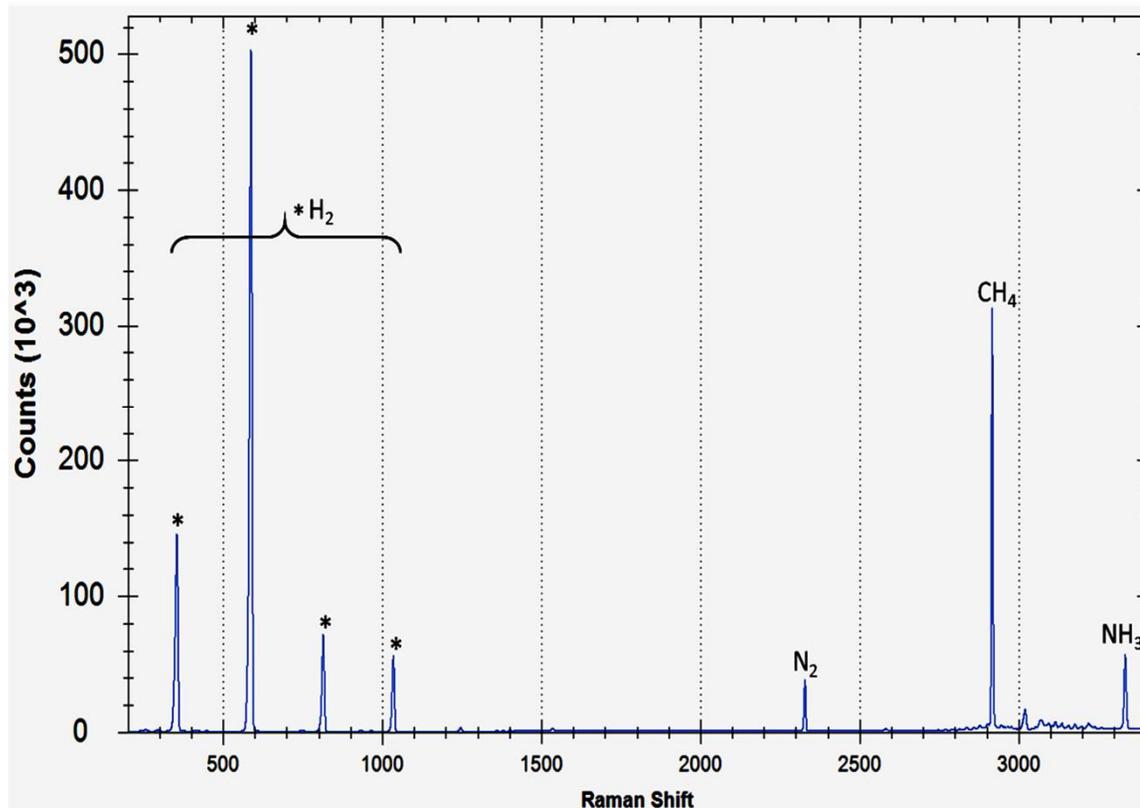


- Single probe can do total measurement
- Only spectroscopic technique that can analyze H₂, N₂, O₂
- Complete baseline separation
- Spectrum has the simplicity of a chromatogram
- Use peak areas for concentrations (no complex models to maintain)

Ammonia Plant Applications

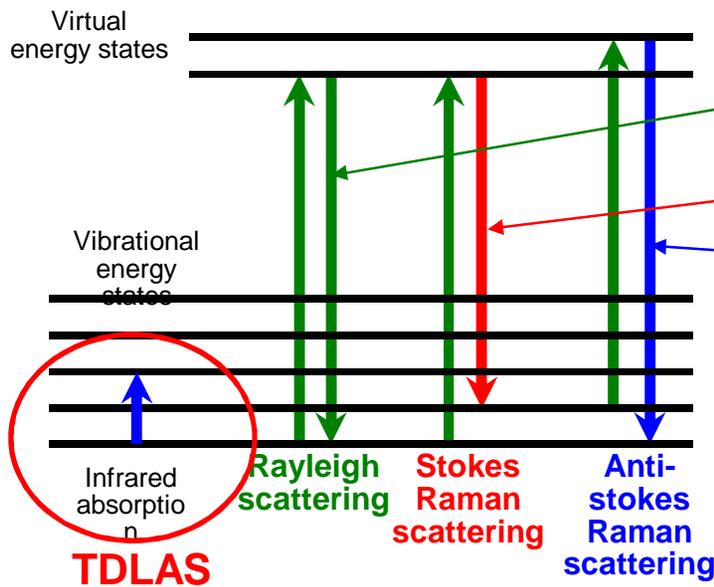
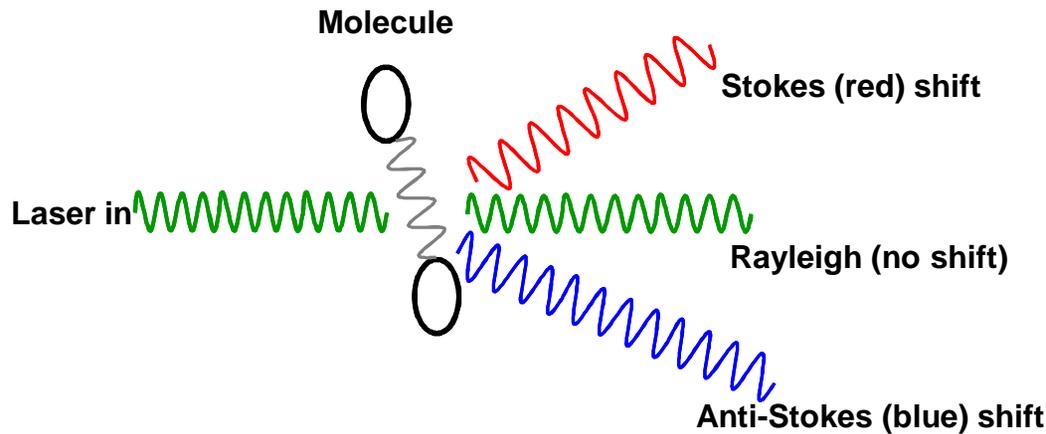
	Stream Service	Key Measurement Parameter	Pressure* (barg)	Temperature* (°C)	Recommended Sampling Interface
1	Natural Gas Feed to Reformer	Carbon Number	26	25	OptoAST
2	Fuel Gas to Reformer Furnace	BTU	6	40	OptoAST
3	Syngas from Primary Reformer	Composition/CH ₄	36	800	OptoDRS
4	Syngas from Secondary Reformer/ Inlet to HT Shift Converter	Composition/CO	35	370	OptoDRS
5	HT Shift Converter Outlet/LT Shift Converter Inlet	Composition/CO	34	445	OptoDRS
6	LT Shift Converter Outlet/CO ₂ Absorber Inlet	Composition/CO ₂	32	220	OptoDRS
7	CO ₂ Absorber Outlet/Methanator Inlet	Composition/CO ₂	31	25	OptoAST
8	Methanator Outlet/Purified Syngas	Composition/H ₂ /N ₂	30	330	OptoAST
9	Feed to Synthesis Loop	H ₂ /N ₂ Ratio	57	400	OptoAST
10	Synthesis Loop Recycle Gas	Composition/ Impurities	220	440	OptoAST
11	Synthesis Loop Purge Gas	CH ₄ Impurities	150	25	OptoAST

Typical composition picture for Ammonia Synthesis Loop Recycle



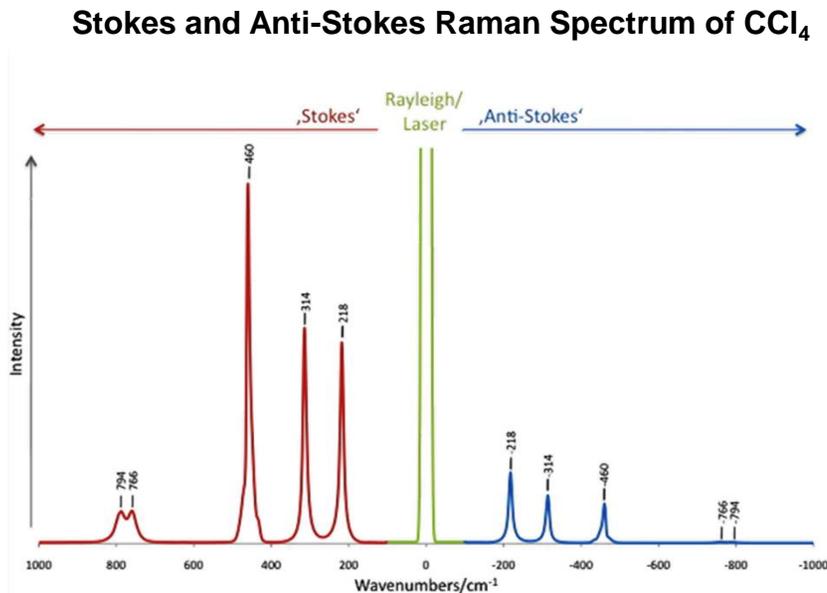
- Single probe can do total measurement
- Only spectroscopic technique that can analyze H₂, N₂, O₂
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Basics of Raman Spectroscopy



- Molecular bonds vibrate
- Different bonds have different strength, or energy
- Raman scatter involves transfer of energy to these bonds
- There are three main types of Raman scattering
 - Rayleigh - No change in wavelength
 - Stokes Raman – Shift to lower energy (red-shifted)
 - Anti-Stokes – Shift to higher energy (blue-shifted)
- The color shift is unique to each chemical species
- The Optograf™ Analyzer uses Stokes Raman

What Does a Raman Spectrum Look Like?

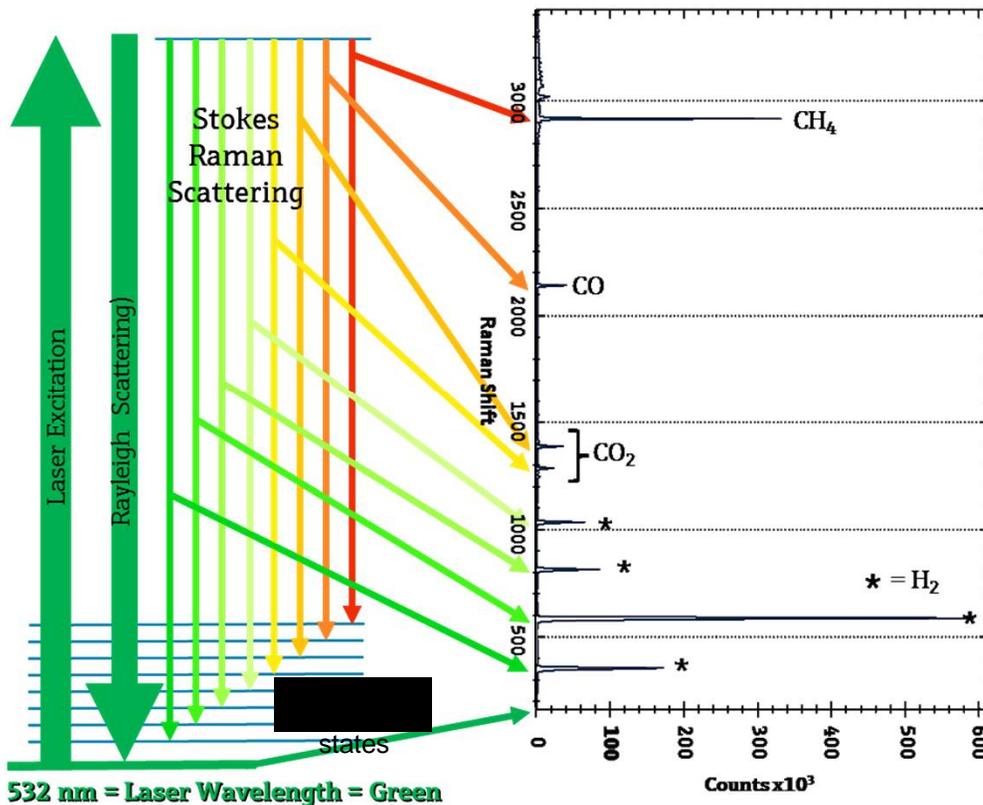


- Raman data are plotted as Raman Shift versus Intensity
- Raman shift is the amount of energy transferred to a bond
- Rayleigh scatter has no energy transfer, so has a Raman Shift of 0.
- Stokes and Anti-stokes lines have the same Raman Shift values
- Anti-stokes peaks are less intense

NOTE: Ratio of Anti-stokes to Stokes line intensities can be used to measure temperature

What about the Raman Spectrum of a Gas Mixture?

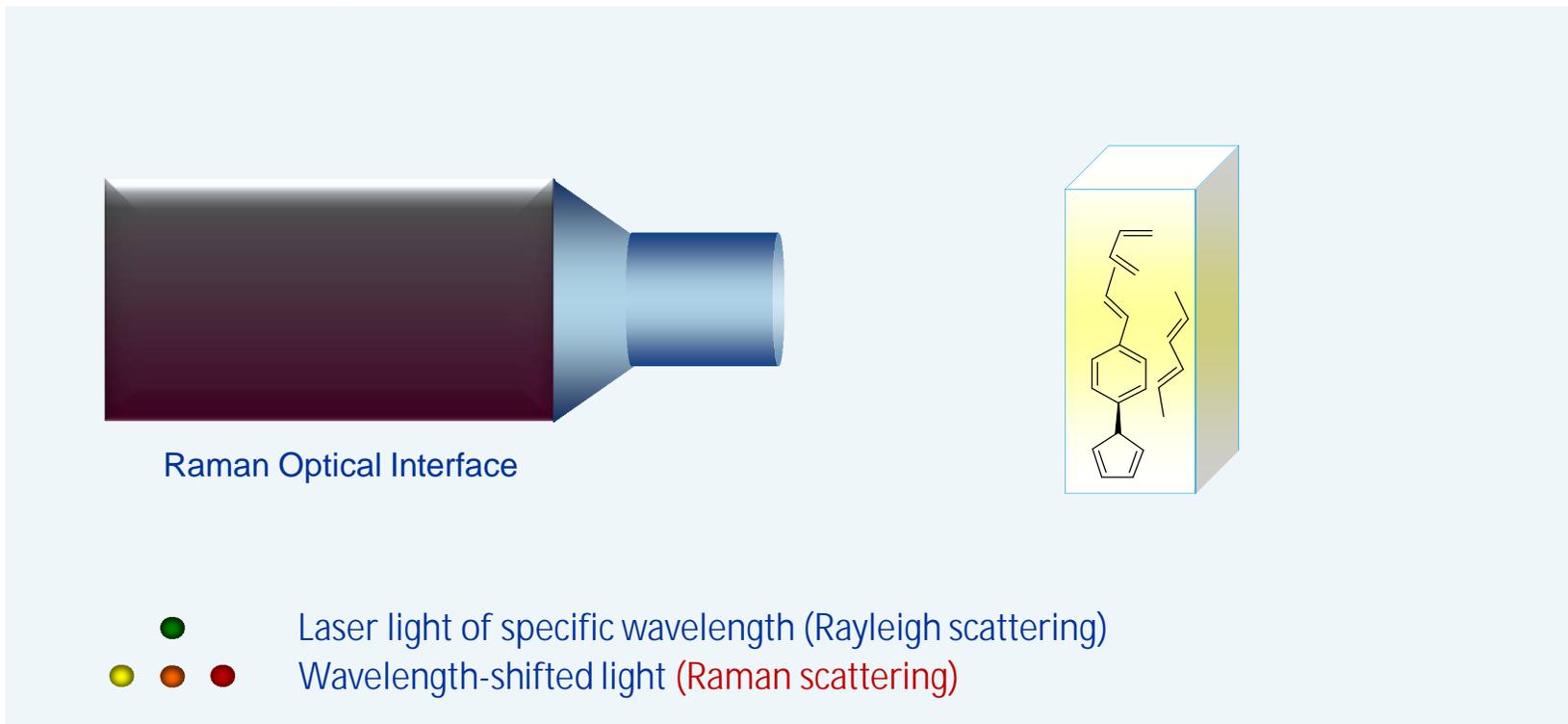
'One Color In, Multiple Colors Out'



- Simple molecular gases (e.g. CO, CO₂, H₂, N₂, NH₃) have simple Raman spectra, often a single peak
- Raman scatter from each type of molecule in the sample results in a different wavelength (color) of light emitted from the sample
- The Optograf analyzer measures all of these gases simultaneously, separating them 'by color'
- Simple spectra allow for the use of simple 'Method-based' analysis of mixtures

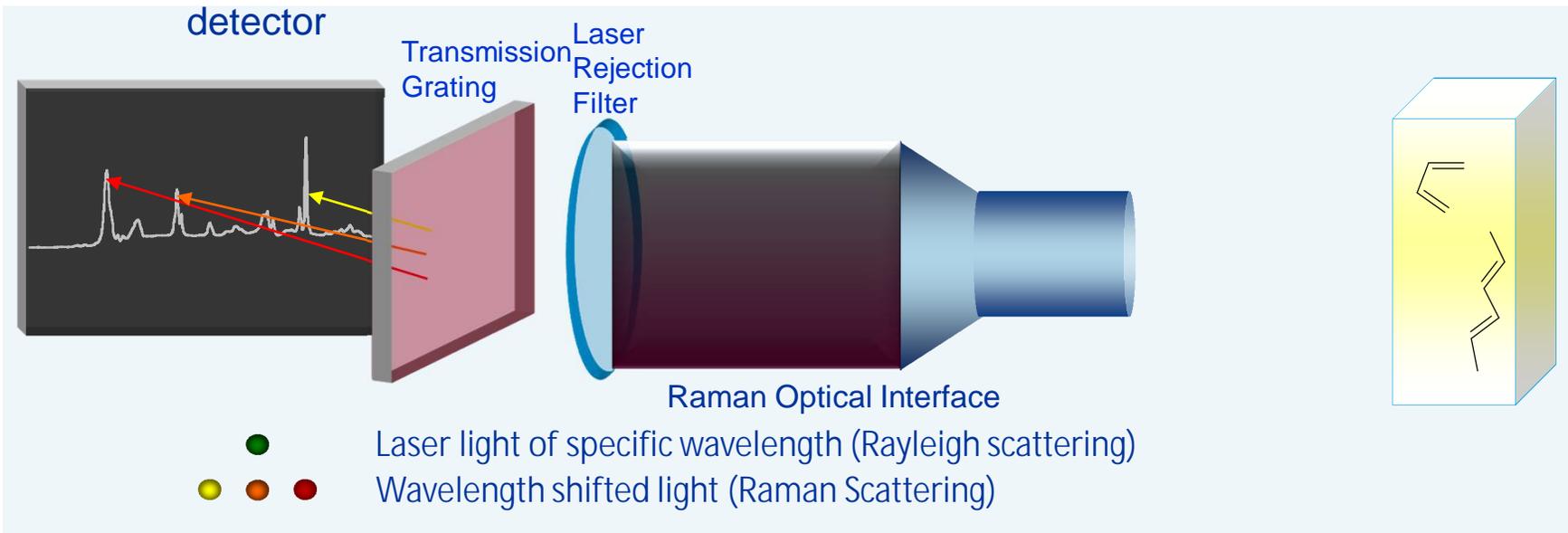
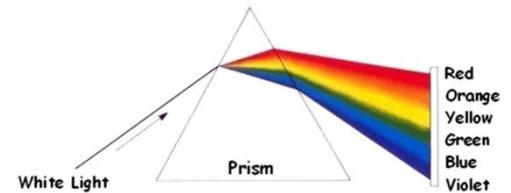
Raman Scattering – How is Raman Data Generated?

- Laser light of a specific wavelength interacts with molecules
- A number of scattered photons are shifted in wavelength due to specific energy transfer to molecules in the sample



Raman Scattering – How is Raman Data Collected?

- Only back scattered photons are collected
- Laser Rejection Filter removes original laser light
- Light is spatially separated using a Transmission Holographic Grating, like a Prism separates sunlight into colors
- Raman photons are simultaneously detected with a CCD detector

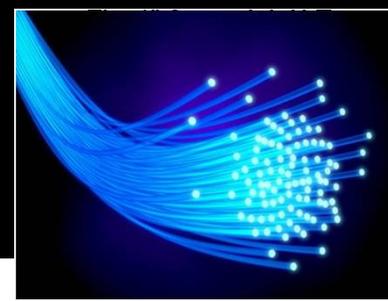
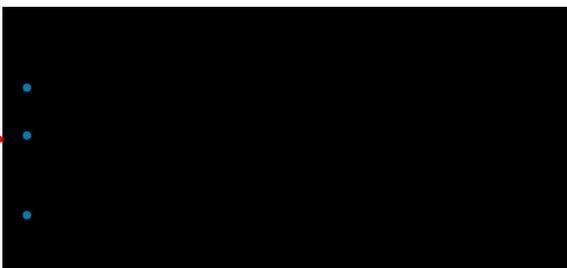
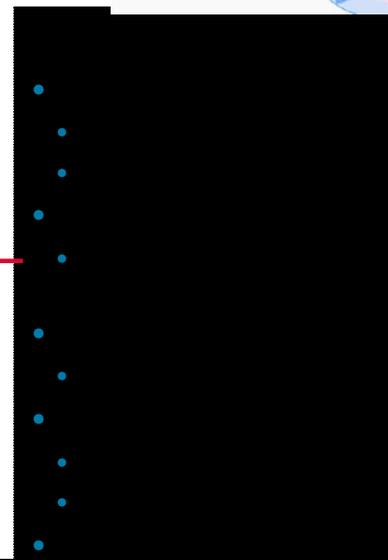
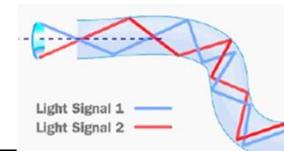
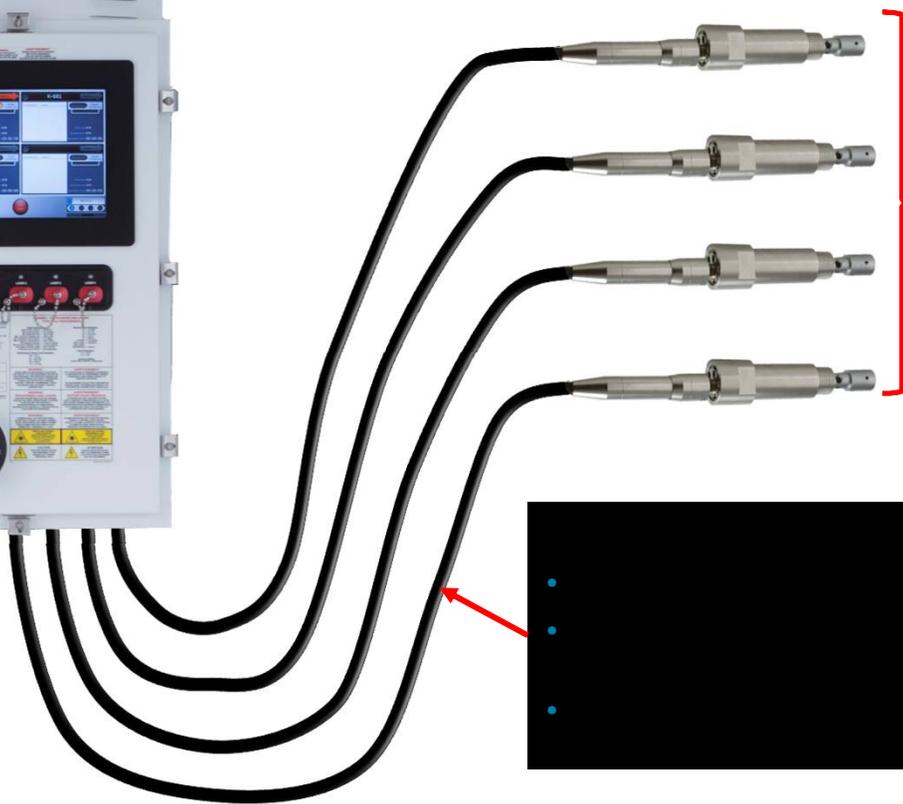


The Optograf Analyzer

Optograf Base Unit
(18"W x 10"D x 32"H)



Fiberoptic Sensor = Sample Interface
Pipe-centric Enabler



The complete Raman solution

Optical fiber
<100m

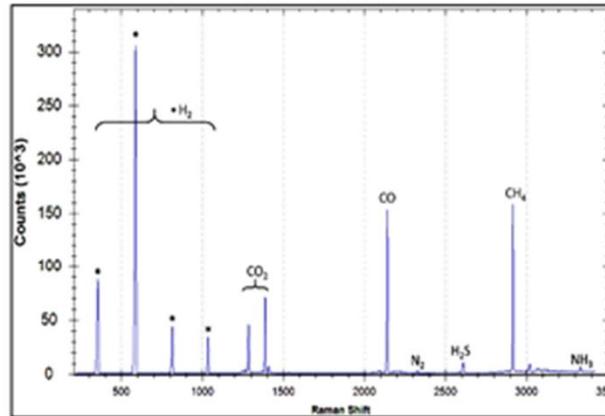
Sample Probe

- Installed at the sample tap
- Sample Streams include
 - Reformers
 - Gasifiers
 - Shift Converters
 - PSA
 - Methanators
 - Synthesis Loops
 - Ammonia
 - Methanol



Sample Interface

- Non-extractive Optical Probe
- Multiple Options
 - Conventional
 - OptoAST
 - Moderate Temp and Dry
 - OptoDRS
 - Particle-laden, Hot, Wet
- Works at process P and T
- No Flare
- Class 1/Div 1; Zone 1



Optogram

Analysis Result

- Full stream composition
- Baseline peak separation
- Peak areas proportional to concentration
- Simple method-based analysis (no complex models)
- Can output BTU, Wobbe Index as derived values



Base Unit

- Laser-based analyzer
- No columns
- No stream switching
- No sample transport
- No carrier gas
- Analyze 4 independent streams at one time
- Class 1/Div 2; Zone 2

Syngas Installation Example – Fieldside mounting arrangement of Raman Pro



Optograf Installation Example – Base Unit



Optograf™ Analyzer for Syngas – Advantages

- Reliability of a spectroscopic system (no critical moving parts)
- Unique spectroscopic capability of measuring the diatomics H₂, N₂
- Transparent to moisture vapor – streams do not have to be bone dry
- Speciation similar to a GC but without valves, columns, carrier gas
- Maintenance skill level required is less than that for a GC
- No consumables
- No time penalty for update times when analyzing multiple streams; all streams analyzed simultaneously
- No regular calibration required
- No chemometrics modelling or model maintenance is required

Syngas Pipe-Centric Analysis – Advantages

- More reliable analysis: sampling and conditioning at the sample tap (no sample transport and conditioning issues at the analyzer/shelter)
- Reduces infrastructure costs
 - Minimize sample conditioning – working at line pressure
 - Sample transport – no need heat traced tubing
 - No shelter – in many parts of the world a shelter is optional
- Sampling and measuring at line temperature/pressure provides for better opportunities to return sample to the process (no flaring)
- Analysis update times are reduced (no lag time, no sample transport)
- Improved safety, as any hazardous/explosive or toxic samples are contained at the sampling point

Thank you very much for your attention

